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Intellectual Property Office

Trade mark number

UK00003356885

Status

Registered

Overview

Trade mark



Dates

Filing date

28 November 2018

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28 November 2028

Goods and services

Classes and terms

Class 1

Flavour enhancers for food;Protein for food for human consumption [raw material];Protein for use in the

manufacture of foodstuffs;Proteins for the food industry;Proteins for use in the manufacture of food products;Proteins for use in the manufacture of food supplements;Food protein as a raw material.

Class 5

Food for medically restricted diets;Food supplements;Food supplements consisting of amino acids;Food supplements for medical purposes;Food supplements for non-medical purposes;Food supplements for veterinary use.

Class 29

Meat-based snack foods.

Class 30

Flavourings for foods.

Names and addresses

Owner(s) name

Ag-Row Consultants Ltd

Bellinster Business Pk, Berners Cross, Winkleigh, Devon, EX19 8DH, United Kingdom

Country of Incorporation

United Kingdom

Company registration number

7425637

Publications

First advert

Journal

2018/049

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Case Ref	Instructor	Instructor's Reference	Owner	Official No.	Title	Country	Application No.	Application Date	Registration Date	Registration No.	Status Sum	Next Renewal Date
P068248.AUP	Elemental Digest Limited	Elemental - Fertiliser - Australia	Elemental Digest Limited	2014282964	Fertiliser	Australia	2014282964	19-Jun-14	21-Nov-19	2014282964	Registered	19-Jun-25
P068248.BRP	Elemental Digest Limited	Elemental - Fertiliser - Brazil	Elemental Digest Limited	BR 112015032008-2	Fertiliser	Brazil	BR 112015032008-2	19-Jun-14	13-Oct-21	BR 112015032008-2	Registered	19-Jun-25
P068248.CAP	Elemental Digest Limited	Elemental - Fertiliser - Canada	Elemental Digest Limited	2 915 964	Fertiliser	Canada	2 915 964	19-Jun-14	16-Nov-21	2 915 964	Registered	19-Jun-25
P068248.DEP	Elemental Digest Limited	Elemental - Fertiliser - Europe	Elemental Digest Limited	3 010 871	Fertiliser	Germany	14732609.4 	19-Jun-14	16-Jun-21	3 010 871	Registered	30-Jun-25
P068248.EAP	Elemental Digest Limited	Elemental - Fertiliser - Eurasia	Elemental Digest Limited	29332	Fertiliser	Eurasian Patent Organization	201690065	19-Jun-14	30-Mar-18	29332	Registered	19-Jun-25
P068248.FRP	Elemental Digest Limited	Elemental - Fertiliser - Europe	Elemental Digest Limited	3 010 871 (E)	Fertiliser	France	14732609.4 	19-Jun-14	16-Jun-21	3 010 871 (E)	Registered	30-Jun-25
P068248.GBP	Elemental Digest Limited	Elemental - Fertiliser - UK	Elemental Digest Limited	2 531 463	Fertiliser	United Kingdom	1601070.4	19-Jun-14	09-Jun-21	2 531 463	Registered	30-Jun-25
P068248.GBP1	Elemental Digest Limited	Elemental - Fertiliser - Europe	Elemental Digest Limited	3 010 871 (E)	Fertiliser	United Kingdom	14732609.4 	19-Jun-14	16-Jun-21	3 010 871 (E)	Registered	30-Jun-25
P068248.IEP	Elemental Digest Limited	Elemental - Fertiliser - Europe	Elemental Digest Limited	3 010 871 (E)	Fertiliser	Ireland	14732609.4 	19-Jun-14	16-Jun-21	3 010 871 (E)	Registered	30-Jun-25
P068248.INP	Elemental Digest Limited	Elemental - Fertiliser - India	Elemental Digest Limited	322699	Fertiliser	India	2.01627E+11	19-Jun-14	14-Oct-19	322699	Registered	19-Jun-25
P068248.ITP	Elemental Digest Limited	Elemental - Fertiliser - Europe	Elemental Digest Limited	3 010 871 (E)	Fertiliser	Italy	14732609.4 	19-Jun-14	16-Jun-21	3 010 871 (E)	Registered	30-Jun-25
P068248.MXP	Elemental Digest Limited	Elemental - Fertiliser - Mexico	Elemental Digest Limited	366240	Fertiliser	Mexico	MX/a/2015/017871	19-Jun-14	03-Jul-19	366240	Registered	30-Jun-29
P068248.NLP	Elemental Digest Limited	Elemental - Fertiliser - Europe	Elemental Digest Limited	3 010 871 (E)	Fertiliser	Netherlands	14732609.4 	19-Jun-14	16-Jun-21	3 010 871 (E)	Registered	30-Jun-25
P068248.NZP	Elemental Digest Limited	Elemental - Fertiliser - New Zealand	Elemental Digest Limited	716125	Fertiliser	New Zealand	716125	19-Jun-14	28-Sep-21	716125	Registered	19-Jun-25
P068248.PLP	Elemental Digest Limited	Elemental - Fertiliser - Europe	Elemental Digest Limited	3 010 871 (E)	Fertiliser	Poland	14732609.4 	19-Jun-14	16-Jun-21	3 010 871 (E)	Registered	19-Jun-25
P068248.USP	Elemental Digest Limited	Elemental - Fertiliser - USA	Elemental Digest Limited	10 087 115	Fertiliser	United States of America	14/900 337	19-Jun-14	02-Oct-18	10 087 115	Registered	02-Apr-26
P068248.ZAP	Elemental Digest Limited	Elemental - Fertiliser - South Africa	Elemental Digest Limited	2016/00418	Fertiliser	South Africa	2016/00418	19-Jun-14	27-Jul-22	2016/00418	Registered	19-Jun-25

14/900 337

2. US20160145164 - ORGANIC WASTE PROCESSING

[National Biblio. Data](#)
[Description](#)
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PermaLink

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United States of America

Application Number

14900337

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IPC

C05F 1/00	C05F 9/00	C05F 11/00
C05G 3/00	C05D 3/02	C05F 3/00

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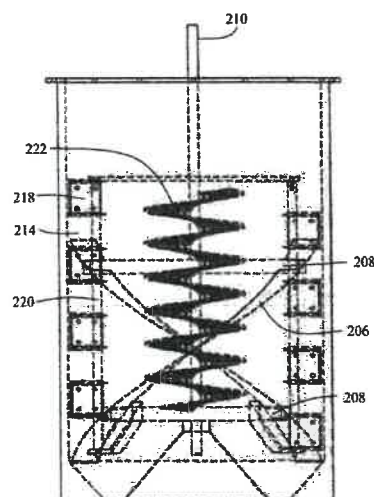
C05D 3/02	C05F 1/005	C05F 11/00
C05F 11/08	C05F 3/00	C05F 9/00

[View more classifications](#)**Applicants**ELEMENTAL DIGEST LIMITED
Elemental Digital Limited**Inventors**

Bruce William Palmer

Agents

Boutware & Valoir

Priority Data1311108.3 21.06.2013 GB
1322874.7 23.12.2013 GB**Title****[EN]** Organic waste processing**Abstract****[EN]**

The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertilizer from organic waste, especially slow release fertilizer. The present invention also relates to fertilizers made by the method of the invention, especially slow release fertilizer. The method of processing organic waste to provide a slow-release fertilizer may comprise at least one hydrolysis step to provide hydrolyzed organic waste and a setting step to provide a slow release fertilizer.

Related patent documents

[MYPI 2015704654](#) [MY-179142-A](#) [PL3010871](#) [W0/2014/202986](#) [JP2016527168](#) [BR112015032008](#)
[TH159506](#) [PH1/2015/502840](#) [GB2531463](#) [AU2014282964](#) [EA201690065](#) [CN105452196](#)
[EP3010871](#) [CA2915964](#) [MX366240](#) [IN201627001450](#) [NZ716125](#) [C07590580](#)



2. US20160145164 - ORGANIC WASTE PROCESSING



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Machine translation

Note: Text based on automatic Optical Character Recognition processes. Please use the PDF version for legal matters

[EN]

CROSS REFERENCE TO RELATED APPLICATION

This application is a national stage application filing under 35 U.S.C. § 371 of PCT International application PCT/GB2014/051883 designating the Great Britain, filed Jun. 19, 2014, which claims the benefit under 35 U.S.C. § 119(a) of G.B. application Ser. No. 1311108.3, filed Jun. 21, 2013, and 1322874.7, filed Dec. 23, 2013, each of which is incorporated here by reference in its entirety for all purposes.

FIELD OF THE INVENTION

The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertiliser from organic waste, especially slow release fertiliser. The present invention also relates to fertilisers made by the method of the invention, especially slow release fertiliser.

BACKGROUND TO THE INVENTION

Organic waste may be animal by-products, vegetable waste, food waste or a mixture of both.

During the processing of animals for meat production or other uses, relatively large amounts of the animal are removed, discarded and not sold on to an end user. Such discarded by-products include animal by-products such as bone, blood, gut content, wool, fur and feathers for example. Animal by-products also include animal waste such as chicken-litter, cow slurry and pig or horse manure. Vegetable waste may include kitchen waste, bedding from animals, fruit or vegetable processing waste, for example, fruit peels, sugar cane waste. This represents a significant waste of the resources that could be harnessed for useful purposes. There is also a cost involved in storing and disposing of such by-products as waste.

Various processes are known for processing organic waste. It is known to use incineration and composting to process the waste into a usable form or extract energy from waste.

Organic waste comprises valuable nutrients and there is a need for methods of processing organic waste to provide useful products. There is also a need for and environmentally friendly alternatives to conventional fertiliser for crops.

Slow release or controlled release fertilisers are particularly valuable because they release nutrients to plants over a period of time, for example several months. This means that there is less danger of toxicity to plants and of fertiliser running off the land before it is taken up by plants. Slow release fertilisers can provide a sustained supply of nutrients through a longer period of the growing season compared to fertilisers that are easily soluble and release nutrients quickly.

It is an aim of the present invention to provide a method for disposing of organic waste and a method for processing organic waste to provide a slow release fertiliser.

The present invention provides a method of processing organic waste comprising the step of mixing the organic waste with a hydrolysing composition.

The method of processing organic waste may be a method of making a slow release fertiliser.

The method may be carried out in a suitable apparatus comprising one or more vessels. The one or more vessels may be equipped with a mechanism for lifting the mixture upwards in a central region of the vessel by means of a lifting mechanism, resisting downward movement of the mixture in an outer region of the vessel by means of a heat transfer member, and transferring heat within the mixture using the heat transfer member. The vessels may also be equipped with a heating mechanism to heat the contents of the vessel.

The heat transfer member may be rotatably mounted in the vessel and may have a leading end and a trailing end. The leading end may be below the trailing end such that rotation of the heat transfer member generates a lifting force acting on the by-product, which may be arranged to resist its downward movement.

The heat transfer member may be formed from a strip of metal. The heat transfer member may be part helical.

The lifting mechanism may comprise a helical lifting member rotatable to lift the by-product. The lifting mechanism and the heat transfer member may be mounted on a common apparatus may comprise scraping means, for example one or more scrapers, arranged to scrape the inside

of the vessel wall, or to scrape by-product, or other contents of the vessel, from the side wall. The scraping means may be mounted on the common rotatable support means for rotation with the heat transfer member.

The vessel wall may be cylindrical, for example having a central axis that is vertical. The vessel may further comprise heating means arranged to heat the by-product in the vessel.

The apparatus may further comprise a further processing vessel also comprising: a vessel wall; a lifting mechanism, which may be located in a central region of the vessel, arranged to lift the by-product within the vessel; a metal heat transfer member, which may extend around the lifting mechanism and may be in an outer region of the vessel, the heat transfer member optionally being rotatable around the outer region of the vessel to resist downward movement of the by-product in the outer region of the vessel and to dissipate heat through the by-product in the outer region of the vessel.

The vessel, or one of the vessels, may have an inlet connected to a supply of water. The vessel, or one of the vessels, may have an inlet connected to a supply of steam. The vessel, or one of the vessels, may have inlet means connected to a supply of acid and hydrogen peroxide.

The method may further comprise heating the mixture in the vessel.

The method may be carried out in a plant according to the invention.

The present invention may be carried out in an apparatus for processing organic waste the apparatus comprising a heating vessel having a product inlet arranged to receive the by-product, a water inlet arranged to receive water, and a steam inlet arranged to receive steam, and heating means arranged to heat the first heating vessel, a separator arranged to receive the contents of the first heating vessel and separate it into first and second components, and an acidification vessel having a product inlet arranged to receive the solid component, and acid inlet means arranged to receive acid. The acid inlet means may further be arranged to receive hydrogen peroxide.

Preferably the apparatus further comprises means to deliver a lime mix to the acidified by-product.

Preferably the apparatus further comprises means to deliver magnesium salts to the acidified by-product.

The apparatus may comprise a dryer operative to receive and dry the acidified by-product.

A by-product sizing apparatus may be provided operative to re-size the by-product into pieces below a predetermined size.

The sizing apparatus may comprise a crusher operative to crush the raw organic waste into pieces below a predetermined size.

The sizing apparatus may comprise a plurality of crushers arranged in sequence and operative to sequentially crush the raw organic waste into smaller pieces.

Preferably the sizing apparatus is operative to output pieces of less than 40 mm.

A metal detector may be provided at the inlet operative to detect any metal content in the raw organic waste and to generate a corresponding signal that is used to prevent raw organic waste entering the by-product sizing apparatus if metal content is detected.

Preferably the apparatus further comprises a pre-heating vessel operative to pre-heat the sized by-product and to deliver the sized by-product to the heating vessel.

The apparatus may further comprise a supply of at least one additive and means to deliver the additive to the further heated by-product to alter the properties of the further heated by-product.

The apparatus may comprise a pellet press or granulator operative to receive dried by-product and to press or granulate the dried by-product into pellets or granules.

The apparatus may further comprise a packaging unit or bagging machine operative to receive and package or bag the pellets or granules.

The method may comprise a step of sizing the organic waste, for example by crushing, grinding or chopping solid waste to provide suitable sized pieces. Suitable sized pieces may be less than 200 mm, less than 150 mm, less than 100 mm, less than 50 mm or less than 20 mm sized pieces.

Water may be added to the organic waste before or after sizing the organic waste.

The method may comprise a heating step. The organic waste may be heated, for example to between 70° C. and 200° C. The organic waste may be heated for between 10 and 60 minutes. The organic waste may be heated for 20 minutes at 135° C. and 3 bar pressure to provide microbiological safety. The organic waste may be heated for a time and/or temperature as specified in legislation regarding disposal of organic waste or animal waste and/or microbiological safety. The organic waste may be heated by adding steam to the organic waste, for example steam may be added in a vessel as shown at [18] or [20] in FIG. 1. The organic waste may be heated at higher than atmospheric pressure, for example in a pressurised vessel shown at [20] in FIG. 1. The organic waste may be heated by heating the vessel in which the organic waste is held. The organic waste may be heated by an exothermic reaction when a hydrolysing composition is added to the organic waste, for example in a vessel as shown at [22] in FIG. 1. The method may comprise one or more, two or more, three or more, four or more or five or more heating steps. Each heating step may be performed in a different vessel for example a heating step may be performed in each of vessels [18], [19] and [20] shown in FIG. 1. The method may comprise two or more heating steps performed at different temperatures or pressures. Heating steps may be used to remove gelatine and fat; degrade the organic waste and/or provide a microbiologically safe product from organic waste.

The heating step may be an independent step in the method of processing organic waste performed at a different time from other steps. The heating step may be performed at the same time as another step in the method, for example the hydrolysis step. Combining the organic waste with a hydrolysing composition may cause an exothermic reaction that heats the organic waste. During the heating step the organic waste may be heated using a combination of heating methods.

During the acid-peroxide/high temp/high pressure and/or digestion step a metal catalyst may be added. The metal catalyst may be any suitable metal or metal compound that is able to increase the rate of reaction with respect to breaking apart protein material such as the hydroxyproline in collagen into the constituent amino acid building blocks. The subsequently released nitrogen atoms from the peptide links are released in the form of ammonium ions. In particular these catalysts could be very useful in the dissolving of bone in acid/peroxide digestion step. The catalysts may include copper or copper compounds specifically copper II sulphate or copper II oxide, selenium or selenium compounds, titanium and/or titanium compounds but specifically titanium dioxide as either a powder or a nano-powder. Some of the catalysts, such as selenium and copper, would add benefit to the micronutrient composition of the final fertiliser product. The catalyst may be added in very small quantities, for example the mass of catalyst may be in the range of 0.1%-0.01% of the mass of undigested organic waste, for example bone paste.

The method may comprise a step of draining liquid from the organic waste. The step of draining liquid from the organic waste may be performed after the heating step. Liquid may be removed from the mixture. The liquid may comprise fat and/or gelatine which may be separated and processed separately.

The method may comprise one or more hydrolysis steps where a hydrolysing composition is combined with the organic waste. The hydrolysing composition may be for example:

a combination of an acid and hydrogen peroxide. These may be added to the organic waste together or sequentially. For example Sulphuric acid H_2SO_4 50-100% [for example 95%] followed by or mixed with Hydrogen Peroxide H_2O_2 5-100% [for example 30% v/v] in a ratio of 1:1 to 1:10 sulphuric acid solution:hydrogen peroxide or for example, nitric acid 37% nitric acid [70% or 1.42 specific gravity] followed by or mixed with 7% hydrogen peroxide [35% w/w];

a combination of a base and hydrogen peroxide. These may be added to the organic waste together or sequentially. For example ammonium hydroxide NH_4OH and Hydrogen Peroxide H_2O_2 30%;

a base, for example Potassium hydroxide KOH or sodium hydroxide NaOH; or

an acid, for example Phosphoric acid H₃PO₄; or Hydrochloric acid HCL, Nitric Acid HNO₃, Sulphuric acid H₂SO₄.

a combination of an acid and hydrogen peroxide. These may be added to the organic waste together or sequentially. For example Sulphuric acid H₂SO₄ 50-100% (for example 95%) followed by or mixed with Hydrogen Peroxide H₂O₂ 5-100% (for example 30% v/v) in a ratio of 1:1 to 1:10 sulphuric acid solution:hydrogen peroxide or for example, nitric acid 37% nitric acid (70% or 1.42 specific gravity) followed by or mixed with 7% hydrogen peroxide (35% w/w);

a combination of a base and hydrogen peroxide. These may be added to the organic waste together or sequentially. For example ammonium hydroxide NH₄ OH and Hydrogen Peroxide H₂O₂ 30%;

a base, for example Potassium hydroxide KOH or sodium hydroxide NaOH; or

an acid, for example Phosphoric acid H₃PO₄; or Hydrochloric acid HCL, Nitric Acid HNO₃, Sulphuric acid H₂SO₄.

The method may comprise one hydrolysis step, two hydrolysis steps or three hydrolysis steps. The hydrolysis steps may be performed sequentially. For example, the method may comprise an acid hydrolysis step and a base hydrolysis step performed sequentially. The method may comprise an acid or a base hydrolysis step and a hydrolysis step using acid or base and hydrogen peroxide.

The hydrolysis step may be an exothermic reaction and may cause the organic waste mixture to heat up. Heat may be removed from the organic waste mixture during this step to keep the mixture within suitable temperature limits.

If the hydrolysis step is exothermic it may contribute heat to the heating step.

The one or more hydrolysis steps may be performed in a vessel, for example as shown at [22] in FIG. 1.

After the one or more hydrolysis steps the hydrolysed organic waste may provide an acidified hydrolysed organic waste.

The method may comprise a step of draining liquid from the organic waste. The step of draining liquid from the organic waste may be performed after the hydrolysing step. Liquid may be removed from the mixture. The liquid may comprise fat and/or gelatine and/or other components which may be separated from each other and processed separately.

The method may comprise an acidification step where the hydrolysed organic waste is combined with additional acid to bring the pH of the organic waste to an acid pH below pH 7. An acid, for example phosphoric acid, nitric acid, or sulphuric acid may be added if the organic waste mixture is not acidic after the hydrolysis step or if the organic waste mixture is not acidic enough after the hydrolysis step. The hydrolysed organic waste may be acidified to a pH between pH ~2 and pH 7, between pH 1 and pH 7, between pH 2 and pH 6, between pH 3 and pH 6 or between pH 4 and pH 5. The amount of acid or base added may be enough to make the organic waste mixture slightly acid, for example pH 6.5 to pH 4.

The one or more acidification or neutralization steps may be performed in a vessel, for example as shown at [22] in FIG. 1.

The method may further comprise a step of adding nutrients or minerals, for example magnesium salts to the organic waste mixture. This is advantageous because additional nutrients that are required in a fertiliser may be added to the processed organic waste so that it can be made into a fertiliser with a suitable balance of nutrients. Nutrients or minerals may be selected in order to complement the known components of the organic waste to provide nutrients or minerals that are known to be lacking in the organic waste. Nutrients or minerals may be selected to address a deficiency in the area where the processed organic waste is going to be used as a fertiliser.

The method may further comprise a step of adding ammonia and/or other basic or carbonate materials to increase the nitrogen or other minerals in the organic waste so that the final fertiliser has higher nitrogen or other minerals. For example ammonia and/or other minerals may be added to the organic waste at any point in the processing in order to increase the nitrogen content, the phosphate content or the potassium content of the final fertiliser product. Ammonia and/or other minerals may be added to the organic waste at any point in the processing in order to produce a final fertiliser product with a particular balance of nitrogen phosphate and potassium (NPK content) that is suitable for a specific soil type and/or crop type.

The method may comprise the step of drying the neutralized organic waste mixture. This may be done by a combination of heating and the addition of a dehydrating granulating mixture followed by further heating in the granule drying stage of the process. The acidified organic waste mixture may be dried until it comprises 10-12% moisture.

The method may comprise a step of setting the organic waste mixture by combining acidic, hydrolysed organic waste with a setting mixture or liming mixture. This may be any compound or composition that can form a lime mortar.

The method may comprise a re-sizing step to re-size the organic waste mixture into pieces below a predetermined size. The re-sizing step may, for example, comprise crushing, chopping or grinding the product. The product may be re-sized to pieces of less than 40 mm.

The method may further comprise adding at least one additive to alter the properties of the organic waste mixture. For example, an unpalatable ingredient may be added to deter animals from eating the fertiliser.

The method may comprise the step of pelleting or granulating the organic waste mixture after the setting step.

Granular fertilisers may be treated with hydrophobic anti-caking agents such as Kaolin by externally coating of the granules, or incorporated as part of the granulation mix; and or paraffin based additives which may be sprayed onto hot granules.

The method may further comprise a packaging step where the pellets or granules are packaged.

Organic waste may comprise, for example, waste from abattoirs, waste from meat processing or packaging, fallen stock, animal carcasses, food waste, animal excrement, for example, cow slurry, pig or cow dung, poultry litter, animal bedding, waste from the food industry, abattoir blood waste. Organic waste may be animal waste, vegetable waste or mixed animal and vegetable waste.

The method of processing organic waste may be a method of making fertiliser. The hydrolysing composition may be chosen in order to provide suitable Ingredients to the fertiliser product, for example, use of nitric acid in the hydrolysing composition may provide additional nitrogen in the fertiliser product. Use of phosphoric acid in the hydrolysing composition may provide additional phosphorus in the fertiliser product. Use of sulphuric acid in the hydrolysing composition may provide additional sulphur in the fertiliser product.

The hydrolysing composition, neutralizing composition and additives may be chosen to provide a suitable level of potassium, phosphorous and nitrogen in the final fertiliser when they are combined with the organic waste.

The hydrolysing composition may also be selected because of its suitability for hydrolysing the particular components of the organic waste that it is combined with. For example a hydrolysing composition comprising a base is particularly suitable for hydrolysing organic waste comprising cellulose. A hydrolysing composition comprising acid may be particularly suitable for hydrolysing organic waste comprising bone.

Ammonium hydroxide NH₄ OH and Hydrogen Peroxide H₂O₂ 30% may be particularly suitable for hydrolysing poultry litter, pig manure, cow manure, horse manure and/or contents of intestinal tract.

The method of the present invention may comprise a hydrolysis step using potassium hydroxide KOH and then a neutralizing step using nitric acid to bring the pH close to neutral but keep the mixture slightly alkaline or slightly acidic as required for granulation into granules. The mixture may be set using lime, plaster of paris CaSO₄ ½H₂O, Calcium carbonate CaCO₃, Calcium oxide CaO or Calcium hydroxide Ca(OH)₂.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one hydrolysis step.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step and at least one hydrolysis step.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step, at least one hydrolysis step and at least one neutralizing step.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step, at least one hydrolysis step, at least one neutralizing step and at least one pelleting step.

The method of the present invention may be a method of processing organic waste to provide a slow-release fertiliser comprising at least one hydrolysis step to provide hydrolysed organic waste and one setting step to provide a slow release fertiliser.

The at least one hydrolysis step may comprise an acid hydrolysis step to provide an acidified hydrolysed organic waste.

The at least one hydrolysis step may comprise an acid hydrolysis step and a base hydrolysis step to produce an acidified hydrolysed organic waste.

The at least one hydrolysis step may comprise an acid hydrolysis step using any acid, for example sulphuric acid, nitric acid and/or phosphoric acid to hydrolyse organic waste and produce an acidified hydrolysed organic waste. The at least one hydrolysis step may comprise an acid hydrolysis step using sulphuric acid, to hydrolyse organic waste and produce an acidified hydrolysed organic waste.

The method may comprise an optional acidification step to make the pH of the acidified organic waste below pH 7.

The acidified hydrolysed organic waste may comprise sulphuric acid, phosphoric acid and nitric acid.

The method may comprise a step of setting the acidified hydrolysed organic waste mixture by combining acidic, hydrolysed organic waste with a setting mixture or a liming mixture. This may be any compound or composition that can form a lime mortar. The organic waste may be bonded to the lime mortar or embedded in it. The liming mixture may comprise, for example lime, calcium sulphate CaSO_4 , Calcium carbonate CaCO_3 , Calcium oxide CaO and/or Calcium hydroxide Ca(OH)_2 .

The liming mixture may further comprise clay bases materials such as kaolin.

The liming mixture may set to provide a lime mortar. During the setting step the organic waste and any additives that have been added to it may become bonded to the lime mortar or set within the lime mortar to provide a slow release fertiliser.

The lime mortar may degrade when left exposed to the elements or in the soil over a period of 1 to 6 months, or over a period of more than 9 months, or over a period of 3 to 9 months, or over a period of 3 to 6 months, or over a period of 3 to 5 months or over a period of 4 to 5 months. The lime mortar may degrade more quickly in soil that is more acidic.

The composition of the setting mixture or liming mixture may be adjusted to provide a slow release fertiliser that degrades faster or slower depending on what is required. In order to adjust the time that the fertiliser takes to degrade different proportions of each of the liming mixture ingredients may be added to the acidified organic waste. For example, the liming mixture can be adjusted so that more calcium sulphate is produced in the setting step to provide a fertiliser that breaks down more slowly. More calcium sulphate may be produced by reducing the amount of lime in the liming mixture so that the excess sulphuric acid reacts with the lime to make more calcium sulphate. The liming mixture can be adjusted so that it degrades faster by including more lime in the liming mixture.

The specific conditions where the slow release fertiliser will be used may be considered when selecting the composition of the organic waste and the composition of the liming mixture. The specific conditions may be the pH of the soil, the expected rain fall, the type of crop, the nutrient requirements of the crop, the nutrient profile of the soil, the ambient temperature that the crop will grow at, the length of growing time of the crop and/or the rate at which the crop requires nutrients during particular phases of its growth.

The percentage of liming mixture compared to acidified organic waste may be adjusted to provide a fertiliser that degrades faster or slower. The percentage of liming mixture used to set the acidified organic waste may be 10% to 50%, may be 20% to 50%, or may be 30% to 50%, or may be 40% to 50% of the mass of the organic waste.

The percentage of liming mixture added compared to the amount of acidified organic waste may be adjusted to provide a fertiliser with a higher or lower neutralizing value. Neutralizing value may be the amount of base compounds in the fertiliser. For example, a higher neutralizing value may be required to partially neutralize acid soil or to provide lime to crops that prefer alkaline conditions. The percentage of liming mixture used to set the acidified organic waste may be 10% to 50% of the mass of the organic waste.

Lime mortar may comprise lime, calcium sulphate CaSO_4 , Calcium carbonate CaCO_3 , Calcium oxide CaO and/or Calcium hydroxide Ca(OH)_2 , which react with water and/or acid and set to form solid lime mortar. The liming mixture may further comprise clay bases materials such as kaolin to change the properties of the lime mortar. Other components of the hydrolysed organic waste may become part of the lime mortar by chemically bonding to components of the lime mortar or by being immobilised inside the lime mortar but not chemically bonded to it. As the lime mortar degrades in soil it may release lime into the soil and may also release components of the organic waste into the soil which provide nutrients and act as a fertiliser.

The setting step may comprise admixing the acidic mixture with lime and/or one or more calcium compounds wherein the calcium compounds are able to react with water and/or acid to form lime mortar which sets the mixture to form a slow release fertiliser. One or more calcium compounds within the liming mixture may react with sulphuric acid to form calcium sulphate which sets the mixture to form part of the lime mortar of the slow release fertiliser.

The lime mortar may set the mixture to form a solid slow release fertiliser with the components of the acidic mixture bonded inside the lime mortar matrix.

The setting step may comprise admixing the acidic hydrolysed organic waste mixture with one or more selected from calcium oxide, calcium carbonate and calcium hydroxide to form a slow release fertiliser.

The setting step may comprise admixing the acidic mixture with calcium oxide and calcium carbonate to form a slow release fertiliser.

The setting step may comprise admixing the acidic mixture with calcium oxide, and calcium hydroxide to form a slow release fertiliser.

The setting step may comprise admixing the acidic mixture with calcium carbonate and calcium hydroxide to form a slow release fertiliser.

The proportions of acidic mixture, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product is slightly alkali.

The proportions of acidic mixture, lime, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product takes more than a month, more than two months, more than 3 months, more than 4 months, more than 5 months or more than 6 months to degrade in soil. The soil may be between pH 5 and pH7. The soil may be between pH 5 and pH 9.

The proportions of acidic mixture, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product takes between one and 6 months, between 1 and 4 months or between 2 and 3 months to degrade in slightly acidic soil. For example the soil may be between pH 5 and pH7.

The method may comprise the step of adding one or more minerals to provide suitable nutrients in the slow release fertiliser.

The method may comprise the step of adding one or more fungicides to provide antifungal properties to the slow release fertiliser.

The method may comprise the step of adding one or more pesticides to provide pesticide properties to the slow release fertiliser.

The method may comprise the step of adding one or more herbicides to provide herbicide properties to the slow release fertiliser.

The method may comprise the step of adding one or more unpalatable compounds deter animals from eating the slow release fertiliser.

The method may comprise the step of processing the slow release fertiliser, for example by pelleting, granulating, press-forming or powderising the slow release fertiliser.

The method may comprise the step of coating the fertiliser, for example coating pellets or granules of the fertiliser, with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds.

In a second aspect the present invention provides a slow release fertiliser provided by the method of the present invention.

In a further aspect the present invention provides a slow release fertiliser comprising a proportion of lime mortar to set the fertiliser.

The slow release fertiliser may comprise more than 10%, more than 20%, more than 30%, more than 40%, more than 50%, more than 60% or more than 70% lime mortar.

The slow release fertiliser may comprise between 10% and 70%, between 20% and 50% or between 30% and 40% lime mortar.

The slow release fertiliser may comprise lime mortar that takes between one and 6 months, between 1 and 4 months or between 2 and 3 months to degrade in slightly acidic soil. For example the soil may be between pH 5 and pH 7.

The slow release fertiliser may further comprise one or more fungicides, one or more pesticides, one or more herbicides, one or more pest control agents and/or one or more unpalatable compounds. Pellets of the fertiliser may be coated with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds.

In a further aspect the present invention provides a composition obtained or obtainable by the method of the invention.

In a further aspect the present invention provides a fertiliser composition obtained or obtainable by the method of the present invention.

There now follows by way of example only a detailed description of the present invention with reference to the accompanying drawings, in which;

FIG. 1 is a schematic view of an apparatus for processing organic waste;

FIG. 2 is a cut-away side view of a heating vessel forming part of the plant of **FIG. 1**;

FIG. 3 is a plan view of the vessel of **FIG. 2**;

FIG. 4 is a cut-away side view of a heating vessel of **FIG. 2** with further components shown; and

FIG. 5 is a plan view of the vessel of **FIG. 4**;

FIG. 6 shows a beaker containing 95 g of congealed frozen blood and 30 g of water;

FIG. 7 shows a view of the hydrolysis and heating stage being performed on 95 g of blood with 30 g of water. 10 g of potassium Hydroxide pellets have been added and the mixture was stirred for two minutes to form a deep red solution. This reaction is very exothermic and so hydrolysis and heating happen at the same time;

FIG. 8 shows the mixture of **FIG. 7** with 8 g of 70% (1.42 specific gravity) nitric acid added to neutralize the potassium hydroxide this step is also very exothermic and is the neutralization step and a heating step at the same time. The resulting mixture is an alkaline solution of potassium nitrate containing dissolved hydrolysed blood;

FIG. 9 shows a view of the solid set product that results when the mixture shown in **FIG. 8** is combined with $\text{CaSO}_4 \cdot \frac{1}{2}\text{H}_2\text{O}$ (left hand side) or with CaO , CaCO_3 and $\text{Ca}(\text{OH})_2$;

FIG. 10 shows a view of 63 g of chicken litter and bedding;

FIG. 11 shows a view of the 63 g of chicken litter and bedding mixed with 150 ml and blended to form a workable paste;

FIG. 12 shows a view of the blended mixture shown in **FIG. 11** with 3 g (0.0306 mole) of H_3PO_4 (orthophosphoric acid (1.75 specific gravity) added and the mixture brought to the boil. This is the heating stage and a first hydrolysis stage (acid hydrolysis) done at the same time;

FIG. 13 shows a view of the mixture in **FIG. 12** with 0.0918 mol of potassium hydroxide (5.14 g) added to neutralize the acid, then 5 g more potassium hydroxide added to increase the pH to 14 and hydrolyse fat, proteins, lipids and nucleic acids, this stage is highly exothermic and is therefore a second hydrolysis step (base hydrolysis) performed at the same time as a heating step;

FIG. 14 shows a view of the resulting mixture after pH of the mixture shown in **FIG. 13** is neutralized using 8 g of 70% (1.42 specific gravity) nitric acid;

FIG. 15 shows a view of the mixture shown in **FIG. 14** after addition of calcium carbonate, calcium hydroxide and calcium oxide to produce a solid pellet; and

FIG. 16 shows a pellet of fertiliser made from chicken litter using a process of the present invention.

The method of the present invention may be carried out in an apparatus for processing organic waste, an example of which is set out below and in **FIGS. 1 to 5**.

Referring to **FIG. 1**, an apparatus for processing organic waste comprises an inlet **10** which receives raw organic waste. A metal detector **12** is provided at the inlet **10** to detect any metal in the raw organic waste and to generate a control signal indicative of any metal content detected. The inlet **10** feeds into a particle sizing module, which in this case is a crushing module **14**, arranged to reduce the raw product to pieces of a suitable size. In this embodiment the crushing module **14** comprises a first crusher arranged to crush the raw organic waste into pieces of less than a predetermined size, said size being 150 mm in this example, and a second crusher arranged to crush the raw organic waste into smaller pieces, of less than 40 mm in this example. A conveyor **16**, which may be a screw conveyor, is arranged to transfer the crushed product from the outlet of the crushing module **14** to the first **18** of three heating vessels **18**, **20**, **22**.

The first heating vessel **18** comprises an enclosed steel container having an inlet **24** at the top through which the crushed product can be introduced, a further inlet **26**, also at the top, which is connected to a water supply via a pump **28** so that water can be added to the product in the vessel **18**, a further inlet **29** connected to a supply of steam, and an outlet **31** at the bottom. Flow control valves are provided in the water and steam inlets **26**, **29**, and the outlet **31**. A heater **30** is provided around the wall of the vessel, and a temperature gauge **32** is arranged to measure the temperature in the vessel **18**. A controller **34** is arranged to control the operation of the whole plant, including all the conveyors, pumps and heaters, and the inlet and outlet valves, as will be described in more detail below.

A first separation vessel **36** is arranged to receive the product from the outlet **31** from the first heating vessel, and is arranged to hold the solid product and allow liquid to drain into a collection tank **38** situated below it. A second conveyor **40** is arranged to transport the solid product from the separation vessel **36** to the inlet **42** of the second heating vessel **20**.

The second heating vessel **20** is pressurized, and has a pressure vent controlled by a vent valve **43** so that the pressure in the vessel **20** can be controlled. It also has a further inlet **44** for water, a further inlet **45** for steam, and an outlet **46** at its bottom end, each with a flow control valve controlled by the controller **34**. It also has a heater **48** around its side walls, and a temperature gauge **50** and pressure gauge **52** arranged to measure the temperature and pressure of the contents of the second vessel **20**. These are connected to the controller **34** which is arranged to control the temperature and pressure, as well as the quantities of water, steam, and product, in the second vessel **20** as required.

A second separation vessel **56** is arranged to receive the product from the outlet **46** from the second heating vessel **20**, and is arranged to hold the solid product and allow liquid to drain into a collection tank **58** situated below it. A third conveyor **60** is arranged to transport the solid product from the separation vessel **56** to a second crusher unit **62**. The second crusher unit **62** is arranged to break down the solid product into pieces no larger than 2 mm.

The outlet from the crusher unit **62** is arranged to output the product into a pan **64**, which may be arranged to pre-heat the product prior to it being received in the third heating vessel **22**. A further conveyor **66** is arranged to transport the product from the pan **64** to the inlet **68** of the third heating vessel **22**.

The third vessel **68** also has a further inlet **70** at its top end connected to a source **72** of acid via a pump **74**, and a further inlet **76** connected to a source **78** of hydrogen peroxide, via pump **80**. These are controlled by the controller **34** so that acid and hydrogen peroxide can be added into the vessel **22** in the required amounts and rates and at the required time. The third vessel also has a heater **82** and a temperature gauge **84** connected to the controller **34**. An outlet **84** at the bottom end of the vessel **22** is connected to a neutralization vessel **86** which is arranged to receive the product from the third heating vessel **22**, and also arranged to receive an alkaline additive, in this embodiment a lime mix, via a conveyor **88**, from an alkaline additive source **90**, under the control of the controller **34**.

The final neutralized product is in the form of a paste and an extruder **92** is arranged to receive this product from the neutralization vessel **86** and extrude it into pellets, and a packaging module **94** is arranged to receive the extruded pellets and package them for transport away from the plant.

Each of the liquid collection vessels **38**, **58** has an outlet that is connected to a separator in the form of a centrifuge **96** which is arranged to separate out the lighter and heavier components of the liquid. The lighter components will generally comprise fats, and the heavier components generally comprise gelatine, and these components are collected in separate collection vessels **98**, **100**.

Operation of the plant will now be described with particular reference to the treatment of bone, although it will be appreciated that it can be used for a wide variety of other waste products.

Raw bone is introduced into the crusher module **14** where it is crushed into pieces no larger than 40 mm. From there it is transferred into the first vessel **18**.

The controller **34** is arranged to monitor the amount of product entering the vessel **18**, for example using a load cell to measure its weight, and to stop the transfer of product when a specified amount has entered the vessel. The controller **34** is also arranged to control the introduction of water and steam into the vessel **18**, and to heat the vessel **18** to the required temperature. The bone and water mixture is heated to approximately 95° C. to form an organic waste slurry which is then transferred to the separation vessel **36**.

In the separation vessel **36**, liquid drains from the slurry into the collection tank **38** while the slurry is transferred along the vessel. The drier slurry is then transferred via the conveyor **40** to the second heating vessel **20**. The amount of slurry introduced into the second vessel **20** is controlled by the controller, for example using weighing cells to weigh the amount added, and then controlled amounts of water and steam are added by controlled operation of the inlet valves. The second vessel **20** is then heated up to about 135° C., and the vent valve **43** closed so that the pressure increases to about 5 bar. This temperature and pressure is maintained for about twenty minutes.

At the end of the treatment in the second vessel **20**, the vent valve **43** is opened to reduce the pressure in the vessel, and the outlet **46** is opened to allow the product, which is still in the form of a slurry, to empty into the second separation vessel **56**. From that vessel **56**, liquid drains into the second collection vessel **58**, and the solid component of the slurry is transferred via conveyor **60** into the second crusher unit **62**, where it is crushed to a particle size of about 2 mm and transferred to the pre-heating vessel **64**, where it is pre-heated.

From the pre-heating vessel **64** the solid product is transferred via conveyor **66** into the third heating vessel **22** via the conveyor **66** and the inlet **68**. Again the amount of product introduced is controlled, and also sulphuric acid and hydrogen peroxide are added in controlled amounts via the inlets **70**, **76**. The reaction in the vessel **22** is exothermic and the temperature rises to about 95° C., and then falls off when the reaction is complete. This heat and the acid and hydrogen peroxide kills any disease or bacteria in the product, and also oxidises any carbon in the product resulting in an inorganic product which is suitable for use as a fertiliser.

When the acidification step is complete, the outlet **22** from the third vessel **22** is opened and the product transferred to the neutralization vessel **86**, to which lime mix is added in controlled quantities in order to increase the pH of the mixture to neutralize the product. The lime mix may comprise a mixture of chalk and lime, or quick lime for example. The lime mix may also comprise additives to alter the properties of the mix, such additives for example comprising selected macro/micro nutrients. The lime mix, and/or the additives, can be varied as required to deliver the required fertilizer properties. The lime mix causes an exothermic reaction. This serves to at least partially dry the acidified organic waste.

The increased pH mixture, which is typically in the form of a paste, is then dried if necessary, and passed into the extruder **92**, from which it is extruded as pellets or granules, and packaged at packaging module **94** for transport.

The liquid from the collection vessels **38**, **58** is separated in the centrifuge **96**, with fat being transferred to fat storage tank **98** and gelatine to the gelatine storage tank **100**.

The third heating vessel **22** may also comprise additive inlets arranged to deliver further additives to the slurry. These additives can be of any suitable ingredient to adjust the mineral or nutritional content of the solid slurry particles to suit a particular use or location of use. Thus the additives can be controlled to alter the macro/micro nutrient bases of the solid slurry particles to suit different soil or ground conditions for example, or to provide optimum nutrition for a particular type of crop.

Referring to **FIGS. 2 and 3**, the first heating vessel **18** comprises a cylindrical steel body **200**, about 1.25 m high and 750 mm in diameter, arranged so that its central axis **202** and curved side walls **204** extend vertically. A pair of part-helical heat transfer blades **206** are mounted by means of cross members **208** on a central rotatable shaft **210** which is located on the central axis **202**. Each of the heat transfer blades **206** is formed of a strip of sheet metal **209** about 15 mm wide, which is formed into a part-helix shape extending around the outer part of the vessel close to the side wall **204** but spaced slightly from it. The blades each extend from a point about two thirds of the way up the side of the vessel, down to close to the bottom of the vessel, and through half a turn around the central axis. The two blades **206** are diametrically opposite each other, so that they extend around opposite sides of the vessel.

A pair of scrapers **212** are also supported on the cross members **208**. Each scraper **212** comprises a flat scraper blade **214** extending vertically down the side wall **204** of the vessel with an outer scraping edge **216** just clear of the side wall **204**. Each of the scraper blades **214** is supported on a set of support brackets **218**, which in turn are supported on a vertical support pole **220** which extends vertically between the cross members **208** near their outer ends.

Referring to **FIGS. 4 and 5**, an inner helical lifting flight **222** is also mounted on the shaft **210**. The lifting flight **222** is formed of a flat strip of metal **224** about 100 mm wide, formed into a helix centred on the shaft **210**, and turning through about seven turns. The top of the lifting flight **222** is about level with the top of the heat transfer blades **206** and the bottom of the lifting flight **222** is slightly above the bottom of the heat transfer blades **206**. The turning sense of the lifting flight **222** is the same as both of the heat transfer blades **206**. A motor **224** is arranged to rotate the shaft **210**, so that the lower end of the lifting flight forms a leading end and the upper end of the lifting flight forms a trailing end, and the lifting flight provides lift in the central part of the vessel **18** close to the central axis **202**. At the same time, the lower ends of the heat transfer blades **206** form leading ends and the upper ends of the heat transfer blades form trailing ends, and the heat transfer blades also provide a small amount of lift, or in fact resistance to downward flow, in the outer part of the vessel closer to the outer wall **204**. As the shaft rotates, the scrapers **212** move around the wall **204** of the vessel, scraping from the wall any of the by-product or other contents of the vessel that may have accumulated on the wall.

The central shaft **210** is rotated as the product is introduced into the vessel **18** through the inlet **24**, the water is introduced through the inlet **26**, this forms a slurry, and the steam is introduced through the inlet **28** which helps to heat the slurry. As the heater **30** heats up the contents of the vessel **18** and the steam is injected into it, the central lifting flight **222** causes a general upward flow of the mixture in the central area of the vessel. The mixture then moves outwards at the top of the vessel **18** and cascades downwards in the outer region of the vessel **18** close to the wall **204**. The rotating heat exchange blades **206** provide some resistance to the downward flow, and therefore mix the mixture, but also help to conduct heat through the mixture quickly allowing it to be heated quickly to the required temperature. This helps to reduce the time required for the heating step carried out in the first heating vessel **18**.

The second vessel **20** has the same mixing and heat exchange mechanism as the first vessel **18** as shown in **FIGS. 2 to 5**. In this vessel the heating process is similar to the first vessel, though at increased pressure, and the heat exchange coils and the lifting flight serve to distribute heat from the heater and steam throughout the vessel in a similar way. Similarly the third vessel **22** also has the same mixing and heat exchange mechanisms as shown in **FIGS. 2 to 5**. In this case the heat is generated by the exothermic reaction in the vessel. However, rapid heat dissipation is still helpful to ensure that the temperature remains approximately equal throughout the vessel, and that the reaction proceeds at about the same rate throughout the vessel.

It will be appreciated that various modifications can be made to the embodiment described above, and that the design of the heating vessel or vessels can be varied as appropriate for the process and products to be treated. For example, the two heat transfer blades could be replaced by one continuous helical coil, or more than two blades. Also the blades may not be strictly part-helical. For example the angle of the blade or blades to the vertical may vary along its length. Similarly the lifting flight may be of a different shape, or may be rotated independently of the heat transfer coils. Indeed a different type of lifting mechanism may be provided which is not a helical flight.

A number of pilot studies, described in the examples below, were performed to assess the effectiveness of different compositions for use in the hydrolysis step of the present method. The hydrolysis step may take place in one of the reaction vessels, for example vessel **[22]** in **FIG. 1**. The pilot studies were mostly done on a small scale but may be scaled up and the method may be performed on large quantities of organic waste, for example in an apparatus as shown in **FIG. 1**. The inlets for acid and hydrogen peroxide described in relation to **FIG. 1** may alternatively be used to add other ingredients for the hydrolysis and neutralization steps.

Example 1 Acid and Hydrogen Peroxide

Bone Paste

The preferred ranges of the temperature and reaction times.

Operating temperature range: Start temp of bone paste 10° C. to 100° C. giving an end temp after the hydrolysis step of 100° C. to 200° C.

Preferred temp for process: Start temperature 50-90° C.-end temp 130-170° C.

Ratio of wet bone paste to acid to hydrogen peroxide by mass 5-20:5:1 [wet bone:acid:peroxide], for example 13.5:5:1 [wet bone:acid:peroxide].

For example the ratio may be: 1000 kg bone paste [wet] to 370 kg of 95% sulphuric acid to 74 kg of 30% hydrogen peroxide.

Treatment time Broad range—1 min to 30 mins, for example, the treatment time may be 5-10 mins.

The amounts of acid and hydrogen peroxide—Broad Range—1:1 to 1:10 Concentrated sulphuric acid to hydrogen peroxide solution.

Preferred ratio is 1:5 concentrated sulphuric acid to hydrogen peroxide for bone digestion but will vary for other waste streams.

Concentrated sulphuric acid used is 95% (1.83 g/ml) but this could work over a range of 50-100%.

Concentrated hydrogen peroxide used is 30% w/v (weight to volume) but concentrations between 5-100% w/v although greater than 30% can carries an explosion risk.

Example 2 Base and Hydrogen Peroxide

Base Peroxide

Broad range of concentrated ammonium hydroxide, and/or potassium hydroxide and/or sodium with hydrogen peroxide from 1:1 to 1:10 using 0.880 sg (specific gravity) ammonium hydroxide [ammonia solution] and 30% hydrogen peroxide solution.

Preferred mixture 3:1 of 0.880 sg ammonia solution to 30% hydrogen peroxide solution for the purpose of dissolving for example poultry litter, pig manure, cow slurry, cow manure, horse manure, contents of intestinal tract material from abattoir waste etc.

Example 3 Hydrolysis of Blood

Overview—blood contents are hydrolysed using sodium hydroxide or potassium hydroxide. The resulting solution is neutralized using nitric acid or sulfuric acid forming a potassium nitrate, sodium nitrate, sodium sulphate or potassium sulphate solution depending on the acid and base used solution containing the dissolved, heat treated, hydrolysed blood components.

Method:

To 100 g of congealed blood add 30 g of water and bring to the boil (water not required if blood is fresh and uncongealed). NB—frozen blood used can be seen melting in **FIG. 1**.

Hydrolysis: Add 10 g of Potassium Hydroxide pellets KOH (solid, form preferred) or 7 g of sodium hydroxide [solid form preferred] and stir for two minutes until deep red solution forms—very exothermic reaction. The product of this reaction is shown in **FIG. 2**.

Neutralisation: Add 16 g of 70% 1.42 specific gravity nitric acid to neutralize the potassium hydroxide (concentrated) although a dilute nitric acid solution would suffice—very exothermic. The resulting mixture is shown in **FIG. 3**.

The resultant mixture is then treated with further nitric acid to achieve the desired pH for granulating—this could be acidic or alkaline.

Solution of potassium nitrate containing dissolved, hydrolysed blood. If a small amount of solid precipitates from the blood it may be separated and hydrolysed/oxidised by a small, additional dose of nitric [or sulphuric] acid and hydrogen peroxide in a ratio of 5:1 before recombining with original solution for pasting or setting.

The neutralized hydrolysed blood may be blended with other soluble ingredients and the resultant solution sold as a liquid fertiliser or the neutralized hydrolysed blood may be granulated and sold as a solid granular form after setting using plaster of paris [CaSO₄.½H₂O] **FIG. 4** left or Calcium carbonate/oxide/hydroxide CaCO₃/CaO/Ca(OH)₂ **FIG. 4** right.

Example 4

Acid hydrolysis using phosphoric acid (other acids could be used—nitric acid or sulphuric acid) followed by alkaline hydrolysis using potassium hydroxide or sodium hydroxide—resultant mixture neutralized using nitric acid or sulphuric acid.

Phosphoric acid increases the phosphorus content (P) and hydrolyses cellulose content. Nitric acid increases the N content.

Potassium hydroxide increases the K content and hydrolyses bulk content. Nitric acid neutralizes the potassium hydroxide or sodium hydroxide forming potassium nitrate or sodium nitrate and increases the N content.

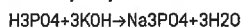
Approximate ratios by mass—42 (chicken litter):100 (water):2 (phosphoric acid):6.6 (potassium hydroxide) or 4.7 sodium hydroxide:8.0 (nitric acid).

Overview 63 g of chicken litter and bedding is shown in **FIG. 5**.

Addition of 150 ml of water followed by 'blending' to form a suitable, workable paste shown in **FIG. 6**.

Acid hydrolysis:—high-heat stage of process. 3 g (0.0306 mole) of H₃PO₄ (Orthophosphoric acid 1.75 specific gravity) or 2.75 g 70% nitric acid is added to drop pH to pH 3-4 and mixture brought to the boil (high heat stage of process).

Alkaline hydrolysis—in heliotherm stage. First the phosphoric acid is neutralized by potassium hydroxide:



3.00 g of phosphoric acid (0.0306 mol) requires 0.0918 mol of potassium hydroxide=5.14 g then an addition 5 g of KOH is required to facilitate the increase in pH to 14 and hydrolyse fat/protein/lipids/nucleic acids etc (total 10 g of KOH)—lots of heat evolved NaOH could also be used (less required —7.14 g).

Neutralisation—still in heliotherm stage.

Neutralisation using 17 g of 70% (1.42 specific gravity) nitric acid.

Resultant mixture is an alkaline, sweet (molasses type smell) brown paste containing potassium nitrate, potassium phosphate and all the components of chicken litter suitable for plant uptake.

Addition of calcium carbonate/hydroxide/oxide/calcium sulphate heptahydrate mix to produce the solid pellet form is shown in **FIG. 10**.

Sodium hydroxide is useful for the hydrolysis of blood or chicken litter followed by neutralization using sulphuric acid. For example 100 g of blood requires 5 g of sodium hydroxide at around 90° C. to hydrolyse, followed by neutralization using concentrated sulphuric acid 3.6 ml (6.83 g) and then adjustment of pH to desired level.

The present invention may be further described in the following numbered paragraphs:

1. A method of processing organic waste comprising one or more step of combining the organic waste with a hydrolysing composition.
2. The method according to paragraph 1 further comprising a heating step.
3. The method according to paragraph 2 wherein the organic waste is heated to between 70° C. and 200° C.
4. The method according to paragraph 2 or paragraph 3 wherein the organic waste is heated for between 10 and 60 minutes.
5. The method according to any one of the preceding paragraphs further comprising the step of removing liquid comprising fat and/or gelatine from the organic waste.
6. The method according to any one of the preceding paragraphs wherein the hydrolysing composition comprises: an acid and hydrogen peroxide; a strong base and hydrogen peroxide; a base; an acid; or hydrogen peroxide.
7. The method according to any one of the preceding paragraphs further comprising a sizing step wherein the organic waste is sized to pieces less than 50 mm.
8. The method according to any one of the preceding paragraphs further comprising a neutralization step, wherein acid or base is added to the mixture to bring the pH near to neutral.
9. The method according to any one of the preceding paragraphs further comprising a step of adding minerals or nutrients to the mixture.
10. The method according to any one of the preceding paragraphs further comprising drying the organic waste mixture.
11. The method according to any one of the preceding paragraphs further comprising adding a setting agent such as calcium carbonate, calcium hydroxide, calcium sulphate hemihydrate and/or calcium oxide to the organic waste in a quantity suitable to set the mixture.
12. The method according to any one of the preceding paragraphs further comprising the step of re-sizing the set organic waste into pieces of less than 40 mm.
13. The method according to any one of the preceding paragraphs further comprising adding one or more unpalatable ingredients to the organic waste mixture.
14. The method according to any one of the preceding paragraphs further comprising a step of pelleting or granulating the organic waste mixture.
15. The method according to any one of the preceding paragraphs wherein the method of processing organic waste is a method of producing fertiliser.
16. An organic waste composition obtained by the method according to any one of the preceding paragraphs.
17. A fertiliser composition obtained by the method of any one of the preceding paragraphs.
1. A method of processing organic waste comprising one or more step of combining the organic waste with a hydrolysing composition.
2. The method according to paragraph 1 further comprising a heating step.
3. The method according to paragraph 2 wherein the organic waste is heated to between 70° C. and 200° C.
4. The method according to paragraph 2 or paragraph 3 wherein the organic waste is heated for between 10 and 60 minutes.
5. The method according to any one of the preceding paragraphs further comprising the step of removing liquid comprising fat and/or gelatine from the organic waste.

6. The method according to any one of the preceding paragraphs wherein the hydrolysing composition comprises: an acid and hydrogen peroxide; a strong base and hydrogen peroxide; a base; an acid; or hydrogen peroxide.
7. The method according to any one of the preceding paragraphs further comprising a sizing step wherein the organic waste is sized to pieces less than 50 mm.
8. The method according to any one of the preceding paragraphs further comprising a neutralization step, wherein acid or base is added to the mixture to bring the pH near to neutral.
9. The method according to any one of the preceding paragraphs further comprising a step of adding minerals or nutrients to the mixture.
10. The method according to any one of the preceding paragraphs further comprising drying the organic waste mixture.
11. The method according to any one of the preceding paragraphs further comprising adding a setting agent such as calcium carbonate, calcium hydroxide, calcium sulphate hemihydrate and/or calcium oxide to the organic waste in a quantity suitable to set the mixture.
12. The method according to any one of the preceding paragraphs further comprising the step of re-sizing the set organic waste into pieces of less than 40 mm.
13. The method according to any one of the preceding paragraphs further comprising adding one or more unpalatable ingredients to the organic waste mixture.
14. The method according to any one of the preceding paragraphs further comprising a step of pelleting or granulating the organic waste mixture.
15. The method according to any one of the preceding paragraphs wherein the method of processing organic waste is a method of producing fertiliser.
16. An organic waste composition obtained by the method according to any one of the preceding paragraphs.
17. A fertiliser composition obtained by the method of any one of the preceding paragraphs.



2. US20160145164 - ORGANIC WASTE PROCESSING

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[EN]

Claims

1. A method of processing organic waste to provide a slow-release fertiliser comprising at least one digesting step to provide dissolved organic waste and a setting step to provide a slow release fertiliser, wherein said digesting step comprises combining the organic waste with a metal catalyst, concentrated sulfuric acid and concentrated hydrogen peroxide and heating to a temperature between 70 °C. and 200°C., wherein the sulfuric acid and hydrogen peroxide are added to the organic waste in a ratio of 5-20:5:1 organic waste:sulfuric acid:hydrogen peroxide, and wherein said catalyst is titanium dioxide, copper or a copper compound; and, wherein the setting step comprises combining the dissolved organic waste with a liming mixture that can form a lime mortar and comprises calcium carbonate.
2. The method according to **claim 1**, wherein said sulfuric acid has a concentration of to 50 to 100%.
3. The method according to **claim 1**, wherein the liming mixture further comprises lime, calcium sulphate, calcium oxide and/or calcium hydroxide.
4. The method according to **claim 1**, wherein the liming mixture further comprises clay base materials.
5. The method according to **claim 1**, wherein the liming mixture sets to provide a lime mortar with the acidified dissolved organic waste bonded to the lime mortar or set within the lime mortar to provide a slow release fertiliser.
6. The method according to **claim 1**, wherein the liming mixture is composed to provide a lime mortar that degrades, when left exposed to the elements or in the soil, over a period of 1 to 9 months.
7. The method according to **claim 1**, wherein the mass of liming mixture compared to the mass of acidified dissolved organic waste is adjusted to provide a fertiliser that degrades faster or slower and/or to provide a fertiliser with a higher or lower neutralising value.
8. The method according to **claim 7**, wherein the mass of said liming mixture is 10% to 50% of the mass of the acidified dissolved organic waste.
9. The method according to **claim 1** further comprising the step of adding one or more minerals at any stage in the method to provide nutrients in the slow release fertiliser.
10. The method according to **claim 1** further comprising the step of adding one or more fungicides, pesticides, herbicides or unpalatable compounds to provide desirable properties to the slow release fertiliser.
11. The method according to **claim 1** further comprising the step of processing the slow release fertiliser by pelleting, granulating, press-forming or powderising the slow release fertiliser.
12. The method according to **claim 1**, further comprising the step of coating the fertiliser with a coating comprising one or more of bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, or unpalatable compounds.
13. The method of **claim 1**, wherein said hydrogen peroxide has a concentration of 30-100%.
14. The method of **claim 1**, wherein the ratio of sulfuric acid to hydrogen peroxide is between 1:1 and 1:10.
15. The method of **claim 1**, wherein the copper II compound or titanium dioxide is added at 0.1%-0.01% of the weight of organic waste.
16. The method of **claim 1**, wherein the method comprises a sizing step wherein said organic material is crushed, ground or chopped before the digesting step.
17. The method of **claim 1**, wherein the method further comprises a step of concentrating the dissolved organic waste by evaporation where 10-40% of the water is removed by evaporation after the digesting step.
18. The method of **claim 1**, further comprising a drying step to remove water after the setting step.
19. The method of **claim 1**, wherein the pH of the dissolved organic waste is increased by mixing it with chalk or lime.
20. The method according to **claim 4**, wherein the clay base materials include kaolin.

1. AU2014282964

Figure. 2

National Public Date Description

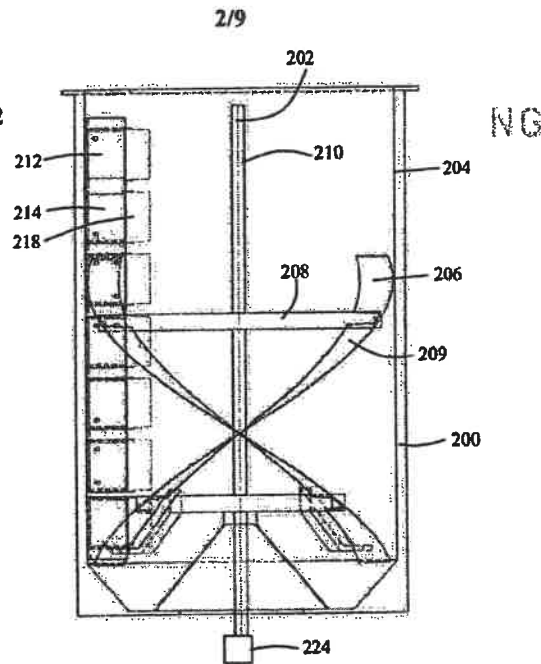


Figure 3

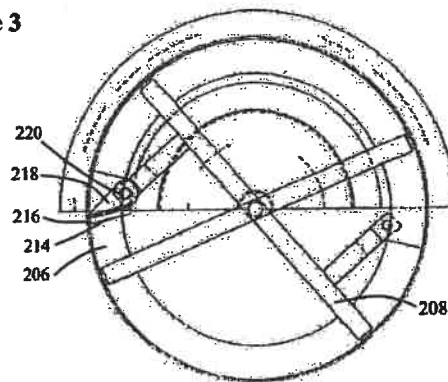


Image 2 of 3



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Figure 4

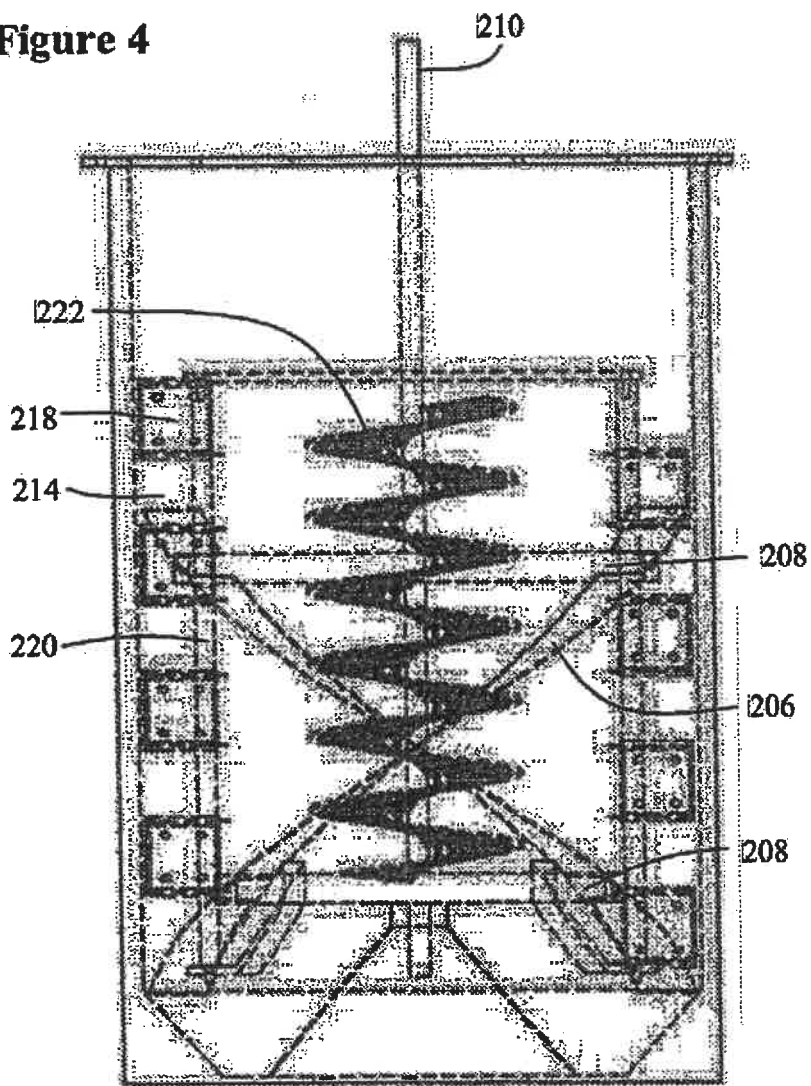
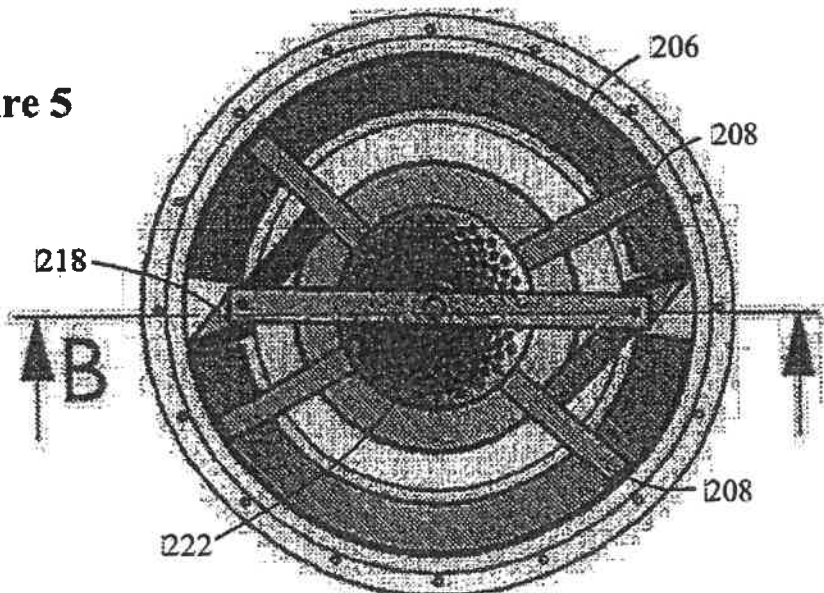


Figure 5



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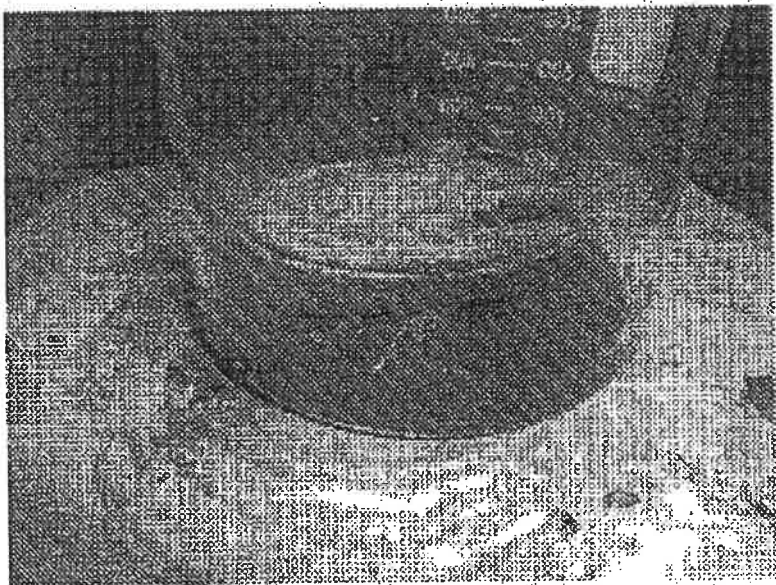


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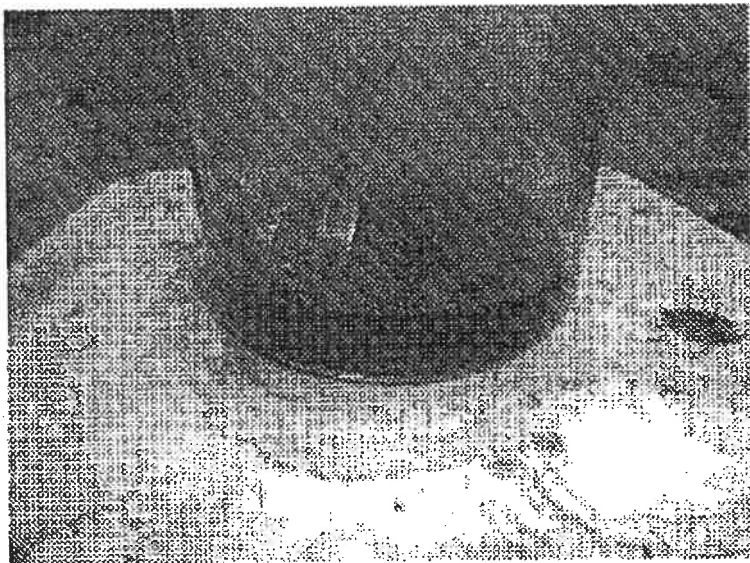


Figure 7

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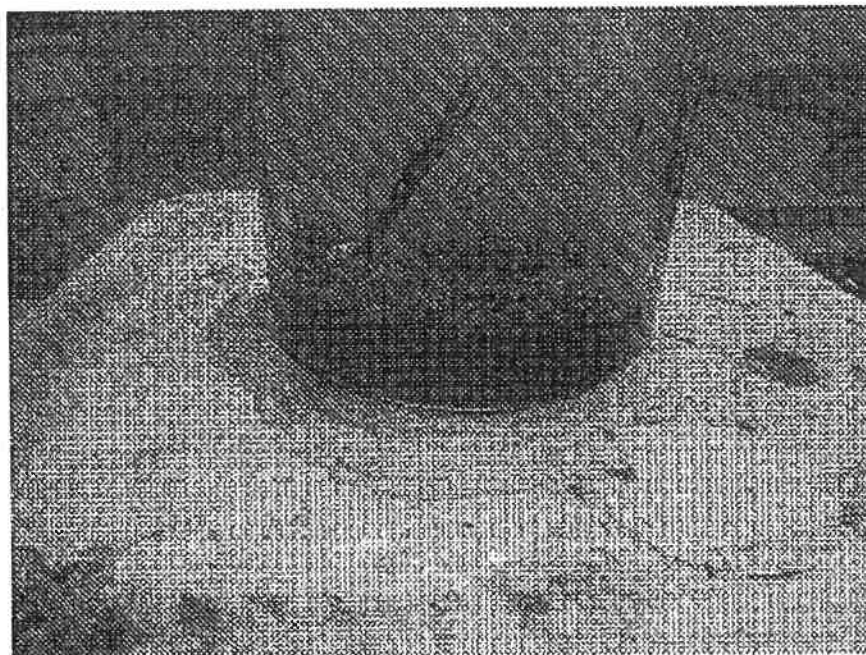


Figure 8

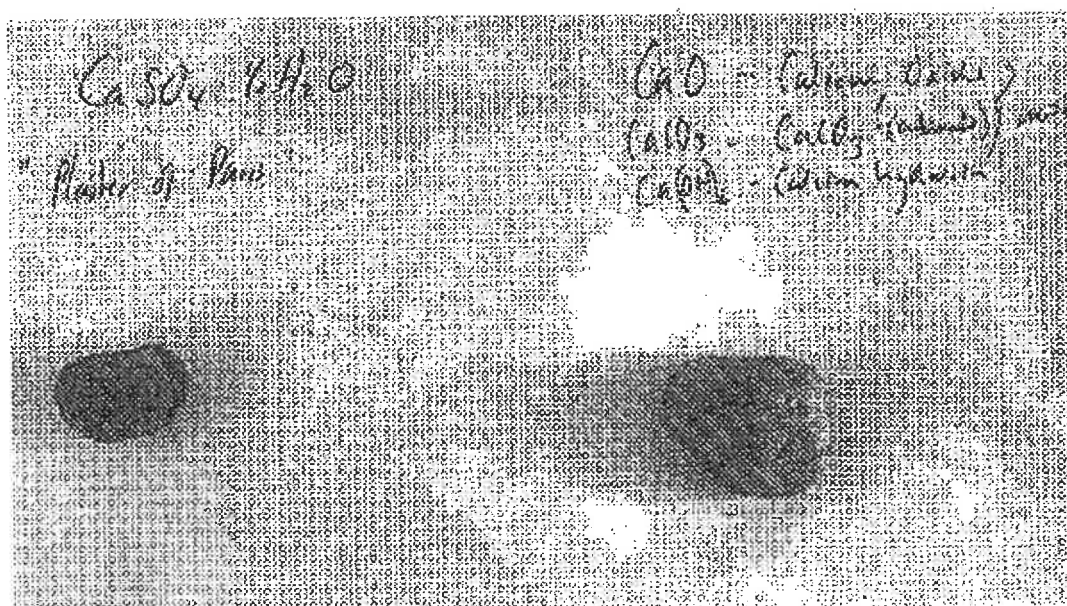


Figure 9

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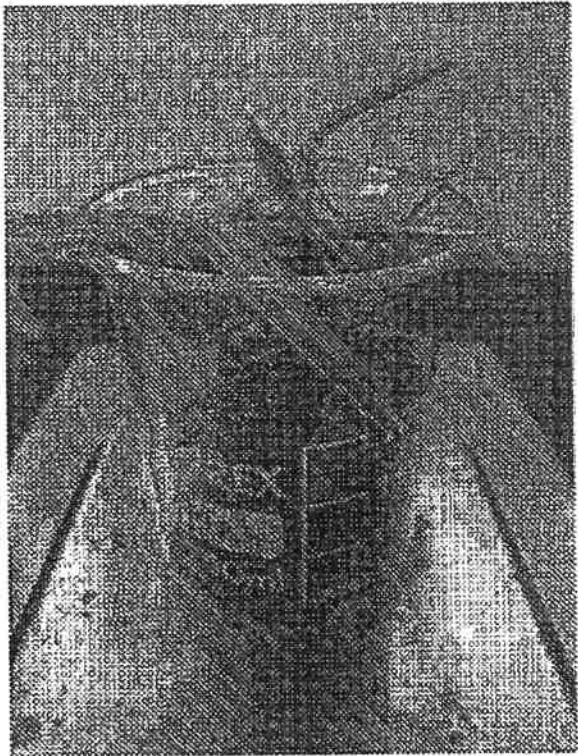


Figure 10

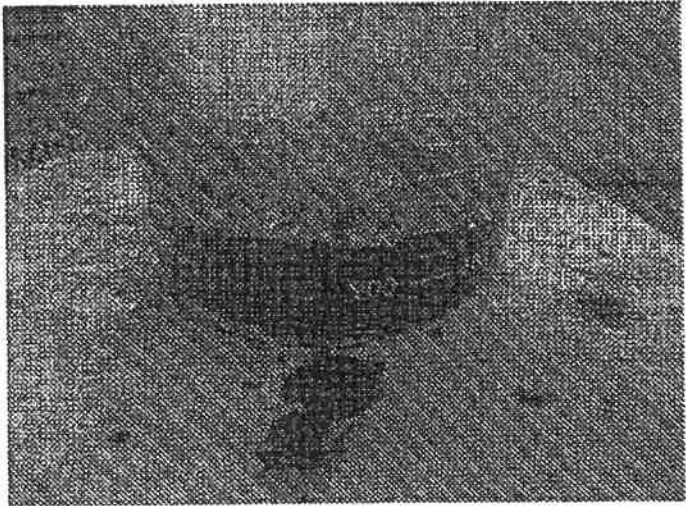


Figure 11

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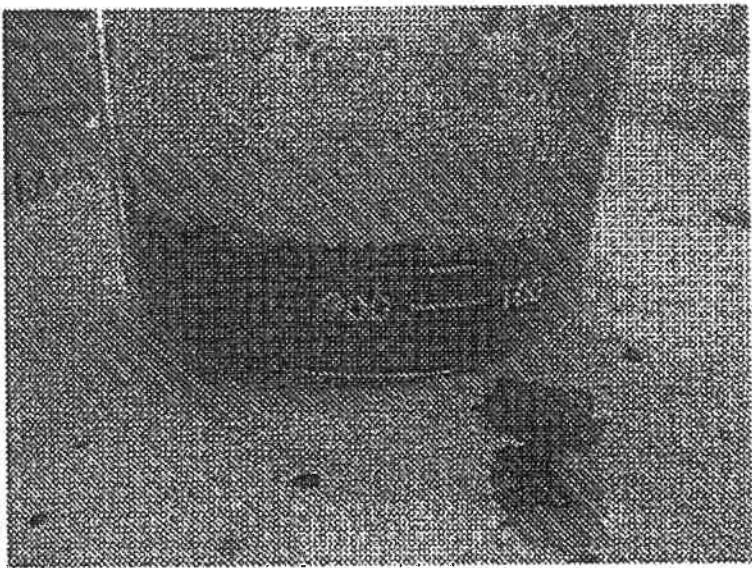


Figure 12

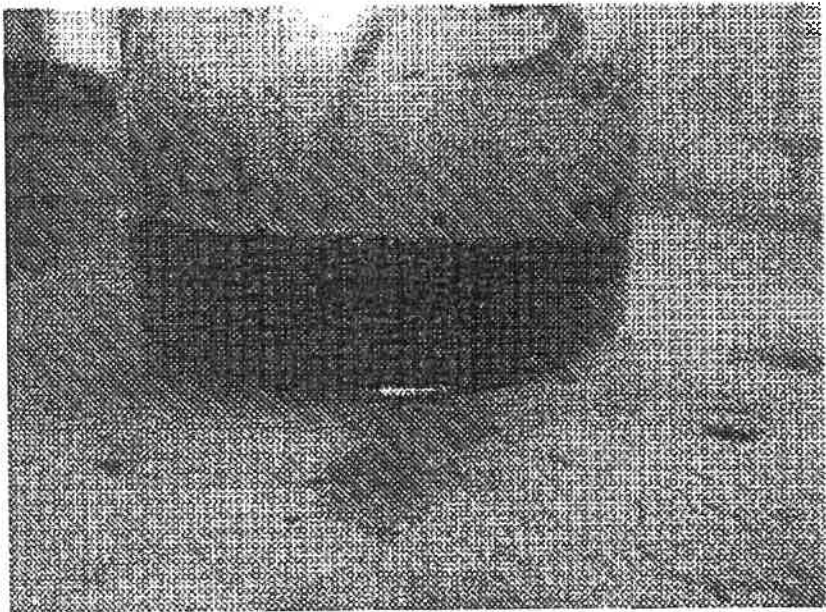


Figure 13

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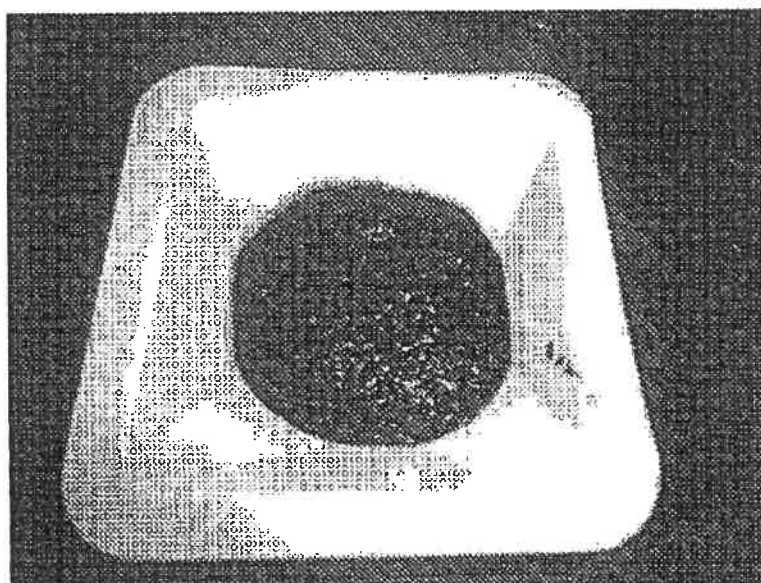


Figure 14

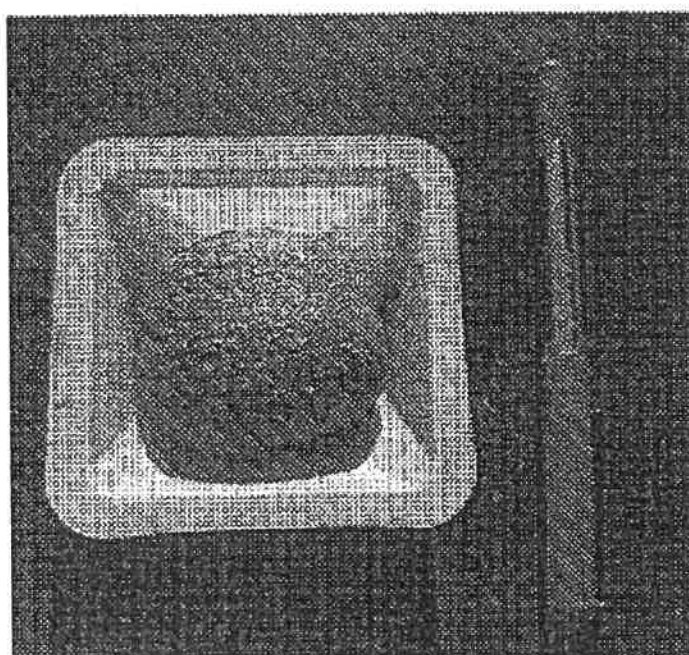


Figure 15

1. AU2014282964

National Biblio Date Description

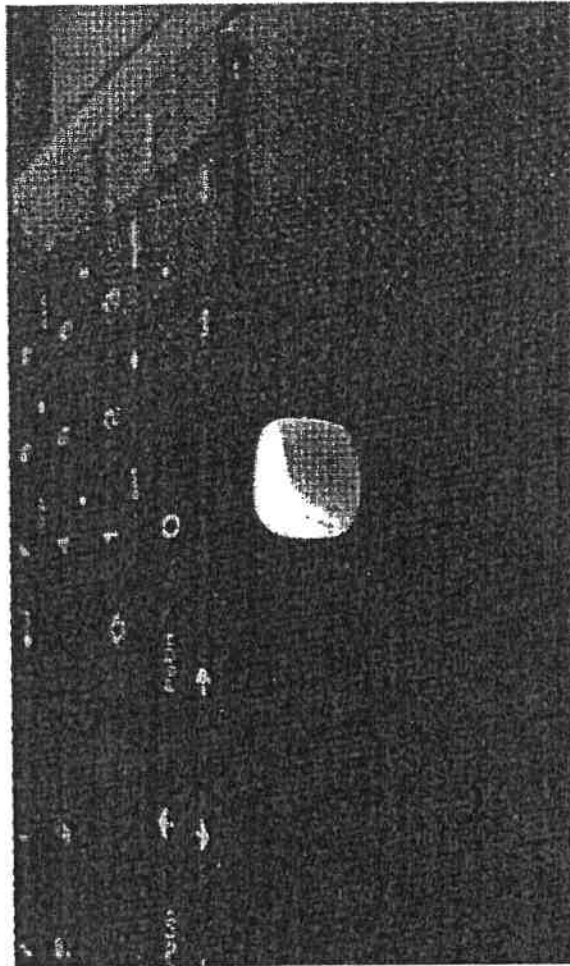


Figure 16

Image 9 of 9



5. NZ716125 - ORGANIC WASTE PROCESSING



National Biblio. Data Patent Family Documents

PermaLink Machine translation

Office

New Zealand

Application Number

716125

Application Date

18.01.2016

Publication Number

716125

Publication Date

28.05.2019

Publication Kind

A

IPC

C05B 17/00C05D 3/02C05F 1/00C05F 3/00

CPC

C05D 3/02C05F 1/005C05F 3/00C05F 11/08C05F 9/00C05F 11/00

View more classifications

Applicants

ELEMENTAL DIGEST LIMITED

Inventors

PALMER, Bruce William

Agents

PHILLIPS ORMONDE FITZPATRICK

Priority Data

1311108.3 21.06.2013 GB1322874.7 23.12.2013 GB

Title

[EN] ORGANIC WASTE PROCESSING

Abstract

[EN] The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertiliser from organic waste, especially slow release fertiliser. The present invention also relates to fertilisers made by the method of the invention, especially slow release fertiliser. The method of processing organic waste to provide a slow-release fertiliser may comprise at least one hydrolysis step to provide hydrolysed organic waste and a setting step to provide a slow release fertiliser.

Related patent documents

MYPI2015704654 MY-179142-A PL3010871 WO/2014/202986 JP2016527168 BR112015032008 TH159506 PH1/2015/502840 GB2531463 AU2014282964 EA201690065 US20160145164 CN105452196 EP3010871 CA2915964 MX366240 IN201627001450 C07590580



MX/A/2015/017871

5. MX366240 - PROCESAMIENTO DE RESIDUOS ORGÁNICOS.

National Biblio. Data

Patent Family

PermaLink

Machine translation

Office

Mexico

Application Number

2015017871

Application Date

18.12.2015

Publication Number

366240

Publication Date

10.11.2016

Grant Number

366240

Grant Date

03.07.2019

Publication Kind

B

IPC

C05B 17/00

CPC

C05D 3/02

C05F 1/005

C05F 3/00

C05F 11/08

C05F 9/00

C05F 11/00

[View more classifications](#)**Applicants**

ELEMENTAL DIGEST LIMITED

Inventors

Bruce William PALMER

Priority Data

1311108.3 21.06.2013 GB

1322874.7 21.06.2013 GB

Title**[EN]** PROCESAMIENTO DE RESIDUOS ORGÁNICOS.**[ES]** PROCESAMIENTO DE RESIDUOS ORGÁNICOS.**Abstract**

[EN] The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertiliser from organic waste, especially slow release fertiliser. The present invention also relates to fertilisers made by the method of the invention, especially slow release fertiliser. The method of processing organic waste to provide a slow-release fertiliser may comprise at least one hydrolysis step to provide hydrolysed organic waste and a setting step to provide a slow release fertiliser.

[ES] La presente invención se refiere a métodos para procesamiento de desechos orgánicos, en particular se refiere a un método para hacer fertilizante como desechos orgánicos, especialmente fertilizante de liberación lenta. La presente invención también se refiere a fertilizantes hechos mediante el método de la invención, especialmente fertilizante de liberación lenta. El método de procesamiento de desechos orgánicos para proporcionar un fertilizante de liberación lenta puede comprender al menos un paso de hidrólisis para proporcionar desechos orgánicos hidrolizados y un paso de ajuste para proporcionar un fertilizante de liberación lenta.

Related patent documents[MYPI 2015704654](#) [MY-179142-A](#) [PL3010871](#) [WO/2014/202986](#) [JP2016527168](#) [BR112015032008](#)[TH159506](#) [PH1/2015/502840](#) [GB2531463](#) [AU2014282964](#) [EA201690065](#) [US20160145164](#)[CN105452196](#) [EP3010871](#) [CA2915964](#) [IN201627001450](#) [NZ716125](#) [C07590580](#)

BR 112015032008-2

1. BR112015032008 - PROCESSAMENTO DE RESÍDUO ORGÂNICO

National Biblio. Data Patent Family

PermaLink Machine translation

Office
Brazil

Application Number
112015032008

Application Date
19.06.2014

Publication Number
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Grant Number
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B1

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C05B 17/00 C05D 3/00 C05F 3/00
C05F 1/00

CPC
C05D 3/02 C05F 1/005 C05F 3/00
C05F 11/08 C05F 9/00 C05F 11/00

View more classifications

Applicants
ELEMENTAL DIGEST LIMITED

Inventors
Bruce William PALMER

Agents
LEONOR MAGALHÃES PERES GALVÃO DE BOTTON

Priority Data
1311108.3 21.06.2013 GB
1322874.7 23.12.2013 GB

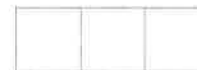
Title
[PT] processamento de resíduo orgânico

Abstract
[PT] processamento de resíduo orgânicoa presente invenção se refere a processamento de resíduo orgânico, em particular ela se refere a um método para produzir fertilizante a partir de resíduo orgânico, especialmente fertilizante de liberação lenta, a presente invenção também se refere a fertilizantes produzidos pelo método da invenção, especialmente fertilizante de liberação lenta, o método de processamento de resíduo orgânico para proporcionar um fertilizante de liberação lenta pode compreender pelo menos uma etapa de hidrólise para proporcionar resíduo orgânico hidrolisado e uma etapa de endurecimento para proporcionar fertilizante de liberação lenta.

Related patent documents
[MYPI2015704654](#) [MY-179142-A](#) [PL3010871](#) [W0/2014/202986](#) [JP2016527168](#) [TH159506](#)
[PH1/2015/502840](#) [GB2531463](#) [AU2014282964](#) [EA201690065](#) [US20160145164](#) [CN105452196](#)
[EP3010871](#) [CA2915964](#) [MX366240](#) [IN201627001450](#) [NZ716125](#) [C07590580](#)

2 915 964

6. CA2915964 - METHOD OF PROCESSING ORGANIC WASTE TO PROVIDE A SLOW-RELEASE FERTILISER


[National Biblio. Data](#)
[Description](#)
[Claims](#)
[Drawings](#)
[Patent Family](#)
[Documents](#)
[PermaLink](#)
[Machine translation](#)
Office

Canada

Application Number

2915964

Application Date

19.06.2014

Publication Number

2915964

Publication Date

24.12.2014

Grant Number**Grant Date**

16.11.2021

Publication Kind

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IPC

C05B 17/00 C05D 3/02 C05F 1/00

C05F 3/00

CPC

C05D 3/02 C05F 1/005 C05F 3/00

C05F 11/08 C05F 9/00 C05F 11/00

[View more classifications](#)**Applicants**

ELEMENTAL DIGEST LIMITED

Inventors

PALMER, BRUCE WILLIAM

Priority Data1311108.3 21.06.2013 GB
1322874.7 23.12.2013 GB**Title****(EN)** METHOD OF PROCESSING ORGANIC WASTE TO PROVIDE A SLOW-RELEASE FERTILISER**(FR)** PROCEDE DE TRAITEMENT DE DECHETS ORGANIQUES DESTINE A FOURNIR UN ENGRAIS A LIBERATION LENTE

Figure 16

Abstract

(EN) The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertiliser from organic waste, especially slow release fertiliser. The present invention also relates to fertilisers made by the method of the invention, especially slow release fertiliser. The method of processing organic waste to provide a slow-release fertiliser may comprise at least one hydrolysis step to provide hydrolysed organic waste and a setting step to provide a slow release fertiliser.

(FR) La présente invention concerne des procédés de traitement de déchets organiques, en particulier, elle concerne un procédé de fabrication d'engrais à partir de déchets organiques, en particulier d'engrais à libération lente. La présente invention concerne également des engrais fabriqués au moyen du procédé de l'invention, en particulier un engrais à libération lente. Le procédé de traitement de déchets organiques destiné à fournir un engrais à libération lente peut comprendre au moins une étape d'hydrolyse pour fournir des déchets organiques hydrolysés et une étape de solidification pour fournir un engrais à libération lente.

Related patent documents

[MYPI2015704654](#) [MY-179142-A](#) [PL3010871](#) [WO/2014/202986](#) [JP2016527168](#) [BR112015032008](#)
[TH159506](#) [PH1/2015/502840](#) [GB2531463](#) [AU2014282964](#) [EA201690065](#) [US20160145164](#)
[CN105452196](#) [EP3010871](#) [MX366240](#) [IN201627001450](#) [NZ716125](#) [C07590580](#)



6. CA2915964 - METHOD OF PROCESSING ORGANIC WASTE TO PROVIDE A SLOW-RELEASE FERTILISER



National Biblio. Data Description Claims Drawings Patent Family Documents

PermaLink

Machine translation

Note: Text based on automatic Optical Character Recognition processes. Please use the PDF version for legal matters

[EN]

METHOD OF PROCESSING ORGANIC WASTE TO PROVIDE A SLOWRELEASE FERTILISER

Field of the Invention

The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertiliser from organic waste, especially slow release fertiliser. The present invention also relates to fertilisers made by the method of the invention, especially slow release fertiliser.

Background to the Invention

Organic waste may be animal by-products, vegetable waste, food waste or a mixture of both.

During the processing of animals for meat production or other uses, relatively large amounts of the animal are removed, discarded and not sold on to an end user. Such discarded byproducts include animal by-products such as bone, blood, gut content, wool, fur and feathers for example. Animal by-products also include animal waste such as chickenlitter, cow slurry and pig or horse manure. Vegetable waste may include kitchen waste, bedding from animals, fruit or vegetable processing waste, for example, fruit peels, sugar cane waste. This represents a significant waste of the resources that could be harnessed for useful purposes. There is also a cost involved in storing and disposing of such by-products as waste.

Various processes are known for processing organic waste. It is known to use incineration and composting to process the waste into a usable form or extract energy from waste.

Date recue/Date Received 2020-08-28

Organic waste comprises valuable nutrients and there is a need for methods of processing organic waste to provide useful products. There is also a need for and environmentally friendly alternatives to conventional fertiliser for crops.

Slow release or controlled release fertilisers are particularly valuable because they release nutrients to plants over a period of time, for example several months. This means that there is less danger of toxicity to plants and of fertiliser running off the land before it is taken up by plants. Slow release fertilisers can provide a sustained supply of nutrients through a longer period of the growing season compared to fertilisers that are easily soluble and release nutrients quickly.

It is an aim of the present invention to provide a method for disposing of organic waste and a method for processing organic waste to provide a slow release fertiliser.

The present invention provides a method of processing organic waste comprising the step of mixing the organic waste with a hydrolysing composition.

The method of processing organic waste may be a method of making a slow release fertiliser.

The method may be carried out in a suitable apparatus comprising one or more vessels. The one or more vessels may be equipped with a mechanism for lifting the mixture upwards in a central region of the vessel by means of a lifting mechanism, resisting downward movement of the mixture in an outer region of the vessel by means of a heat transfer member, and transferring heat within the mixture using the heat transfer member. The vessels may also be equipped with a heating mechanism to heat the contents of the vessel.

The heat transfer member may be rotatably mounted in the vessel and may have a leading end and a trailing end. The leading end may be below the trailing end such that rotation of the heat transfer member generates a lifting force acting on the by-product, which may be arranged to resist its downward movement.

The heat transfer member may be formed from a strip of metal. The heat transfer member may be part helical.

The lifting mechanism may comprise a helical lifting member rotatable to lift the by-product. The lifting mechanism and the heat transfer member may be mounted on a common apparatus may comprise scraping means, for example one or more scrapers, arranged to scrape the inside of the vessel wall, or to scrape by-product, or other contents of the vessel, from the side wall. The scraping means may be mounted on the common rotatable support means for rotation with the heat transfer member.

The vessel wall may be cylindrical, for example having a central axis that is vertical. The vessel may further comprise heating means arranged to heat the by-product in the vessel.

The apparatus may further comprise a further processing vessel also comprising, a vessel wall; a lifting mechanism, which may be located in a central region of the vessel, arranged to lift the by-product within the vessel; a metal heat transfer member, which may extend around the lifting mechanism and may be in an outer region of the vessel, the heat transfer member optionally being rotatable around the outer region of the vessel to resist downward movement of the by-product in the outer region of the vessel and to dissipate heat through the by-product in the outer region of the vessel.

The vessel, or one of the vessels, may have an inlet connected to a supply of water. The vessel, or one of the vessels, may have an inlet connected to a supply of steam. The vessel, or one of the vessels, may have inlet means connected to a supply of acid and hydrogen peroxide.

The method may further comprise heating the mixture in the vessel.

The method may be carried out in a plant according to the invention.



The present invention may be carried out in an apparatus for processing organic waste the apparatus comprising a heating vessel having a product inlet arranged to receive the by-product, a water inlet arranged to receive water, and a steam inlet arranged to receive steam, and heating means arranged to heat the first heating vessel, a separator arranged to receive the contents of the first heating vessel and separate it into first and second components, and an acidification vessel having a product inlet arranged to receive the solid component, and acid inlet means arranged to receive acid. The acid inlet means may further be arranged to receive hydrogen peroxide.

Preferably the apparatus further comprises means to deliver a lime mix to the acidified by-product

Preferably the apparatus further comprises means to deliver magnesium salts to

the acidified by-product.

The apparatus may comprise a dryer operative to receive and dry the acidified by-product

A by-product sizing apparatus may be provided operative to re-size the byproduct into pieces below a predetermined size.

The sizing apparatus may comprise a crusher operative to crush the raw organic waste into pieces below a predetermined size.

The sizing apparatus may comprise a plurality of crushers arranged in sequence

and operative to sequentially crush the raw organic waste into smaller pieces.

Preferably the sizing apparatus is operative to output pieces of less than

A metal detector may be provided at the inlet operative to detect any metal content in the raw organic waste and to generate a corresponding signal that used to prevent raw organic waste entering the by-product sizing apparatus if metal content is detected. 5 Preferably the apparatus further comprises a pre-heating vessel operative

pre-heat the sized by-product and to deliver the sized by-product to the heating vessel.

The apparatus may further comprise a supply of at least one additive and means

to deliver the additive to the further heated by-product to alter the properties of the further heated by-product.

The apparatus may comprise a pellet press or granulator operative to receive dried by-product and to press or granulate the dried by-product into pellets granules.

The apparatus may further comprise a packaging unit or bagging machine operative to receive and package or bag the pellets or granules

The method may comprise a step of sizing the organic waste, for example by crushing, grinding or chopping solid waste to provide suitable sized pieces.

Suitable sized pieces may be less than 200mm, less than 150 mm, less than 100 mm, less than 50 mm or less than 20 mm sized pieces.

Water may be added to the organic waste before or after sizing the organic waste.

The method may comprise a heating step. The organic waste may be heated, for example to between 70 C and 200 C. The organic waste may be heated for between 10 and 60 minutes. The organic waste may be heated for 20 minutes at 135 C and 3 bar pressure to provide microbiological safety. The organic waste may be heated for a time and/or temperature as specified in legislation regarding disposal of organic waste or animal waste and/or microbiological safety. The organic waste may be heated by adding steam to the organic waste, for example steam may be added in a vessel as shown at [18] or [20] in Figure 1. The organic waste may be heated at higher than atmospheric pressure, for example in a pressurised vessel shown at [20] in Figure 1. The organic waste may be heated by heating the vessel in which the organic waste is held. The organic waste may be heated by an exothermic reaction when a hydrolysing composition is added to the organic waste, for example in a vessel as shown at

[22] in Figure 1 The method may comprise one or more, two or more, three or more, four or more or five or more heating steps. Each heating step may be performed in a different vessel for example a heating step may be performed in

each of vessels [18], [19] and [20] shown in Figure 1. The method may comprise two or more heating steps performed at different temperatures or pressures. Heating steps may be used to remove gelatine and fat; degrade the organic waste and/or provide a microbiologically safe product from organic waste.

The heating step may be an independent step in the method of processing organic waste performed at a different time from other steps. The heating step may be performed at the same time as another step in the method, for example the hydrolysis step. Combining the organic waste with a hydrolysing composition may cause an exothermic reaction that heats the organic waste.

During the heating step the organic waste may be heated using a combination of heating methods.

During the acid-peroxide/high temp/high pressure and/or digestion step a metal catalyst may be added. The metal catalyst may be any suitable metal or metal compound that is able to increase the rate of reaction with respect to breaking apart protein material such as the hydroxyproline in collagen into the constituent amino acid building blocks. The subsequently released nitrogen atoms from the peptide links are released in the form of ammonium ions. In particular these catalysts could be very useful in the dissolving of bone in acid/peroxide digestion step. The catalysts may include copper or copper compounds specifically copper II sulphate or copper II oxide, selenium or selenium compounds, titanium and/or titanium compounds but specifically titanium dioxide as either a powder or a nano-powder. Some of the catalysts, such as selenium and copper, would add benefit to the micronutrient composition of the final fertiliser product. The catalyst may be added in very small quantities, for example the mass of catalyst may be in the range of 0.1% 0.01% of the mass of undigested organic waste, for example bone paste.

The method may comprise a step of draining liquid from the organic waste.

The step of draining liquid from the organic waste may be performed after the heating step. Liquid may be removed from the mixture. The liquid may comprise fat and/or gelatine which may be separated and processed separately.

The method may comprise one or more hydrolysing steps where a hydrolysing composition is combined with the organic waste. The hydrolysing composition may be for example, a combination of an acid and hydrogen peroxide. These may be added to the organic waste together or sequentially. For example Sulphuric acid

H2SO4 50-100% (for example 95%) followed by or mixed with Hydrogen

Peroxide H2O2 5- 100% (for example 30% w/v) in a ratio of 1:1 to 1:10 sulphuric acid solution : hydrogen peroxide or for example, nitric acid 37% nitric acid (70% or 1.42 specific gravity) followed by or mixed with 7% hydrogen peroxide (35% w/w); a combination of a base and hydrogen peroxide. These may be added to the organic waste together or sequentially. For example ammonium hydroxide NH4 OH and Hydrogen Peroxide H2O2 30%; a base, for example Potassium hydroxide KOH or sodium hydroxide

NaOH; or an acid, for example Phosphoric acid H3PO4; or Hydrochloric acid HCL,

Nitric Acid HNO3, Sulphuric acid H2SO4.

The method may comprise one hydrolysis step, two hydrolysis steps or three hydrolysis steps. The hydrolysis steps may be performed sequentially. For example, the method may comprise an acid hydrolysis step and a base hydrolysis step performed sequentially. The method may comprise an acid or a base hydrolysis step and a hydrolysis step using acid or base and hydrogen peroxide.

The hydrolysis step may be an exothermic reaction and may cause the organic waste mixture to heat up. Heat may be removed from the organic waste mixture during this step to keep the mixture within suitable temperature limits.

If the hydrolysis step is exothermic it may contribute heat to the heating step.

The one or more hydrolysis steps may be performed in a vessel, for example as shown at [22] in Figure 1.

After the one or more hydrolysis steps the hydrolysed organic waste may provide an acidified hydrolysed organic waste.

The method may comprise a step of draining liquid from the organic waste.

The step of draining liquid from the organic waste may be performed after the hydrolysing step. Liquid may be removed from the mixture. The liquid may comprise fat and/or gelatine and/or other components which may be separated from each other and processed separately.

The method may comprise an acidification step where the hydrolysed organic waste is combined with additional acid to bring the pH of the organic waste to an acid pH below pH 7. An acid, for example phosphoric acid, nitric acid, or sulphuric acid may be added if the organic waste mixture is not acidic after the hydrolysis step or if the organic waste mixture is not acidic enough after the hydrolysis step. The hydrolysed organic waste may be acidified to a pH between pH -2 and pH 7, between pH 1 and pH 7, between pH 2 and pH 6, between pH 3 and pH 6 or between pH 4 and pH 5. The amount of acid or base added may be enough to make the organic waste mixture slightly acid, for example pH 6.5 to pH 4.

The one or more acidification or neutralisation steps may be performed in a vessel, for example as shown at [22] in Figure 1.

The method may further comprise a step of adding nutrients or minerals, for example magnesium salts to the organic waste mixture. This is advantageous because additional nutrients that are required in a fertiliser may be added to the processed organic waste so that it can be made into a fertiliser with a suitable balance of nutrients. Nutrients or minerals may be selected in order to complement the known components of the organic waste to provide nutrients or minerals that are known to be lacking in the organic waste. Nutrients or minerals may be selected to address a deficiency in the area where the processed organic waste is going to be used as a fertiliser.

The method may further comprise a step of adding ammonia and/or other basic or carbonate materials to increase the nitrogen or other minerals in the organic waste so that the final fertiliser has higher nitrogen or other minerals. For example ammonia and/or other minerals may be added to the organic waste at any point in the processing in order to increase the nitrogen content, the phosphate content or the potassium content of the final fertiliser product.

Ammonia and/or other minerals may be added to the organic waste at any point in the processing in order to produce a final fertiliser product with a particular balance of nitrogen phosphate and potassium (NPK content) that is suitable for a specific soil type and/or crop type.

The method may comprise the step of drying the neutralised organic waste mixture. This may be done by a combination of heating and the addition of a dehydrating granulating mixture followed by further heating in the granule drying stage of the process. The acidified organic waste mixture may be dried until it comprises 10-12% moisture.

The method may comprise a step of setting the organic waste mixture by combining acidic, hydrolysed organic waste with a setting mixture or liming 5 mixture. This may be any compound or composition that can form a lime mortar.

The method may comprise a re-sizing step to re-size the organic waste mixture into pieces below a predetermined size. The re-sizing step may, for example, comprise crushing, chopping or grinding the product. The product may be re-sized to pieces of less than 40mm.

The method may further comprise adding at least one additive to alter the properties of the organic waste mixture. For example, an unpalatable ingredient may be added to deter animals from eating the fertiliser.

The method may comprise the step of pelleting or granulating the organic waste mixture after the setting step.

Granular fertilisers may be treated with hydrophobic anti-caking agents such

Kaolin by externally coating of the granules, or incorporated as part of the granulation mix; and or paraffin based additives which may be sprayed onto hot granules.

The method may further comprise a packaging step where the pellets or granules are packaged.

Organic waste may comprise, for example, waste from abattoirs, waste from meat processing or packaging, fallen stock, animal carcasses, food waste, animal excrement, for example, cow slurry, pig or cow dung, poultry litter, animal bedding, waste from the food industry, abattoir blood waste. Organic waste may be animal waste, vegetable waste or mixed animal and vegetable waste.

The method of processing organic waste may be a method of making fertiliser.

The hydrolysing composition may be chosen in order to provide suitable ingredients to the fertiliser product, for example, use of nitric acid in the hydrolysing composition may provide additional nitrogen in the fertiliser product. Use of phosphoric acid in the hydrolysing composition may provide additional phosphorus in the fertiliser product. Use of sulphuric acid in the hydrolysing composition may provide additional sulphur in the fertiliser product.

The hydrolysing composition, neutralising composition and additives may be chosen to provide a suitable level of potassium, phosphorous and nitrogen in the final fertiliser when they are combined with the organic waste.

The hydrolysing composition may also be selected because of its suitability for hydrolysing the particular components of the organic waste that it is combined

with. For example a hydrolysing composition comprising a base is particularly suitable for hydrolysing organic waste comprising cellulose. A hydrolysing composition comprising acid may be particularly suitable for hydrolysing organic waste comprising bone.

Ammonium hydroxide NH_4OH and Hydrogen Peroxide H_2O_2 30% may be particularly suitable for hydrolysing poultry litter, pig manure, cow manure, horse manure and/or contents of intestinal tract.

The method of the present invention may comprise a hydrolysis step using potassium hydroxide KOH and then a neutralising step using nitric acid to bring the pH close to neutral but keep the mixture slightly alkaline or slightly acidic as required for granulation into granules. The mixture may be set using lime, plaster of paris $\text{CaSO}_4 \cdot 1/2 \text{H}_2\text{O}$, Calcium carbonate CaCO_3 , Calcium oxide CaO or Calcium hydroxide Ca(OH)_2 .

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one hydrolysis step.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step and least one hydrolysis step.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step, at least one hydrolysis step and at least one neutralising step.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step, at least one hydrolysis step, at least one neutralising step and at least one pelleting step.

The method of the present invention may be a method of processing organic waste to provide a slow-release fertiliser comprising at least one hydrolysis step to provide hydrolysed organic waste and one setting step to provide a slow release fertiliser.

The at least one hydrolysis step may comprise an acid hydrolysis step to provide an acidified hydrolysed organic waste

The at least one hydrolysis step may comprise an acid hydrolysis step and a base hydrolysis step to produce an acidified hydrolysed organic waste.

The at least one hydrolysis step may comprise an acid hydrolysis step using any acid, for example sulphuric acid, nitric acid and/or phosphoric acid to hydrolyse organic waste and produce an acidified hydrolysed organic waste. The at least one hydrolysis step may comprise an acid hydrolysis step using sulphuric acid,

to hydrolyse organic waste and produce an acidified hydrolysed organic waste.

The method may comprise an optional acidification step to make the pH of the acidified organic waste below pH 7.

The acidified hydrolysed organic waste may comprise sulphuric acid, phosphoric acid and nitric acid.

The method may comprise a step of setting the acidified hydrolysed organic waste mixture by combining acidic, hydrolysed organic waste with a setting mixture or a liming mixture. This may be any compound or composition that can form a lime mortar. The organic waste may be bonded to the lime mortar or embedded in it. The liming mixture may comprise, for example lime, calcium sulphate CaSO_4 , Calcium carbonate CaCO_3 , Calcium oxide CaO and/or Calcium hydroxide Ca(OH)_2 .

The liming mixture may further comprise clay bases materials such as kaolin.



The liming mixture may set to provide a lime mortar. During the setting step the organic waste and any additives that have been added to it may become bonded to the lime mortar or set within the lime mortar to provide a slow release fertiliser.

The lime mortar may degrade when left exposed to the elements or in the soil over a period of 1 to 6 months, or over a period of more than 9 months, or over a period of 3 to 9 months, or over a period of 3 to 6 months, or over a period

3 to 5 months or over a period of 4 to 5 months. The lime mortar may degrade more quickly in soil that is more acidic.

The composition of the setting mixture or liming mixture may be adjusted to provide a slow release fertiliser that degrades faster or slower depending on what is required. In order to adjust the time that the fertiliser takes to degrade different proportions of each of the liming mixture ingredients may be added the acidified organic waste. For example, the liming mixture can be adjusted

that more calcium sulphate is produced in the setting step to provide a fertiliser that breaks down more slowly. More calcium sulphate may be produced by reducing the amount of lime in the liming mixture so that the excess sulphuric

acid reacts with the lime to make more calcium sulphate. The liming mixture can be adjusted so that it degrades faster by including more lime in the liming mixture.

The specific conditions where the slow release fertiliser will be used may be considered when selecting the composition of the organic waste and the composition of the liming mixture. The specific conditions may be the pH of the soil, the expected rain fall, the type of crop, the nutrient requirements of the crop, the nutrient profile of the soil, the ambient temperature that the crop will grow at, the length of growing time of the crop and/or the rate at which the crop requires nutrients during particular phases of its growth.

The percentage of liming mixture compared to acidified organic waste may be adjusted to provide a fertiliser that degrades faster or slower. The percentage of liming mixture used to set the acidified organic waste may be 10% to 50%, may be 20% to 50%, or may be 30% to 50%, or may be 40% to 50% of the mass of the organic waste.

The percentage of liming mixture added compared to the amount of acidified organic waste may be adjusted to provide a fertiliser with a higher or lower neutralising value. Neutralising value may be the amount of base compounds in the fertiliser. For example, a higher neutralising value may be required to partially neutralise acid soil or to provide lime to crops that prefer alkaline conditions. The percentage of liming mixture used to set the acidified organic waste may be 10% to 50% of the mass of the organic waste.

Lime mortar may comprise lime, calcium sulphate CaSO_4 , Calcium carbonate

CaCO_3 , Calcium oxide CaO and/or Calcium hydroxide Ca(OH)_2 , which react with water and/or acid and set to form solid lime mortar. The liming mixture may further comprise clay bases materials such as kaolin to change the properties of the lime mortar. Other components of the hydrolysed organic waste may become part of the lime mortar by chemically bonding to components of the lime mortar or by being immobilised inside the lime mortar but not chemically bonded to it. As the lime mortar degrades in soil it may release lime into the soil and may also release components of the organic waste into the soil which provide nutrients and act as a fertiliser.

The setting step may comprise admixing the acidic mixture with lime and/or one or more calcium compounds wherein the calcium compounds are able to react with water and/or acid to form lime mortar which sets the mixture to form a slow release fertiliser. One or more calcium compounds within the liming mixture may react with sulphuric acid to form calcium sulphate which sets the mixture to form part of the lime mortar of the slow release fertiliser.

The lime mortar may set the mixture to form a solid slow release fertiliser with the components of the acidic mixture bonded inside the lime mortar matrix.

The setting step may comprise admixing the acidic hydrolysed organic waste mixture with one or more selected from calcium oxide, calcium carbonate and calcium hydroxide to form a slow release fertiliser.

The setting step may comprise admixing the acidic mixture with calcium oxide and calcium carbonate to form a slow release fertiliser.

The setting step may comprise admixing the acidic mixture with calcium oxide, and calcium hydroxide to form a slow release fertiliser.

The setting step may comprise admixing the acidic mixture with calcium carbonate and calcium hydroxide to form a slow release fertiliser.

The proportions of acidic mixture, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product is slightly alkali.

The proportions of acidic mixture, lime, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product takes more than a month, more than two months, more than 3 months, more than 4 months, more than 5 months or more than 6 months to degrade in soil. The soil may be between pH 5 and pH 7. The soil may be between pH 5 and pH 9.

The proportions of acidic mixture, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product takes between one and 6 months, between 1 and 4 months or between 2 and 3 months to degrade in slightly acidic soil. For example the soil may be between pH 5 and pH 7.

The method may comprise the step of adding one or more minerals to provide suitable nutrients in the slow release fertiliser.

The method may comprise the step of adding one or more fungi sides to provide antifungal properties to the slow release fertiliser.

The method may comprise the step of adding one or more pesticides to provide pesticide properties to the slow release fertiliser.

The method may comprise the step of adding one or more herbicides to provide herbicide properties to the slow release fertiliser.

The method may comprise the step of adding one or more unpalatable compounds deter animals from eating the slow release fertiliser.

The method may comprise the step of processing the slow release fertiliser, for example by pelleting, granulating, press-forming or powderising the slow release fertiliser.

The method may comprise the step of coating the fertiliser, for example coating pellets or granules of the fertiliser, with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds.

In a second aspect the present invention provides a slow release fertiliser provided by the method of the present invention.

In a further aspect the present invention provides a slow release fertiliser comprising a proportion of lime mortar to set the fertiliser.

The slow release fertiliser may comprise more than 10%, more than 20%, more than 30%, more than 40%, more than 50%, more than 60% or more than 70% lime mortar.

The slow release fertiliser may comprise between 10% and 70%, between 20% and 50% or between 30% and 40% lime mortar.

The slow release fertiliser may comprise lime mortar that takes between one and 6 months, between 1 and 4 months or between 2 and 3 months to degrade in slightly acidic soil. For example the soil may be between pH 5 and pH 7.

The slow release fertiliser may further comprise one or more fungicides, one

more pesticides, one or more herbicides, one or more pest control agents and/or one or more unpalatable compounds. Pellets of the fertiliser may be coated with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds

In a further aspect the present invention provides a composition obtained or obtainable by the method of the invention.

In a further aspect the present invention provides a fertiliser composition obtained or obtainable by the method of the present invention

There now follows by way of example only a detailed description of the present invention with reference to the accompanying drawings, in which,

Figure 1 is a schematic view of an apparatus for processing organic waste;

Figure 2 is a cut-away side view of a heating vessel forming part of the plant of Figure 1;



Figure 3 is a plan view of the vessel of Figure 2;

Figure 4 is a cut-away side view of a heating vessel of Figure 2 with further components shown; and

Figure 5 is a plan view of the vessel of Figure 4;

Figure 6 shows a beaker containing 95g of congealed frozen blood and g of water; 25 Figure 7 shows a view of the hydrolysis and heating stage being performed on 95 g of blood with 30 g of water, 10g of potassium

Hydroxide pellets have been added and the mixture was stirred for two minutes to form a deep red solution. This reaction is very exothermic and so hydrolysis and heating happen at the same time;

Figure 8 shows the mixture of figure 7 with 8g of 70% (1.42 specific gravity) nitric acid added to neutralise the potassium hydroxide this step is also very exothermic and is the neutralisation step and a heating step at the same time. The resulting mixture is an alkaline solution of potassium nitrate containing dissolved hydrolysed blood;

Figure 9 shows a view of the solid set product that results when the mixture shown in figure 8 is combined with CaSO_4 , $\frac{1}{2} \text{H}_2\text{O}$ (left hand side) or with CaO , CaCO_3 and Ca(OH)_2 ;

Figure 10 shows a view of 63 g of chicken litter and bedding;

Figure 11 shows a view of the 63 g of chicken litter and bedding mixed with 150 ml and blended to form a workable paste;

Figure 12 shows a view of the blended mixture shown in Figure 11 with 3 g (0.0306 mole) of H_3PO_4 (orthophosphoric acid (1.75 specific gravity) added and the mixture brought to the boil. This is the heating stage and a first hydrolysis stage (acid hydrolysis) done at the same time;

Figure 13 shows a view of the mixture in Figure 12 with 0.0918 mol of potassium hydroxide (5.14 g) added to neutralise the acid, then 5 g more potassium hydroxide added to increase the pH to 14 and hydrolyse fat, proteins, lipids and nucleic acids, this stage is highly exothermic and is therefore a second hydrolysis step (base hydrolysis) performed at the same time as a heating step;

Figure 14 shows a view of the resulting mixture after pH of the mixture shown in Figure 13 is neutralised using using 8 g of 70% (1.42 specific gravity) nitric acid; 5 Figure 15 shows a view of the mixture shown in Figure 14 after addition of calcium carbonate, calcium hydroxide and calcium oxide to produce a solid pellet; and

Figure 16 shows a pellet of fertiliser made from chicken litter using a 10 process of the present invention.

The method of the present invention may be carried out in an apparatus for processing organic waste, an example of which is set out below and in Figures

Referring to Figure 1, an apparatus for processing organic waste comprises an inlet 10 which receives raw organic waste. A metal detector 12 is provided at the inlet 10 to detect any metal in the raw organic waste and to generate a control signal indicative of any metal content detected. The inlet 10 feeds into a particle sizing module, which in this case is a crushing module 14, arranged

reduce the raw product to pieces of a suitable size. In this embodiment the crushing module 14 comprises a first crusher arranged to crush the raw organic waste into pieces of less than a predetermined size, said size being 150mm in this example, and a second crusher arranged to crush the raw organic waste into smaller pieces, of less than 40mm in this example. A conveyor 16, which may be a screw conveyor, is arranged to transfer the crushed product from the outlet of the crushing module 14 to the first 18 of three heating vessels 18, 20, 22.

The first heating vessel 18 comprises an enclosed steel container having an inlet 24 at the top through which the crushed product can be introduced, a further inlet 26, also at the top, which is connected to a water supply via a pump 28 so that water can be added to the product in the vessel 18, a further inlet 29 connected to a supply of steam, and an outlet 31 at the bottom. Flow control valves are provided in the water and steam inlets 26, 29, and the outlet 31. A heater 30 is provided around the wall of the vessel, and a temperature gauge 32 is arranged to measure the temperature in the vessel 18. A controller 34 is arranged to control the operation of the whole plant, including all the conveyors, pumps and heaters, and the inlet and outlet valves, as will be described in more detail below.

A first separation vessel 36 is arranged to receive the product from the outlet 31 from the first heating vessel, and is arranged to hold the solid product and allow liquid to drain into a collection tank 38 situated below it. A second conveyor

is arranged to transport the solid product from the separation vessel 36 to the inlet 42 of the second heating vessel 20.

The second heating vessel 20 is pressurized, and has a pressure vent controlled by a vent valve 43 so that the pressure in the vessel 20 can be controlled. It also has a further inlet 44 for water, a further inlet 45 for steam, and an outlet

its bottom end, each with a flow control valve controlled by the controller

also has a heater 48 around its side walls, and a temperature gauge 50 and pressure gauge 52 arranged to measure the temperature and pressure of the contents of the second vessel 20. These are connected to the controller 34 which is arranged to control the temperature and pressure, as well as the quantities of water, steam, and product, in the second vessel 20 as required.

A second separation vessel 56 is arranged to receive the product from the outlet 46 from the second heating vessel 20, and is arranged to hold the solid product and allow liquid to drain into a collection tank 58 situated below it.

third conveyor 60 is arranged to transport the solid product from the separation vessel 56 to a second crusher unit 62. The second crusher unit 62 is arranged break down the solid product into pieces no larger than 2mm.

The outlet from the crusher unit 62 is arranged to output the product into a pan 64, which may be arranged to pre-heat the product prior to it being received in the third heating vessel 22. A further conveyor 66 is arranged to transport the product from the pan 64 to the inlet 68 of the third heating vessel 22.

The third vessel 68 also has a further inlet 70 at its top end connected to a source 72 of acid via a pump 74, and a further inlet 76 connected to a source of hydrogen peroxide, via pump 80. These are controlled by the controller 34

that acid and hydrogen peroxide can be added into the vessel 22 in the required amounts and rates and at the required time. The third vessel also has a heater

and a temperature gauge 84 connected to the controller 34. An outlet 84 at the

bottom end of the vessel 22 is connected to a neutralization vessel 86 which

arranged to receive the product from the third heating vessel 22, and also arranged to receive an alkaline additive, in this embodiment a lime mix, via a conveyor 88, from an alkaline additive source 90, under the control of the controller 34.

The final neutralized product is in the form of a paste and an extruder 92 is arranged to receive this product from the neutralization vessel 86 and extrude into pellets, and a packaging module 94 is arranged to receive the extruded pellets and package them for transport away from the plant.

Each of the liquid collection vessels 38, 58 has an outlet that is connected

separator in the form of a centrifuge 96 which is arranged to separate out the lighter and heavier components of the liquid. The lighter components will generally comprise fats, and the heavier components generally comprise gelatine, and these components are collected in separate collection vessels

Operation of the plant will now be described with particular reference to the treatment of bone, although it will be appreciated that it can be used for a wide variety of other waste products.

Raw bone is introduced into the crusher module 14 where it is crushed into pieces no larger than 40mm. From there it is transferred into the first vessel

The controller 34 is arranged to monitor the amount of product entering the vessel 18, for example using a load cell to measure its weight, and to stop the transfer of product when a specified amount has entered the vessel. The controller 34 is also arranged to control the introduction of water and steam into the vessel 18, and to heat the vessel 18 to the required temperature. The bone and water mixture is heated to approximately 95 C to form an organic waste slurry which is then transferred to the separation vessel 36.

In the separation vessel 36, liquid drains from the slurry into the collection

tank 38 while the slurry is transferred along the vessel. The drier slurry is then transferred via the conveyor 40 to the second heating vessel 20. The amount of slurry introduced into the second vessel 20 is controlled by the controller, for example using weighing cells to weigh the amount added, and then controlled amounts of water and steam are added by controlled operation of the inlet valves. The second vessel 20 is then heated up to about 135 C, and the vent valve 43 closed so that the pressure increases to about 5bar. This temperature and pressure is maintained for about twenty minutes.

At the end of the treatment in the second vessel 20, the vent valve 43 is opened to reduce the pressure in the vessel, and the outlet 46 is opened to allow the product, which is still in the form of a slurry, to empty into the second separation vessel 56. From that vessel 56, liquid drains into the second collection vessel 58, and the solid component of the slurry is transferred via

conveyor 60 into the second crusher unit 62, where it is crushed to a particle

size of about 2mm and transferred to the pre-heating vessel 64, where it is preheated.

From the pre-heating vessel 64 the solid product is transferred via conveyor

into the third heating vessel 22 via the conveyor 66 and the inlet 68. Again the amount of product introduced is controlled, and also sulphuric acid and hydrogen peroxide are added in controlled amounts via the inlets 70, 76. The reaction in the vessel 22 is exothermic and the temperature rises to about 95 and then falls off when the reaction is complete. This heat and the acid and hydrogen peroxide kills any disease or bacteria in the product, and also oxidises any carbon in the product resulting in an inorganic product which is suitable for use as a fertiliser.

When the acidification step is complete, the outlet 22 from the third vessel

opened and the product transferred to the neutralization vessel 86, to which lime mix is added in controlled quantities in order to increase the pH of the mixture to neutralize the product. The lime mix may comprise a mixture of chalk and lime, or quick lime for example. The lime mix may also comprise additives to alter the properties of the mix, such additives for example comprising selected macro/micro nutrients. The lime mix, and/or the additives, can be varied as required to deliver the required fertilizer properties. The lime mix causes an exothermic reaction. This serves to at least partially dry the acidified organic waste.

The increased pH mixture, which is typically in the form of a paste, is then dried if necessary, and passed into the extruder 92, from which it is extruded pellets or granules, and packaged at packaging module 94 for transport.

The liquid from the collection vessels 38, 58 is separated in the centrifuge

with fat being transferred to fat storage tank 98 and gelatine to the gelatine storage tank 100.

The third heating vessel 22 may also comprise additive inlets arranged to deliver further additives to the slurry. These additives can be of any suitable ingredient to adjust the mineral or nutritional content of the solid slurry particles to suit a particular use or location of use. Thus the additives can controlled to alter the macro/micro nutrient bases of the solid slurry particles to suit different soil or ground conditions for example, or to provide optimum nutrition for a particular type of crop.

Referring to Figures 2 and 3, the first heating vessel 18 comprises a cylindrical steel body 200, about 1.25m high and 750mm in diameter, arranged so that its central axis 202 and curved side walls 204 extend vertically. A pair of parhelical heat transfer blades 206 are mounted by means of cross members 208 on 5 a central rotatable shaft 210 which is located on the central axis 202.

Each of the heat transfer blades 206 is formed of a strip of sheet metal 209 about 15

wide, which is formed into a part-helix shape extending around the outer part

the vessel close to the side wall 204 but spaced slightly from it. The blades each extend from a point about two thirds of the way up the side of the vessel, down 10 to close to the bottom of the vessel, and through half a turn around the central axis. The two blades 206 are diametrically opposite each other, so that they extend around opposite sides of the vessel.

A pair of scrapers 212 are also supported on the cross members 208. Each scraper 212 comprises a flat scraper blade 214 extending vertically down the 15 side wall 204 of the vessel with an outer scraping edge 216 just clear of the side wall 204. Each of the scraper blades 214 is supported on a set of support brackets 218, which in turn are supported on a vertical support pole 220 which extends vertically between the cross members 208 near their outer ends.

Referring to Figures 4 and 5, an inner helical lifting flight 222 is also mounted 20 on the shaft 210. The lifting flight 222 is formed of a flat strip of metal 224 about 100mm wide, formed into a helix centred on the shaft 210, and turning through about seven turns. The top of the lifting flight 222 is about level with the top of the heat transfer blades 206 and the bottom of the lifting flight

slightly above the bottom of the heat transfer blades 206. The turning sense

25 the lifting flight 222 is the same as both of the heat transfer blades

motor 224 is arranged to rotate the shaft 210, so that the lower end of the lifting flight forms a leading end and the upper end of the lifting flight forms a trailing end, and the lifting flight provides lift in the central part of the vessel 18 close to the central axis 202. At the same time, the lower ends of the heat transfer blades 206 form leading ends and the upper ends of the heat transfer blades form trailing ends, and the heat transfer blades also provide a small amount

lift, or in fact resistance to downward flow, in the outer part of the vessel closer to the outer wall 204. As the shaft rotates, the scrapers 212 move around the wall 204 of the vessel, scraping from the wall any of the by-product or other contents of the vessel that may have accumulated on the wall.

The central shaft 210 is rotated as the product is introduced into the vessel

through the inlet 24, the water is introduced through the inlet 26, this forms

slurry, and the steam is introduced through the inlet 29 which helps to heat the slurry. As the heater 30 heats up the contents of the vessel 18 and the steam injected into it, the central lifting flight 222 causes a general upward flow

the mixture in the central area of the vessel. The mixture then moves outwards at the top of the vessel 18 and cascades downwards in the outer region of the vessel 18 close to the wall 204. The rotating heat exchange blades 206 provide

some resistance to the downward flow, and therefore mix the mixture, but also help to conduct heat through the mixture quickly allowing it to be heated quickly to the required temperature. This helps to reduce the time required for the heating step carried out in the first heating vessel 18.

The second vessel 20 has the same mixing and heat exchange mechanism as the first vessel 18 as shown in Figures 2 to 5. In this vessel the heating process similar to the first vessel, though at increased pressure, and the heat exchange coils and the lifting flight serve to distribute heat from the heater and steam throughout the vessel in a similar way. Similarly the third vessel 22 also has the same mixing and heat exchange mechanisms as shown in Figures 2 to 5. In this case the heat is generated by the exothermic reaction in the vessel. However, rapid heat dissipation is still helpful to ensure that the temperature remains approximately equal throughout the vessel, and that the reaction proceeds at about the same rate throughout the vessel.

It will be appreciated that various modifications can be made to the embodiment described above, and that the design of the heating vessel or vessels can be varied as appropriate for the process and products to be treated.

For example, the two heat transfer blades could be replaced by one continuous helical coil, or more than two blades. Also the blades may not be strictly parhelical. For example the angle of the blade or blades to the vertical may vary

along its length. Similarly the lifting flight may be of a different shape, or may be rotated independently of the heat transfer coils. Indeed a different type lifting mechanism may be provided which is not a helical flight.



A number of pilot studies, described in the examples below, were performed to assess the effectiveness of different compositions for use in the hydrolysis step of the present method. The hydrolysis step may take place in one of the reaction vessels, for example vessel [22] in Figure 1. The pilot studies were mostly done on a small scale but may be scaled up and the method may be performed on large quantities of organic waste, for example in an apparatus as shown in Figure 1. The inlets for acid and hydrogen peroxide described in relation to Figure 1 may alternatively be used to add other ingredients for the hydrolysis and neutralisation steps.

Example 1 Acid and Hydrogen Peroxide - Bone Paste

The preferred ranges of the temperature and reaction times.

Operating temperature range: Start temp of bone paste 10 C to 100 C giving an end temp after the hydrolysis step of 100 C to 200 C.

Preferred temp for process: Start temperature 50-90 C " end temp 130-170 C.

Ratio of wet bone paste to acid to hydrogen peroxide by mass 5-20:5:1 (wet bone:acid:peroxide), for example 13.5:5:1 (wet bone : acid : peroxide).

For example the ratio may be: 1000kg bone paste (wet) to 370kg of 95% sulphuric acid to 74kg of 30% hydrogen peroxide.

Treatment time Broad range " 1min to 30mins, for example, the treatment time may be 5-10 mins.

The amounts of acid and hydrogen peroxide - Broad Range " 1:1 to 1:10

Concentrated sulphuric acid to hydrogen peroxide solution.

Preferred ratio is 1:5 concentrated sulphuric acid to hydrogen peroxide for bone digestion but will vary for other waste streams.

Concentrated sulphuric acid used is 95% (1.83g/ml) but this could work over a range of 50" 100%.

Concentrated hydrogen peroxide used is 30%w/v (weight to volume) but concentrations between 5-100% w/v although greater than 30% can carries an explosion risk.

Example 2 Base and Hydrogen Peroxide

Base peroxide

Broad range of concentrated ammonium hydroxide, and/or potassium hydroxide and/or sodium with hydrogen peroxide from 1:1 to 1:10 using 0.880sg (specific gravity) ammonium hydroxide (ammonia solution) and 30% hydrogen peroxide solution.

Preferred mixture 3:1 of 0.880sg ammonia solution to 30% hydrogen peroxide solution for the purpose of dissolving for example poultry litter, pig manure, cow slurry, cow manure, horse manure, contents of intestinal tract material from abattoir waste etc.

Example 3 Hydrolysis of Blood

Overview " blood contents are hydrolysed using sodium hydroxide or potassium hydroxide. The resulting solution is neutralised using nitric acid sulphuric acid forming a potassium nitrate, sodium nitrate, sodium sulphate or potassium sulphate solution depending on the acid and base used solution containing the dissolved, heat treated, hydrolysed blood components.

Method:

To 100g of congealed blood add 30g of water and bring to the boil (water not required if blood is fresh and uncongealed). NB " frozen blood used can be seen melting in Figure 1.

Hydrolysis: Add 10g of Potassium Hydroxide pellets KOH (solid, form preferred) or 7g of sodium hydroxide (solid form preferred) and stir for two minutes until deep red solution forms " very exothermic reaction. The product of this reaction is shown in Figure 2.

Neutralisation: Add 16g of 70% 1.42 specific gravity nitric acid to neutralise the potassium hydroxide (concentrated) although a dilute nitric acid solution would suffice " very exothermic. The resulting mixture is shown in Figure 3.

The resultant mixture is then treated with further nitric acid to achieve the desired pH for granulating " this could be acidic or alkaline.

Solution of potassium nitrate containing dissolved, hydrolysed blood.

If a small amount of solid precipitates from the blood it may be separated and

hydrolysed/oxidised by a small, additional dose of nitric(or sulphuric) acid and hydrogen peroxide in a ratio of 5:1 before recombining with original solution for pasting or setting.

The neutralised hydrolysed blood may be blended with other soluble ingredients

and the resultant solution sold as a liquid fertiliser or the neutralised hydrolysed blood may be granulated and sold as a solid granular form after setting using plaster of paris (CaSO₄.1/2H₂O) Figure 4 left or Calcium carbonate/oxide/hydroxide CaCO₃/CaO/Ca(OH)₂ Figure 4 right.

Example 4 Acid hydrolysis using phosphoric acid (other acids could be used - nitric acid or sulphuric acid) followed by alkaline hydrolysis using potassium hydroxide or sodium hydroxide " resultant mixture neutralised using nitric acid or sulphuric acid.

Phosphoric acid increases the phosphorus content (P) and hydrolyses cellulose content. Nitric acid increases the N content.

Potassium hydroxide increases the K content and hydrolyses bulk content. 10 Nitric acid neutralises the potassium hydroxide or sodium hydroxide forming potassium nitrate or sodium nitrate and increases the N content.

Approximate ratios by mass " 42(chicken litter) : 100(water) : 2(phosphoric acid) : 6.6(potassium hydroxide) or 4.7 sodium hydroxide: 8.0 (nitric acid).

Overview 63g of chicken litter and bedding is shown in Figure 5.

Addition of 150ml of water followed by binding' to form a suitable, workable paste shown in Figure 6.

Acid hydrolysis: - high-heat stage of process, 3g (0.0306 mole) of H₃PO₄ (Orthophosphoric acid 1.75 specific gravity) or 2.75g 70% nitric acid is added drop pH to pH 3-4 and mixture brought to the boil (high heat stage of process).

Alkaline hydrolysis " in heliotherm stage. First the phosphoric acid is neutralised by potassium hydroxide:

H₃PO₄ + 3KOH " Na₃PO₄ + 3H₂O 3.00g of phosphoric acid (0.0306 mol) requires 0.0918 mol of potassium hydroxide = 5.14g then an addition 5g of KOH is required to facilitate the increase in pH to 14 and hydrolyse fat/protein/lipids/ nucleic acids etc (total 10g of KOH) " lots of heat evolved NaOH could also be used (less required ")

Neutralisation " still in heliotherm stage.

Neutralisation using 17g of 70% (1.42 specific gravity) nitric acid.

Resultant mixture is an alkaline, sweet (molasses type smell) brown paste containing potassium nitrate, potassium phosphate and all the components of chicken litter suitable for plant uptake.

Addition of calcium carbonate/hydroxide/oxide/calcium sulphate heptahydrate mix to produce the solid pellet form is shown in Figure 10.

Sodium hydroxide is useful for the hydrolysis of blood or chicken litter followed by neutralisation using sulphuric acid. For example 100g of blood requires 5g of sodium hydroxide at around 90 C to hydrolyse, followed by neutralisation using concentrated sulphuric acid 3.6ml (6.63g) and then adjustment of pH to desired level.

The present invention may be further described in the following numbered paragraphs: 1.A method of processing organic waste comprising one or more step of combining the organic waste with a hydrolysing composition. 2.The method according to paragraph 1 further comprising a heating step. 3.The method

according to paragraph 2 wherein the organic waste is heated to between 70 C and 200 C. 4.The method according to paragraph 2 or paragraph 3 wherein the organic waste is heated for between 10 and 60 minutes.

5.The method according to any one of the preceding paragraphs further comprising the step of removing liquid comprising fat and/or gelatine from the organic waste. 6.The method according to any one of the preceding paragraphs wherein the hydrolysing composition comprises: an acid and hydrogen peroxide; a strong base and hydrogen peroxide; a base; an acid; or hydrogen peroxide. 7.The method according to any one of the preceding paragraphs further comprising a sizing step wherein the organic waste is sized to pieces less than 50mm. 8.The method according to any one of the preceding paragraphs further comprising a neutralisation step, wherein acid or base is added to the mixture to bring the pH near to neutral. 9.The method according to any one of the preceding paragraphs further comprising a step of adding minerals or nutrients to the mixture. 10. The method according to any one of the preceding paragraphs further comprising drying the organic waste mixture. 11 The method according to any one of the preceding paragraphs further comprising adding a setting agent such as calcium carbonate, calcium hydroxide, calcium sulphate hemihydrate and/or calcium oxide to the organic waste in a quantity suitable to set the mixture. 12. The method according to any one of the preceding paragraphs further comprising the step of re-sizing the set organic waste into pieces of less than 40 mm.

13. The method according to any one of the preceding paragraphs further comprising adding one or more unpalatable ingredients to the organic waste mixture. 14 The method according to any one of the preceding paragraphs further comprising a step of pelleting or granulating the organic waste mixture. 15. The method according to any one of the preceding paragraphs wherein the method of processing organic waste is a method of producing fertiliser. 16. An organic waste composition obtained by the method according to any one of the preceding paragraphs. 17. A fertiliser composition obtained by the method of any one of the preceding paragraphs.

6. CA2915964 - METHOD OF PROCESSING ORGANIC WASTE TO PROVIDE A SLOW-RELEASE FERTILISER

National Biblio. Data

Description

Claims

Drawings

Patent Family

Documents

PermaLink

Machine translation

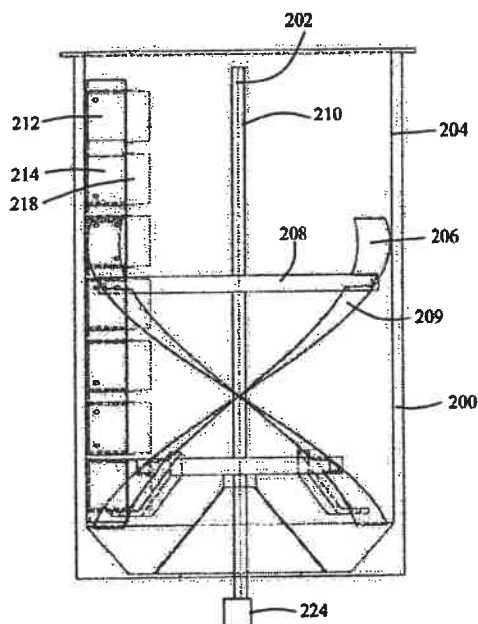
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[EN]
CLAIMS

1. A method of processing organic waste to provide a slow-release fertiliser comprising at least one digesting step to provide digested organic waste and a setting step to provide the slow-release fertiliser, wherein said digesting step comprises combining the organic waste with a metal catalyst, 50-100% concentrate of sulphuric acid and 5-100% concentrate of hydrogen peroxide, wherein the ratio of sulphuric acid to hydrogen peroxide is 1:1 to 1:10, and wherein said metal catalyst is titanium, copper, or compounds thereof; and wherein the setting step comprises combining the digested organic waste into a setting mixture comprising or consisting of one or more of lime, calcium sulphate, calcium carbonate, calcium oxide and calcium hydroxide.
2. The method according to claim 1 wherein the setting mixture further comprises .. clay base materials.
3. The method according to claim 1 or 2 wherein the lime in the setting mixture sets to provide a lime mortar with the digested organic waste bonded to the lime mortar or set within the lime mortar to provide the slow release fertiliser.
4. The method according to any one of claims 1 to 3 further comprising the step of adding one or more minerals at any stage in the method to provide suitable nutrients in the slow release fertiliser.
5. The method according to any one of claims 1 to 4 further comprising the step of adding one or more fungicides, pesticides, and/or herbicides to provide desirable properties to the slow release fertiliser.
6. The method according to any one of claims 1 to 5 further comprising the step of processing the slow release fertiliser by pelleting, granulating, pressforming or powderising the slow release fertiliser.
- Date Recue/Date Received 2021-03-16
7. The method according to any one of claims 1 to 6 further comprising the step of coating the fertiliser with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, and/or pest control agents.
- Date Recue/Date Received 2021-03-16

1. AU2014282964 **Figure. 2**

National Public Data Description



NG

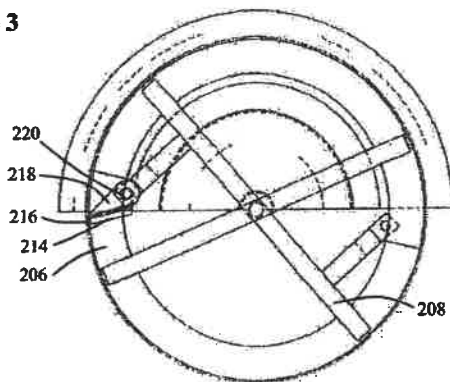
Figure 3

Image 2 of 9



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Figure 4

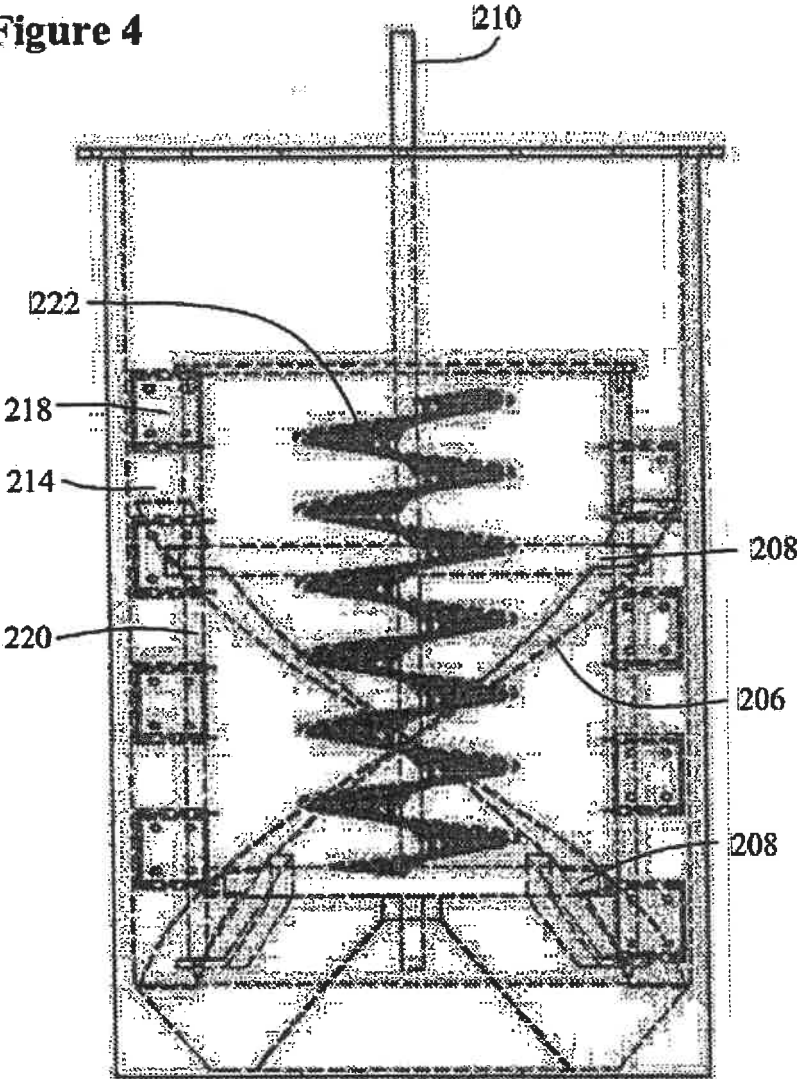
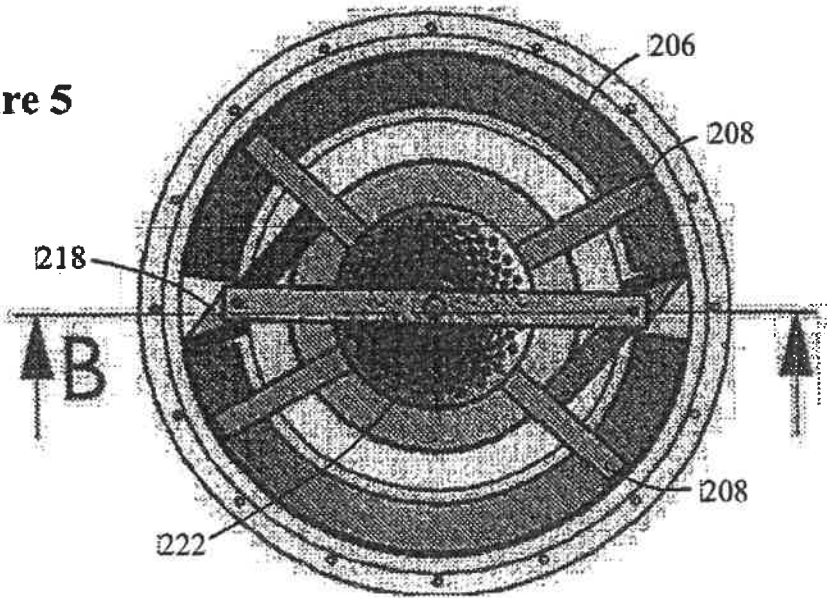


Figure 5



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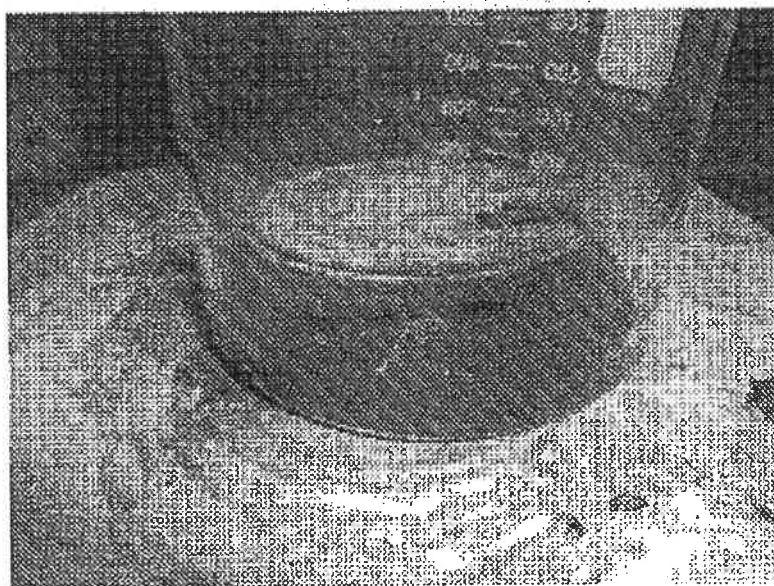


Figure 6

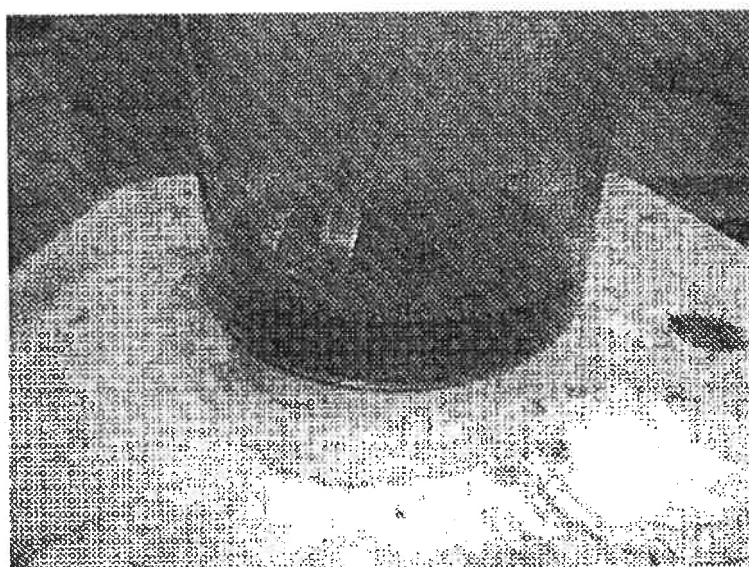


Figure 7

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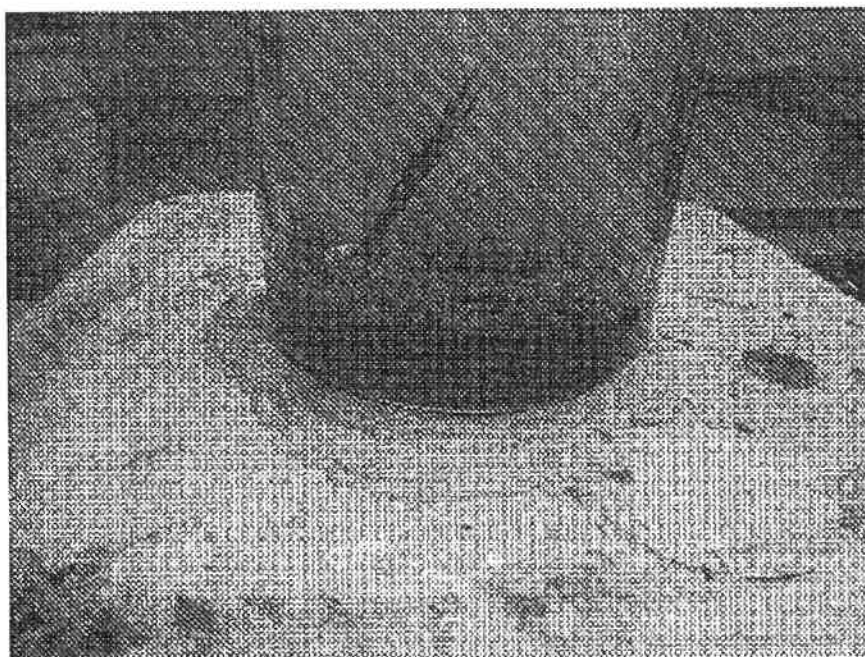


Figure 8

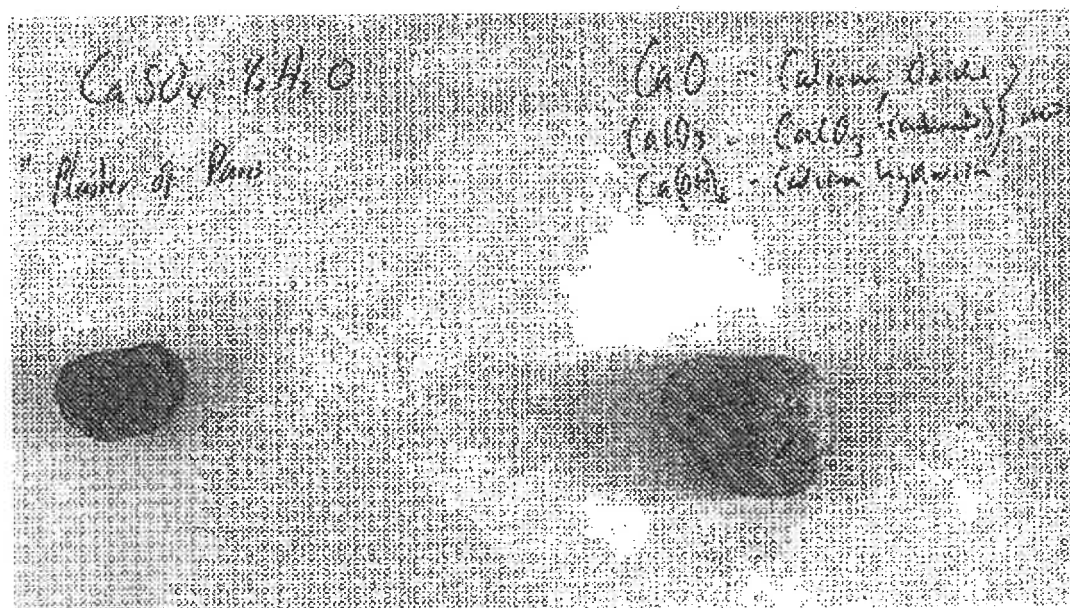


Figure 9

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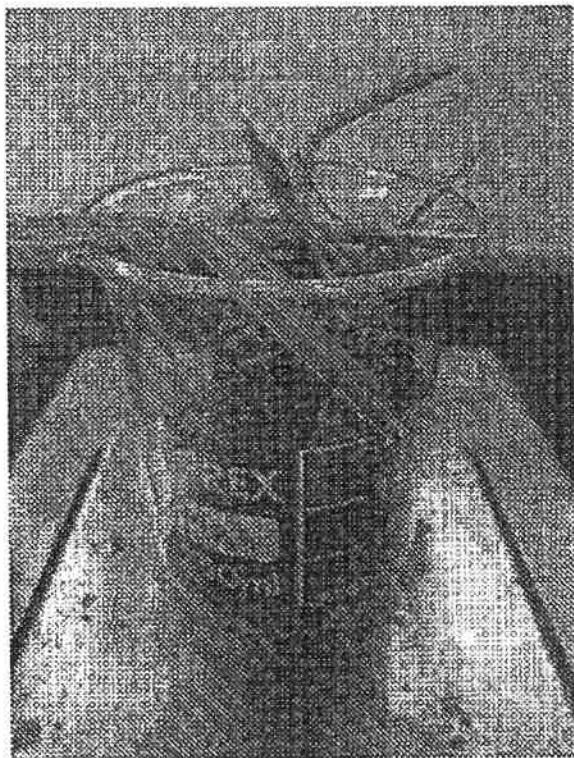


Figure 10



Figure 11

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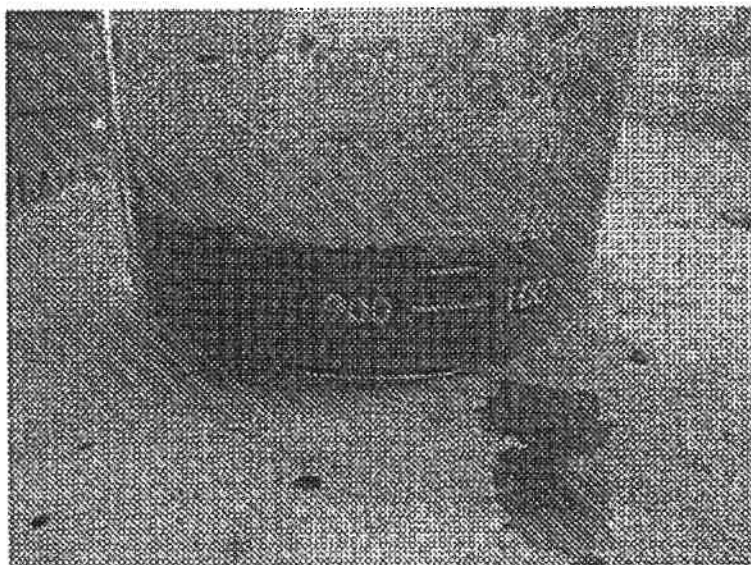


Figure 12



Figure 13

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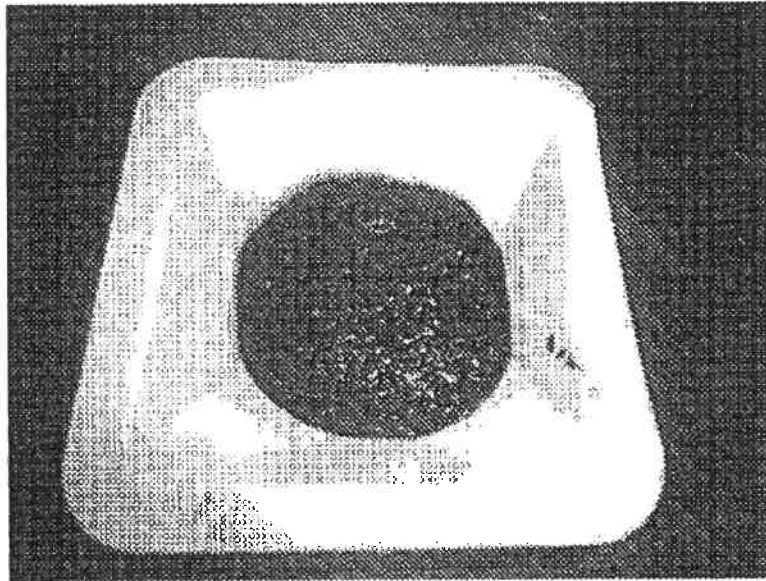


Figure 14

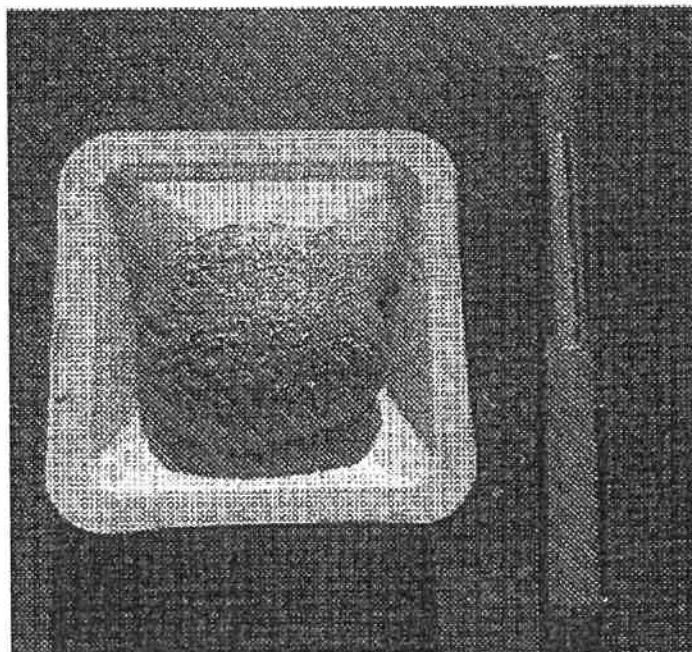


Figure 15

1. AU2014282964

National Bibliography Date Description

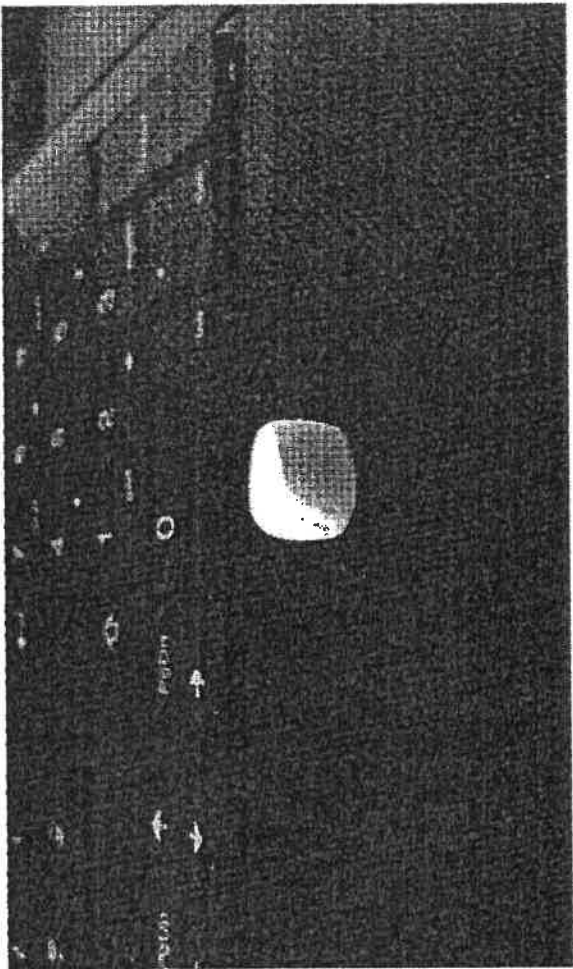


Figure 16

Image 9 of 9



Figure 16



Figure 13



Figure 15

14732609.4

1. EP3010871 - ORGANIC WASTE PROCESSING

National Biblio. DataDescriptionClaimsDrawingsPatent FamilyDocuments

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3010871

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CPC

C05D 3/02C05F 1/005C05F 3/00C05F 11/08C05F 9/00C05F 11/00

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Applicants

ELEMENTAL DIGEST LTD

Inventors

PALMER BRUCE WILLIAM

Designated States

View all

Priority Data

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201322874 23.12.2013 GB

Title

(DE) BEHANDLUNG VON ORGANISCHEN ABFALLSTOFFEN
(EN) ORGANIC WASTE PROCESSING
(FR) TRAITEMENT DE DÉCHETS ORGANIQUES

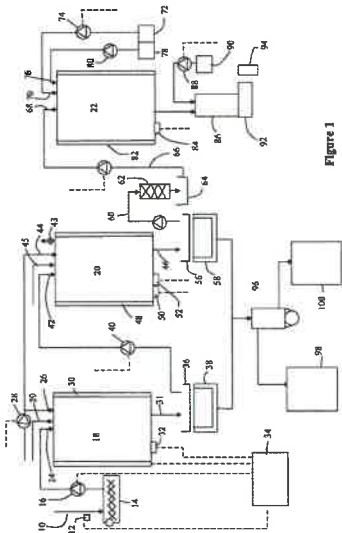


Figure 1

Abstract

(EN) The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertiliser from organic waste, especially slow release fertiliser. The present invention also relates to fertilisers made by the method of the invention, especially slow release fertiliser. The method of processing organic waste to provide a slow-release fertiliser may comprise at least one hydrolysis step to provide hydrolysed organic waste and a setting step to provide a slow release fertiliser.

(FR) La présente invention concerne des procédés de traitement de déchets organiques, en particulier, elle concerne un procédé de fabrication d'engrais à partir de déchets organiques, en particulier d'engrais à libération lente. La présente invention concerne également des engrais fabriqués au moyen du procédé de l'invention, en particulier un engrais à libération lente. Le procédé de traitement de déchets organiques destiné à fournir un engrais à libération lente peut comprendre au moins une étape d'hydrolyse pour fournir des déchets organiques hydrolysés et une étape de solidification pour fournir un engrais à libération lente.

Related patent documents

MYPI2015704654 MY-179142-A PL3010871 WO/2014/202986 JP2016527168 BR112015032008
TH159506 PH1/2015/502840 GB2531463 AU2014282964 EA201690065 US20160145164
CN105452196 CA2915964 MX366240 IN201627001450 NZ716125 C07590580



1. EP3010871 - ORGANIC WASTE PROCESSING



National Biblio. Data

Description

Claims

Drawings

Patent Family

Documents

PermaLink

Machine translation

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[EN]

Description

Field of the Invention

[0001] The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertiliser from organic waste, especially slow release fertiliser. The disclosure also relates to fertilisers made by the method of the invention, especially slow release fertiliser.

Background to the invention

[0002] Organic waste may be animal by-products, vegetable waste, food waste or a mixture of both.

[0003] During the processing of animals for meat production or other uses, relatively large amounts of the animal are removed, discarded and not sold on to an end user. Such discarded by-products include animal by-products such as bone, blood, gut content, wool, fur and feathers for example. Animal by-products also include animal waste such as chicken-litter, cow slurry and pig or horse manure. Vegetable waste may include kitchen waste, bedding from animals, fruit or vegetable processing waste, for example, fruit peels, sugar cane waste. This represents a significant waste of the resources that could be harnessed for useful purposes. There is also a cost involved in storing and disposing of such by-products as waste.

[0004] Various processes are known for processing organic waste. It is known to use incineration and composting to process the waste into a usable form or extract energy from waste.

[0005] Organic waste comprises valuable nutrients and there is a need for methods of processing organic waste to provide useful products. There is also a need for and environmentally friendly alternatives to conventional fertiliser for crops.

[0006] Slow release or controlled release fertilisers are particularly valuable because they release nutrients to plants over a period of time, for example several months. This means that there is less danger of toxicity to plants and of fertiliser running off the land before it is taken up by plants. Slow release fertilisers can provide a sustained supply of nutrients through a longer period of the growing season compared to fertilisers that are easily soluble and release nutrients quickly

[0007] [W02012135317](#) discloses methods of producing fertiliser from organic waste comprising the step of heating the waste in the presence of sulphuric acid.

[0008] It is an aim of the present invention to provide a method for disposing of organic waste and a method for processing organic waste to provide a slow release fertiliser

[0009] The invention is defined by the claims. Any subject matter falling outside the scope of the claims is provided for information purposes only.

[0010] The present invention provides a method of processing organic waste comprising the step of combining the organic waste with a composition according to claim 1, and a setting step according to claim 1.

[0011] The method of processing organic waste may be a method of making a slow release fertiliser.

[0012] The method may be carried out in a suitable apparatus comprising one or more vessels. The one or more vessels may be equipped with a mechanism for lifting the mixture upwards in a central region of the vessel by means of a lifting mechanism, resisting downward movement of the mixture in an outer region of the vessel by means of a heat transfer member, and transferring heat within the mixture using the heat transfer member. The vessels may also be equipped with a heating mechanism to heat the contents of the vessel.

[0013] The heat transfer member may be rotatably mounted in the vessel and may have a leading end and a trailing end. The leading end may be below the trailing end such that rotation of the heat transfer member generates a lifting force acting on the by-product, which may be arranged to resist its downward movement.

[0014] The heat transfer member may be formed from a strip of metal. The heat transfer member may be part helical.

[0015] The lifting mechanism may comprise a helical lifting member rotatable to lift the by-product. The lifting mechanism and the heat transfer member may be mounted on a common apparatus may comprise scraping means, for example one or more scrapers, arranged to scrape the inside of the vessel wall, or to scrape by-product, or other contents of the vessel, from the side wall. The scraping means may be mounted on the common rotatable support means for rotation with the heat transfer member.

[0016] The vessel wall may be cylindrical, for example having a central axis that is vertical. The vessel may further comprise heating means arranged to heat the by-product in the vessel.

[0017] The apparatus may further comprise a further processing vessel also comprising: a vessel wall; a lifting mechanism, which may be located in a central region of the vessel, arranged to lift the by-product within the vessel; a metal heat transfer member, which may extend around the lifting mechanism and may be in an outer region of the vessel, the heat transfer member optionally being rotatable around the outer region of the vessel to resist downward movement of the by-product in the outer region of the vessel and to dissipate heat through the by-product in the outer region of the vessel.

[0018] The vessel, or one of the vessels, may have an inlet connected to a supply of water. The vessel, or one of the vessels, may have an inlet connected to a supply of steam. The vessel, or one of the vessels, may have inlet means connected to a supply of acid and hydrogen peroxide.

[0019] The method may further comprise heating the mixture in the vessel.

[0020] The method may be carried out in a plant.

[0021] The present invention may be carried out in an apparatus for processing organic waste the apparatus comprising a heating vessel having a product inlet arranged to receive the by-product, a water inlet arranged to receive water, and a steam inlet arranged to receive steam, and heating means arranged to heat the first heating vessel, a separator arranged to receive the contents of the first heating vessel and separate it into first and second components, and an acidification vessel having a product inlet arranged to receive the solid component, and acid inlet means arranged to receive acid. The acid inlet means may further be arranged to receive hydrogen peroxide.

[0022] Preferably the apparatus further comprises means to deliver a lime mix to the acidified by-product.

[0023] Preferably the apparatus further comprises means to deliver magnesium salts to the acidified by-product.

[0024] The apparatus may comprise a dryer operative to receive and dry the acidified by-product.

[0025] A by-product sizing apparatus may be provided operative to re-size the by-product into pieces below a predetermined size.

[0026] The sizing apparatus may comprise a crusher operative to crush the raw organic waste into pieces below a predetermined size.

[0027] The sizing apparatus may comprise a plurality of crushers arranged in sequence and operative to sequentially crush the raw organic waste into smaller pieces.

[0028] Preferably the sizing apparatus is operative to output pieces of less than 40mm.

[0029] A metal detector may be provided at the inlet operative to detect any metal content in the raw organic waste and to generate a corresponding signal that is used to prevent raw organic waste entering the by-product sizing apparatus if metal content is detected.

[0030] Preferably the apparatus further comprises a pre-heating vessel operative to pre-heat the sized by-product and to deliver the sized by-product to the heating vessel.

[0031] The apparatus may further comprise a supply of at least one additive and means to deliver the additive to the further heated by-product to alter the properties of the further heated by-product.

[0032] The apparatus may comprise a pellet press or granulator operative to receive dried by-product and to press or granulate the dried by-product into pellets or granules.

[0033] The apparatus may further comprise a packaging unit or bagging machine operative to receive and package or bag the pellets or granules.

[0034] The method may comprise a step of sizing the organic waste, for example by crushing, grinding or chopping solid waste to provide suitable sized pieces. Suitable sized pieces may be less than 200mm, less than 150 mm, less than 100 mm, less than 50 mm or less than 20 mm sized pieces.

[0035] Water may be added to the organic waste before or after sizing the organic waste.

[0036] The method may comprise a heating step. The organic waste may be heated, for example to between 70 °C and 200°C. The organic waste may be heated for between 10 and 60 minutes. The organic waste may be heated for 20 minutes at 135 °C and 3 bar pressure to provide microbiological safety. The organic waste may be heated for a time and/or temperature as specified in legislation regarding disposal of organic waste or animal waste and/or microbiological safety. The organic waste may be heated by adding steam to the organic waste, for example steam may be added in a vessel as shown at [18] or [20] in Figure 1. The organic waste may be heated at higher than atmospheric pressure, for example in a pressurised vessel shown at [20] in Figure 1. The organic waste may be heated by heating the vessel in which the organic waste is held. The organic waste may be heated by an exothermic reaction when a hydrolysing composition is added to the organic waste, for example in a vessel as shown at [22] in Figure 1. The method may comprise one or more, two or more, three or more, four or more or five or more heating steps. Each heating step may be performed in a different vessel for example a heating step may be

performed in each of vessels [18], [19] and [20] shown in Figure 1. The method may comprise two or more heating steps performed at different temperatures or pressures. Heating steps may be used to remove gelatine and fat; degrade the organic waste and/or provide a microbiologically safe product from organic waste.

[0037] The heating step may be an independent step in the method of processing organic waste performed at a different time from other steps. The heating step may be performed at the same time as another step in the method, for example the hydrolysis step. Combining the organic waste with a hydrolysing composition may cause an exothermic reaction that heats the organic waste. During the heating step the organic waste may be heated using a combination of heating methods.

[0038] During the acid-peroxide/high temp/high pressure and/or digesting step a metal catalyst is added. The metal catalyst is a suitable metal or metal compound that is able to increase the rate of reaction with respect to breaking apart protein material such as the hydroxyproline in collagen into the constituent amino acid building blocks. The subsequently released nitrogen atoms from the peptide links are released in the form of ammonium ions. In particular these catalysts could be very useful in the dissolving of bone in acid/peroxide digestion step. The catalysts are copper or copper compounds specifically copper II sulphate or copper II oxide, titanium or titanium compounds but specifically titanium dioxide as either a powder or a nano-powder. Some of the catalysts, such as copper, would add benefit to the micronutrient composition of the final fertiliser product. The catalyst may be added in very small quantities, for example the mass of catalyst may be in the range of 0.1% - 0.01% of the mass of undigested organic waste, for example bone paste.

[0039] The method may comprise a step of draining liquid from the organic waste. The step of draining liquid from the organic waste may be performed after the heating step. Liquid may be removed from the mixture. The liquid may comprise fat and/or gelatine which may be separated and processed separately.

[0040] The method comprises one or more digesting steps where a digesting composition is combined with the organic waste. The composition is a combination of sulphuric acid H_2SO_4 50-100% (for example 95%) followed by or mixed with Hydrogen Peroxide H_2O_2 5- 100% (for example 30% v/v) in a ratio of 1:1 to 1:10 sulphuric acid solution : hydrogen peroxide, and a metal catalyst, wherein the metal catalyst is titanium, copper, or compounds thereof.

[0041] The method may comprise one hydrolysis step, two hydrolysis steps or three hydrolysis steps. The hydrolysis steps may be performed sequentially. For example, the method may comprise an acid digesting step and a base hydrolysis step performed sequentially.

[0042] The digesting step may be an exothermic reaction and may cause the organic waste mixture to heat up. Heat may be removed from the organic waste mixture during this step to keep the mixture within suitable temperature limits.

[0043] If the digesting step is exothermic it may contribute heat to the heating step.

[0044] The one or more digesting steps may be performed in a vessel, for example as shown at [22] in Figure 1.

[0045] After the one or more digesting steps the digested organic waste may provide an acidified digested organic waste.

[0046] The method may comprise a step of draining liquid from the organic waste.

[0047] The step of draining liquid from the organic waste may be performed after the digesting step. Liquid may be removed from the mixture. The liquid may comprise fat and/or gelatine and/or other components which may be separated from each other and processed separately.

[0048] The method may comprise an acidification step where the digested organic waste is combined with additional acid to bring the pH of the organic waste to an acid pH below pH 7. An acid, for example phosphoric acid, nitric acid, or sulphuric acid may be added if the organic waste mixture is not acidic enough after the digesting step. The digested organic waste may be acidified to a pH between pH -2 and pH 7, between pH 1 and pH 7, between pH 2 and pH 6, between pH 3 and pH 6 or between pH 4 and pH 5. The amount of acid or base added may be enough to make the organic waste mixture slightly acid, for example pH 6.5 to pH4.

[0049] The one or more acidification or neutralisation steps may be performed in a vessel, for example as shown at [22] in Figure 1.

[0050] The method may further comprise a step of adding nutrients or minerals, for example magnesium salts to the organic waste mixture. This is advantageous because additional nutrients that are required in a fertiliser may be added to the processed organic waste so that it can be made into a fertiliser with a suitable balance of nutrients. Nutrients or minerals may be selected in order to complement the known components of the organic waste to provide nutrients or minerals that are known to be lacking in the organic waste. Nutrients or minerals may be selected to address a deficiency in the area where the processed organic waste is going to be used as a fertiliser.

[0051] The method may further comprise a step of adding ammonia and/or other basic or carbonate materials to increase the nitrogen or other minerals in the organic waste so that the final fertiliser has higher nitrogen or other minerals. For example ammonia and/or other minerals may be added to the organic waste at any point in the processing in order to increase the nitrogen content, the phosphate content or the potassium content of the final fertiliser product. Ammonia and/or other minerals may be added to the organic waste at any point in the processing in order to produce a final fertiliser product with a particular balance of nitrogen phosphate and potassium (NPK content) that is suitable for a specific soil type and/or crop type.

[0052] The method may comprise the step of drying the neutralised organic waste mixture. This may be done by a combination of heating and the addition of a dehydrating granulating mixture followed by further heating in the granule drying stage of the process. The acidified organic waste mixture may be dried until it comprises 10-12% moisture.

[0053] The method comprises a step of setting the organic waste mixture by combining acidic, digested organic waste with a setting mixture or liming mixture. This may be any compound or composition that can form a lime mortar.

[0054] The method may comprise a re-sizing step to re-size the organic waste mixture into pieces below a predetermined size. The re-sizing step may, for example, comprise crushing, chopping or grinding the product. The product may be resized to pieces of less than 40mm.

[0055] The method may further comprise adding at least one additive to alter the properties of the organic waste mixture. For example, an unpalatable ingredient may be added to deter animals from eating the fertiliser.

[0056] The method may comprise the step of pelleting or granulating the organic waste mixture after the setting step.

[0057] Granular fertilisers may be treated with hydrophobic anti-caking agents such as Kaolin by externally coating of the granules, or incorporated as part of the granulation mix; and or paraffin based additives which may be sprayed onto hot granules.

[0058] The method may further comprise a packaging step where the pellets or granules are packaged.

[0059] Organic waste may comprise, for example, waste from abattoirs, waste from meat processing or packaging, fallen stock, animal carcasses, food waste, animal excrement, for example, cow slurry, pig or cow dung, poultry litter, animal bedding, waste from the food industry, abattoir blood waste. Organic waste may be animal waste, vegetable waste or mixed animal and vegetable waste.

[0060] The method of processing organic waste is a method of making fertiliser.

[0061] The digesting composition according to claim 1 may provide suitable ingredients to the fertiliser product, for example, use of nitric acid in the digesting composition may provide additional nitrogen in the fertiliser product. Use of phosphoric acid in the digesting composition may provide additional phosphorus in the fertiliser product. Use of sulphuric acid in the digesting composition provides additional sulphur in the fertiliser product.

[0062] The digesting composition according to claim 1, neutralising composition and additives may be chosen to provide a suitable level of potassium, phosphorous and nitrogen in the final fertiliser when they are combined with the organic waste.

[0063] The digesting composition is selected because of its suitability for digesting the particular components of the organic waste that it is combined with. The digesting composition according to claim 1 comprising sulphuric acid is particularly suitable for hydrolysing organic waste comprising bone.

[0064] The method according to the present invention is a method of processing organic waste to provide a fertiliser comprising at least one digesting step.

[0065] The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step and at least one digesting step.

[0066] The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step, at least one digesting step and at least one neutralising step.

[0067] The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step, at least one digesting step, at least one neutralising step and at least one pelleting step.

[0068] The method of the present invention may be a method of processing organic waste to provide a slow-release fertiliser comprising at least one digesting step to provide digested organic waste and one setting step to provide a slow release fertiliser.

[0069] The at least one digesting step comprises an acid digesting step according to claim 1 to provide an acidified digested organic waste.

[0070] The at least one digesting step may comprise an acid digesting step and a base hydrolysis step to produce an acidified digested organic waste.

[0071] The at least one digesting step comprises an acid digesting step using sulphuric acid, to digest organic waste and produce an acidified digested organic waste.

[0072] The acidified digested organic waste comprises sulphuric acid.

[0073] The method comprises a step of setting the acidified digested organic waste mixture by combining acidic, digested organic waste with a setting mixture or a liming mixture. This may be any compound or composition that can form a lime mortar. The organic waste may be bonded to the lime mortar or embedded in it. The liming mixture may comprise, for example lime, calcium sulphate CaSO_4 , Calcium carbonate CaCO_3 , Calcium oxide CaO and/or Calcium hydroxide Ca(OH)_2 .

[0074] The liming mixture may further comprise clay bases materials such as kaolin.

[0075] The liming mixture may set to provide a lime mortar. During the setting step the organic waste and any additives that have been added to it may become bonded to the lime mortar or set within the lime mortar to provide a slow release fertiliser.

[0076] The lime mortar may degrade when left exposed to the elements or in the soil over a period of 1 to 6 months, or over a period of more than 9 months, or over a period of 3 to 9 months, or over a period of 3 to 6 months, or over a period of 3 to 5 months or over a period of 4 to 5 months. The lime mortar may degrade more quickly in soil that is more acidic.

[0077] The composition of the setting mixture or liming mixture may be adjusted to provide a slow release fertiliser that degrades faster or slower depending on what is required. In order to adjust the time that the fertiliser takes to degrade different proportions of each of the liming mixture ingredients may be added to the acidified organic waste. For example, the liming mixture can be adjusted so that more calcium sulphate is produced in the setting step to provide a fertiliser that breaks down more slowly. More calcium sulphate may be produced by reducing the amount of lime in the liming mixture so that the excess sulphuric acid reacts with the lime to make more calcium sulphate. The liming mixture can be adjusted so that it degrades faster by including more lime in the liming mixture.

[0078] The specific conditions where the slow release fertiliser will be used may be considered when selecting the composition of the organic waste and the composition of the liming mixture. The specific conditions may be the pH of the soil, the expected rain fall, the type of crop, the nutrient requirements of the crop, the nutrient profile of the soil, the ambient temperature that the crop will grow at, the length of growing time of the crop and/or the rate at which the crop requires nutrients during particular phases of its growth.

[0079] The percentage of liming mixture compared to acidified organic waste may be adjusted to provide a fertiliser that degrades faster or slower. The percentage of liming mixture used to set the acidified organic waste may be 10% to 50%, may be 20% to 50%, or may be 30% to 50%, or may be 40% to 50%

of the mass of the organic waste.

[0080] The percentage of liming mixture added compared to the amount of acidified organic waste may be adjusted to provide a fertiliser with a higher or lower neutralising value. Neutralising value may be the amount of base compounds in the fertiliser. For example, a higher neutralising value may be required to partially neutralise acid soil or to provide lime to crops that prefer alkaline conditions. The percentage of liming mixture used to set the acidified organic waste may be 10% to 50% of the mass of the organic waste.

[0081] Lime mortar may comprise lime, calcium sulphate CaSO_4 , Calcium carbonate CaCO_3 , Calcium oxide CaO and/or Calcium hydroxide Ca(OH)_2 , which react with water and/or acid and set to form solid lime mortar. The liming mixture may further comprise clay bases materials such as kaolin to change the properties of the lime mortar. Other components of the digested organic waste may become part of the lime mortar by chemically bonding to components of the lime mortar or by being immobilised inside the lime mortar but not chemically bonded to it. As the lime mortar degrades in soil it may release lime into the soil and may also release components of the organic waste into the soil which provide nutrients and act as a fertiliser.

[0082] The setting step may comprise admixing the acidic mixture with lime and/or one or more calcium compounds wherein the calcium compounds are able to react with water and/or acid to form lime mortar which sets the mixture to form a slow release fertiliser. One or more calcium compounds within the liming mixture may react with sulphuric acid to form calcium sulphate which sets the mixture to form part of the lime mortar of the slow release fertiliser.

[0083] The lime mortar may set the mixture to form a solid slow release fertiliser with the components of the acidic mixture bonded inside the lime mortar matrix.

[0084] The setting step may comprise admixing the acidic digested organic waste mixture with one or more selected from calcium oxide, calcium carbonate and calcium hydroxide to form a slow release fertiliser.

[0085] The setting step may comprise admixing the acidic mixture with calcium oxide and calcium carbonate to form a slow release fertiliser.

[0086] The setting step may comprise admixing the acidic mixture with calcium oxide, and calcium hydroxide to form a slow release fertiliser.

[0087] The setting step may comprise admixing the acidic mixture with calcium carbonate and calcium hydroxide to form a slow release fertiliser.

[0088] The proportions of acidic mixture, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product is slightly alkali.

[0089] The proportions of acidic mixture, lime, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product takes more than a month, more than two months, more than 3 months, more than 4 months, more than 5 months or more than 6 months to degrade in soil. The soil may be between pH 5 and pH 7. The soil may be between pH 5 and pH 9.

[0090] The proportions of acidic mixture, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product takes between one and 6 months, between 1 and 4 months or between 2 and 3 months to degrade in slightly acidic soil. For example the soil may be between pH 5 and pH 7.

[0091] The method may comprise the step of adding one or more minerals to provide suitable nutrients in the slow release fertiliser.

[0092] The method may comprise the step of adding one or more fungicides to provide antifungal properties to the slow release fertiliser.

[0093] The method may comprise the step of adding one or more pesticides to provide pesticide properties to the slow release fertiliser.

[0094] The method may comprise the step of adding one or more herbicides to provide herbicide properties to the slow release fertiliser.

[0095] The method may comprise the step of adding one or more unpalatable compounds deter animals from eating the slow release fertiliser.

[0096] The method may comprise the step of processing the slow release fertiliser, for example by pelleting, granulating, press-forming or powderising the slow release fertiliser.

[0097] The method may comprise the step of coating the fertiliser, for example coating pellets or granules of the fertiliser, with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds.

[0098] In a second aspect the present disclosure provides a slow release fertiliser provided by the method of the present invention.

[0099] In a further aspect the present disclosure provides a slow release fertiliser comprising a proportion of lime mortar to set the fertiliser.

[0100] The slow release fertiliser may comprise more than 10%, more than 20%, more than 30%, more than 40%, more than 50%, more than 60% or more than 70% lime mortar.

[0101] The slow release fertiliser may comprise between 10% and 70%, between 20% and 50% or between 30% and 40% lime mortar.

[0102] The slow release fertiliser may comprise lime mortar that takes between one and 6 months, between 1 and 4 months or between 2 and 3 months to degrade in slightly acidic soil. For example the soil may be between pH 5 and pH 7.

[0103] The slow release fertiliser may further comprise one or more fungicides, one or more pesticides, one or more herbicides, one or more pest control agents and/or one or more unpalatable compounds. Pellets of the fertiliser may be coated with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds.

[0104] In a further aspect the present disclosure provides a composition obtained or obtainable by the method of the invention.

[0105] In a further aspect the present disclosure provides a fertiliser composition obtained or obtainable by the method of the present invention.

[0106] There now follows by way of example only a detailed description with reference to the accompanying drawings, in which;

Figure 1 is a schematic view of an apparatus for processing organic waste;

Figure 2 is a cut-away side view of a heating vessel forming part of the plant of Figure 1;

Figure 3 is a plan view of the vessel of Figure 2;

Figure 4 is a cut-away side view of a heating vessel of Figure 2 with further components shown; and

Figure 5 is a plan view of the vessel of Figure 4;

Figure 6 shows a beaker containing 95g of congealed frozen blood and 30 g of water;

Figure 7 shows a view of the hydrolysis and heating stage being performed on 95 g of blood with 30 g of water [not according to the invention]. 10g of potassium Hydroxide pellets have been added and the mixture was stirred for two minutes to form a deep red solution. This reaction is very exothermic and so hydrolysis and heating happen at the same time;

Figure 8 shows the mixture of figure 7 with 8g of 70% [1.42 specific gravity] nitric acid added to neutralise the potassium hydroxide this step is also very exothermic and is the neutralisation step and a heating step at the same time [not according to the invention]. The resulting mixture is an alkaline solution of potassium nitrate containing dissolved hydrolysed blood;

Figure 9 shows a view of the solid set product that results when the mixture shown in figure 8 is combined with $\text{CaSO}_4 \cdot \frac{1}{2} \text{H}_2\text{O}$ (left hand side) or with CaO , CaCO_3 and Ca(OH)_2 ;

Figure 10 shows a view of 63 g of chicken litter and bedding;

Figure 11 shows a view of the 63 g of chicken litter and bedding mixed with 150 ml and blended to form a workable paste;

Figure 12 shows a view of the blended mixture shown in Figure 11 with 3 g [0.0306 mole] of H_3PO_4 [orthophosphoric acid [1.75 specific gravity] added and the mixture brought to the boil. This is the heating stage and a first hydrolysis stage [acid hydrolysis] done at the same time [not according to the invention].

Figure 13 shows a view of the mixture in Figure 12 with 0.0918 mol of potassium hydroxide (5.14 g) added to neutralise the acid, then 5 g more potassium hydroxide added to increase the pH to 14 and hydrolyse fat, proteins, lipids and nucleic acids, this stage is highly exothermic and is therefore a second hydrolysis step [base hydrolysis] performed at the same time as a heating step [not according to the invention].

Figure 14 shows a view of the resulting mixture after pH of the mixture shown in Figure 13 is neutralised using using 8 g of 70% [1.42 specific gravity] nitric acid;

Figure 15 shows a view of the mixture shown in Figure 14 after addition of calcium carbonate, calcium hydroxide and calcium oxide to produce a solid pellet; and

Figure 16 shows a pellet of fertiliser made from chicken litter using a process of the present disclosure.

[0107] The method of the present invention may be carried out in an apparatus for processing organic waste, an example of which is set out below and in Figures 1 to 5.

[0108] Referring to Figure 1, an apparatus for processing organic waste comprises an inlet 10 which receives raw organic waste. A metal detector 12 is provided at the inlet 10 to detect any metal in the raw organic waste and to generate a control signal indicative of any metal content detected. The inlet 10 feeds into a particle sizing module, which in this case is a crushing module 14, arranged to reduce the raw product to pieces of a suitable size. In this embodiment the crushing module 14 comprises a first crusher arranged to crush the raw organic waste into pieces of less than a predetermined size, said size being 150mm in this example, and a second crusher arranged to crush the raw organic waste into smaller pieces, of less than 40mm in this example. A conveyor 16, which may be a screw conveyor, is arranged to transfer the crushed product from the outlet of the crushing module 14 to the first 18 of three heating vessels 18, 20, 22.

[0109] The first heating vessel 18 comprises an enclosed steel container having an inlet 24 at the top through which the crushed product can be introduced, a further inlet 26, also at the top, which is connected to a water supply via a pump 28 so that water can be added to the product in the vessel 18, a further inlet 29 connected to a supply of steam, and an outlet 31 at the bottom. Flow control valves are provided in the water and steam inlets 26, 29, and the outlet 31. A heater 30 is provided around the wall of the vessel, and a temperature gauge 32 is arranged to measure the temperature in the vessel 18. A controller 34 is arranged to control the operation of the whole plant, including all the conveyors, pumps and heaters, and the inlet and outlet valves, as will be described in more detail below.

[0110] A first separation vessel 36 is arranged to receive the product from the outlet 31 from the first heating vessel, and is arranged to hold the solid product and allow liquid to drain into a collection tank 38 situated below it. A second conveyor 40 is arranged to transport the solid product from the separation vessel 36 to the inlet 42 of the second heating vessel 20.

[0111] The second heating vessel 20 is pressurized, and has a pressure vent controlled by a vent valve 43 so that the pressure in the vessel 20 can be controlled. It also has a further inlet 44 for water, a further inlet 45 for steam, and an outlet 46 at its bottom end, each with a flow control valve controlled by the controller 34. It also has a heater 48 around its side walls, and a temperature gauge 50 and pressure gauge 52 arranged to measure the temperature and pressure of the contents of the second vessel 20. These are connected to the controller 34 which is arranged to control the temperature and pressure, as well as the quantities of water, steam, and product, in the second vessel 20 as required.

[0112] A second separation vessel 56 is arranged to receive the product from the outlet 46 from the second heating vessel 20, and is arranged to hold the solid product and allow liquid to drain into a collection tank 58 situated below it. A third conveyor 60 is arranged to transport the solid product from the separation vessel 56 to a second crusher unit 62. The second crusher unit 62 is arranged to break down the solid product into pieces no larger than 2mm.

[0113] The outlet from the crusher unit 62 is arranged to output the product into a pan 64, which may be arranged to pre-heat the product prior to it being

received in the third heating vessel 22. A further conveyor 66 is arranged to transport the product from the pan 64 to the inlet 68 of the third heating vessel 22.

[0114] The third vessel 68 also has a further inlet 70 at its top end connected to a source 72 of acid via a pump 74, and a further inlet 76 connected to a source 78 of hydrogen peroxide, via pump 80. These are controlled by the controller 34 so that acid and hydrogen peroxide can be added into the vessel 22 in the required amounts and rates and at the required time. The third vessel also has a heater 82 and a temperature gauge 84 connected to the controller 34. An outlet 84 at the bottom end of the vessel 22 is connected to a neutralization vessel 86 which is arranged to receive the product from the third heating vessel 22, and also arranged to receive an alkaline additive, in this embodiment a lime mix, via a conveyor 88, from an alkaline additive source 90, under the control of the controller 34.

[0115] The final neutralized product is in the form of a paste and an extruder 92 is arranged to receive this product from the neutralization vessel 86 and extrude it into pellets, and a packaging module 94 is arranged to receive the extruded pellets and package them for transport away from the plant.

[0116] Each of the liquid collection vessels 38, 58 has an outlet that is connected to a separator in the form of a centrifuge 96 which is arranged to separate out the lighter and heavier components of the liquid. The lighter components will generally comprise fats, and the heavier components generally comprise gelatine, and these components are collected in separate collection vessels 98, 100.

[0117] Operation of the plant will now be described with particular reference to the treatment of bone, although it will be appreciated that it can be used for a wide variety of other waste products.

[0118] Raw bone is introduced into the crusher module 14 where it is crushed into pieces no larger than 40mm. From there it is transferred into the first vessel 18.

[0119] The controller 34 is arranged to monitor the amount of product entering the vessel 18, for example using a load cell to measure its weight, and to stop the transfer of product when a specified amount has entered the vessel. The controller 34 is also arranged to control the introduction of water and steam into the vessel 18, and to heat the vessel 18 to the required temperature. The bone and water mixture is heated to approximately 95°C to form an organic waste slurry which is then transferred to the separation vessel 36.

[0120] In the separation vessel 36, liquid drains from the slurry into the collection tank 38 while the slurry is transferred along the vessel. The drier slurry is then transferred via the conveyor 40 to the second heating vessel 20. The amount of slurry introduced into the second vessel 20 is controlled by the controller, for example using weighing cells to weigh the amount added, and then controlled amounts of water and steam are added by controlled operation of the inlet valves. The second vessel 20 is then heated up to about 135°C, and the vent valve 43 closed so that the pressure increases to about 5bar. This temperature and pressure is maintained for about twenty minutes.

[0121] At the end of the treatment in the second vessel 20, the vent valve 43 is opened to reduce the pressure in the vessel, and the outlet 46 is opened to allow the product, which is still in the form of a slurry, to empty into the second separation vessel 56. From that vessel 56, liquid drains into the second collection vessel 58, and the solid component of the slurry is transferred via conveyor 60 into the second crusher unit 62, where it is crushed to a particle size of about 2mm and transferred to the pre-heating vessel 64, where it is preheated.

[0122] From the pre-heating vessel 64 the solid product is transferred via conveyor 66 into the third heating vessel 22 via the conveyor 66 and the inlet 68. Again the amount of product introduced is controlled, and also sulphuric acid and hydrogen peroxide are added in controlled amounts via the inlets 70, 76. The reaction in the vessel 22 is exothermic and the temperature rises to about 95°C, and then falls off when the reaction is complete. This heat and the acid and hydrogen peroxide kills any disease or bacteria in the product, and also oxidises any carbon in the product resulting in an inorganic product which is suitable for use as a fertiliser.

[0123] When the acidification step is complete, the outlet 22 from the third vessel 22 is opened and the product transferred to the neutralization vessel 86, to which lime mix is added in controlled quantities in order to increase the pH of the mixture to neutralize the product. The lime mix may comprise a mixture of chalk and lime, or quick lime for example. The lime mix may also comprise additives to alter the properties of the mix, such additives for example comprising selected macro/micro nutrients. The lime mix, and/or the additives, can be varied as required to deliver the required fertilizer properties. The lime mix causes an exothermic reaction. This serves to at least partially dry the acidified organic waste.

[0124] The increased pH mixture, which is typically in the form of a paste, is then dried if necessary, and passed into the extruder 92, from which it is extruded as pellets or granules, and packaged at packaging module 94 for transport.

[0125] The liquid from the collection vessels 38, 58 is separated in the centrifuge 96, with fat being transferred to fat storage tank 98 and gelatine to the gelatine storage tank 100.

[0126] The third heating vessel 22 may also comprise additive inlets arranged to deliver further additives to the slurry. These additives can be of any suitable ingredient to adjust the mineral or nutritional content of the solid slurry particles to suit a particular use or location of use. Thus the additives can be controlled to alter the macro/micro nutrient bases of the solid slurry particles to suit different soil or ground conditions for example, or to provide optimum nutrition for a particular type of crop.

[0127] Referring to Figures 2 and 3, the first heating vessel 18 comprises a cylindrical steel body 200, about 1.25m high and 750mm in diameter, arranged so that its central axis 202 and curved side walls 204 extend vertically. A pair of part-helical heat transfer blades 206 are mounted by means of cross members 208 on a central rotatable shaft 210 which is located on the central axis 202. Each of the heat transfer blades 206 is formed of a strip of sheet metal 209 about 15 mm wide, which is formed into a part-helix shape extending around the outer part of the vessel close to the side wall 204 but spaced slightly from it. The blades each extend from a point about two thirds of the way up the side of the vessel, down to close to the bottom of the vessel, and through half a turn around the central axis. The two blades 206 are diametrically opposite each other, so that they extend around opposite sides of the vessel.

[0128] A pair of scrapers 212 are also supported on the cross members 208. Each scraper 212 comprises a flat scraper blade 214 extending vertically down the side wall 204 of the vessel with an outer scraping edge 216 just clear of the side wall 204. Each of the scraper blades 214 is supported on a set of support brackets 218, which in turn are supported on a vertical support pole 220 which extends vertically between the cross members 208 near their outer ends.

[0129] Referring to Figures 4 and 5, an inner helical lifting flight 222 is also mounted on the shaft 210. The lifting flight 222 is formed of a flat strip of metal 224

about 100mm wide, formed into a helix centred on the shaft 210, and turning through about seven turns. The top of the lifting flight 222 is about level with the top of the heat transfer blades 206 and the bottom of the lifting flight 222 is slightly above the bottom of the heat transfer blades 206. The turning sense of the lifting flight 222 is the same as both of the heat transfer blades 206. A motor 224 is arranged to rotate the shaft 210, so that the lower end of the lifting flight forms a leading end and the upper end of the lifting flight forms a trailing end, and the lifting flight provides lift in the central part of the vessel 18 close to the central axis 202. At the same time, the lower ends of the heat transfer blades 206 form leading ends and the upper ends of the heat transfer blades form trailing ends, and the heat transfer blades also provide a small amount of lift, or in fact resistance to downward flow, in the outer part of the vessel closer to the outer wall 204. As the shaft rotates, the scrapers 212 move around the wall 204 of the vessel, scraping from the wall any of the by-product or other contents of the vessel that may have accumulated on the wall.

[0130] The central shaft 210 is rotated as the product is introduced into the vessel 18 through the inlet 24, the water is introduced through the inlet 26, this forms a slurry, and the steam is introduced through the inlet 29 which helps to heat the slurry. As the heater 30 heats up the contents of the vessel 18 and the steam is injected into it, the central lifting flight 222 causes a general upward flow of the mixture in the central area of the vessel. The mixture then moves outwards at the top of the vessel 18 and cascades downwards in the outer region of the vessel 18 close to the wall 204. The rotating heat exchange blades 206 provide some resistance to the downward flow, and therefore mix the mixture, but also help to conduct heat through the mixture quickly allowing it to be heated quickly to the required temperature. This helps to reduce the time required for the heating step carried out in the first heating vessel 18.

[0131] The second vessel 20 has the same mixing and heat exchange mechanism as the first vessel 18 as shown in Figures 2 to 5. In this vessel the heating process is similar to the first vessel, though at increased pressure, and the heat exchange coils and the lifting flight serve to distribute heat from the heater and steam throughout the vessel in a similar way. Similarly the third vessel 22 also has the same mixing and heat exchange mechanisms as shown in Figures 2 to 5. In this case the heat is generated by the exothermic reaction in the vessel. However, rapid heat dissipation is still helpful to ensure that the temperature remains approximately equal throughout the vessel, and that the reaction proceeds at about the same rate throughout the vessel.

[0132] It will be appreciated that various modifications can be made to the embodiment described above, and that the design of the heating vessel or vessels can be varied as appropriate for the process and products to be treated. For example, the two heat transfer blades could be replaced by one continuous helical coil, or more than two blades. Also the blades may not be strictly part-helical. For example the angle of the blade or blades to the vertical may vary along its length. Similarly the lifting flight may be of a different shape, or may be rotated independently of the heat transfer coils. Indeed a different type of lifting mechanism may be provided which is not a helical flight.

[0133] A number of pilot studies, described in the examples below, were performed to assess the effectiveness of different compositions for use in the hydrolysis step of the present method. The digesting step may take place in one of the reaction vessels, for example vessel (22) in Figure 1. The pilot studies were mostly done on a small scale but may be scaled up and the method may be performed on large quantities of organic waste, for example in an apparatus as shown in Figure 1. The inlets for acid and hydrogen peroxide described in relation to Figure 1 may alternatively be used to add other ingredients for the hydrolysis and neutralisation steps.

Example 1 Acid and Hydrogen Peroxide - Bone Paste (not according to the invention)

[0134] The preferred ranges of the temperature and reaction times.

[0135] Operating temperature range: Start temp of bone paste 10°C to 100°C giving an end temp after the digesting step of 100°C to 200 °C.

[0136] Preferred temp for process: Start temperature 50-90 °C - end temp 130-170 °C.

[0137] Ratio of wet bone paste to acid to hydrogen peroxide by mass 5-20:5:1 (wet bone:acid:peroxide), for example 13.5:5:1 (wet bone : acid : peroxide).

[0138] For example the ratio may be: 1000kg bone paste (wet) to 370kg of 95% sulphuric acid to 74kg of 30% hydrogen peroxide.

[0139] Treatment time Broad range - lmin to 30mins, for example, the treatment time may be 5-10 mins.

[0140] The amounts of acid and hydrogen peroxide - Broad Range - 1:1 to 1:10 Concentrated sulphuric acid to hydrogen peroxide solution.

[0141] Preferred ratio is 1:5 concentrated sulphuric acid to hydrogen peroxide for bone digesting but will vary for other waste streams.

[0142] Concentrated sulphuric acid used is 95% [1.83g/ml] but this could work over a range of 50 - 100%.

[0143] Concentrated hydrogen peroxide used is 30%w/v (weight to volume) but concentrations between 5-100% w/v although greater than 30% can carries an explosion risk.

Example 2 Base and Hydrogen Peroxide (not according to the invention)

Base peroxide

[0144] Broad range of concentrated ammonium hydroxide, and/or potassium hydroxide and/or sodium with hydrogen peroxide from 1:1 to 1:10 using 0.880sg [specific gravity] ammonium hydroxide [ammonia solution] and 30% hydrogen peroxide solution.

[0145] Preferred mixture 3:1 of 0.880sg ammonia solution to 30% hydrogen peroxide solution for the purpose of dissolving for example poultry litter, pig manure, cow slurry, cow manure, horse manure, contents of intestinal tract material from abattoir waste etc.

Example 3 Hydrolysis of Blood (not according to the invention)

[0146] Overview - blood contents are hydrolysed using sodium hydroxide or potassium hydroxide. The resulting solution is neutralised using nitric acid or sulfuric acid forming a potassium nitrate, sodium nitrate, sodium sulphate or potassium sulphate solution depending on the acid and base used solution containing the dissolved, heat treated, hydrolysed blood components.

Method:

[0147] To 100g of congealed blood add 30g of water and bring to the boil [water not required if blood is fresh and uncongealed]. NB - frozen blood used can be seen melting in Figure 1.

[0148] Hydrolysis: Add 10g of Potassium Hydroxide pellets KOH [solid, form preferred] or 7g of sodium hydroxide (solid form preferred) and stir for two minutes until deep red solution forms - very exothermic reaction. The product of this reaction is shown in Figure 2.

[0149] Neutralisation: Add 16g of 70% 1.42 specific gravity nitric acid to neutralise the potassium hydroxide [concentrated] although a dilute nitric acid solution would suffice - very exothermic. The resulting mixture is shown in Figure 3.

[0150] The resultant mixture is then treated with further nitric acid to achieve the desired pH for granulating - this could be acidic or alkaline.

[0151] Solution of potassium nitrate containing dissolved, hydrolysed blood.

If a small amount of solid precipitates from the blood it may be separated and hydrolysed/oxidised by a small, additional dose of nitric(or sulphuric) acid and hydrogen peroxide in a ratio of 5:1 before recombining with original solution for pasting or setting.

[0152] The neutralised hydrolysed blood may be blended with other soluble ingredients and the resultant solution sold as a liquid fertiliser or the neutralised hydrolysed blood may be granulated and sold as a solid granular form after setting using plaster of paris [CaSO₄.1/2H₂O] Figure 4 left or Calcium carbonate/oxide/hydroxide CaCO₃/CaO/Ca[OH]₂ Figure 4 right.

Example 4

[0153] Acid hydrolysis using phosphoric acid [other acids could be used - nitric acid or sulphuric acid] followed by alkaline hydrolysis using potassium hydroxide or sodium hydroxide - resultant mixture neutralised using nitric acid or sulphuric acid [not according to the invention].

[0154] Phosphoric acid increases the phosphorus content [P] and hydrolyses cellulose content. Nitric acid increases the N content.

[0155] Potassium hydroxide increases the K content and hydrolyses bulk content. Nitric acid neutralises the potassium hydroxide or sodium hydroxide forming potassium nitrate or sodium nitrate and increases the N content.

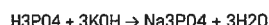
[0156] Approximate ratios by mass - 42(chicken litter) : 100(water) : 2(phosphoric acid) : 6.6(potassium hydroxide) or 4.7 sodium hydroxide: 8.0 (nitric acid).

[0157] Overview 63g of chicken litter and bedding is shown in Figure 5.

[0158] Addition of 150ml of water followed by 'blending' to form a suitable, workable paste shown in Figure 6.

[0159] Acid hydrolysis: - high-heat stage of process. 3g [0.0306 mole] of H₃PO₄ [Orthophosphoric acid 1.75 specific gravity] or 2.75g 70% nitric acid is added to drop pH to pH 3-4 and mixture brought to the boil [high heat stage of process].

[0160] Alkaline hydrolysis - in heliotherm stage. First the phosphoric acid is neutralised by potassium hydroxide:



3.00g of phosphoric acid [0.0306 mol] requires 0.0918 mol of potassium hydroxide = 5.14g then an addition 5g of KOH is required to facilitate the increase in pH to 14 and hydrolyse fat/protein/lipids/ nucleic acids etc (total 10g of KOH) - lots of heat evolved NaOH could also be used [less required - 7.14g].

[0161] Neutralisation - still in heliotherm stage.

[0162] Neutralisation using 17g of 70% [1.42 specific gravity] nitric acid.

[0163] Resultant mixture is an alkaline, sweet [molasses type smell] brown paste containing potassium nitrate, potassium phosphate and all the components of chicken litter suitable for plant uptake.

[0164] Addition of calcium carbonate/hydroxide/oxide/calcium sulphate heptahydrate mix to produce the solid pellet form is shown in Figure 10.

[0165] Sodium hydroxide is useful for the hydrolysis of blood or chicken litter followed by neutralisation using sulphuric acid. For example 100g of blood requires 5g of sodium hydroxide at around 90°C to hydrolyse, followed by neutralisation using concentrated sulphuric acid 3.6ml [6.63g] and then adjustment of pH to desired level.

1. EP3010871 - ORGANIC WASTE PROCESSING

National Biblio. DataDescriptionClaimsDrawingsPatent FamilyDocuments

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[EN] [DE] [FR]
Claims

1. A method of processing organic waste to provide a slow-release fertiliser comprising at least one digesting step to provide digested organic waste and a setting step to provide a slow release fertiliser, wherein said digesting step comprises combining the organic waste with a metal catalyst, 50-100% concentrate of sulphuric acid and 5-100% concentrate of hydrogen peroxide, wherein the ratio of sulphuric acid to hydrogen peroxide is 1:1 to 1:10, and wherein said catalyst is titanium, copper, or compounds thereof.
2. The method according to claim 1 wherein the method comprises a step of setting the digested organic waste by combining the digested organic waste with a liming mixture, optionally wherein the liming mixture is a compound or composition that can form a lime mortar.
3. The method according to claim 2 wherein the liming mixture comprises lime, calcium sulphate, Calcium carbonate, Calcium oxide and/or Calcium hydroxide, optionally wherein the liming mixture further comprises clay base materials such as kaolin.
4. The method according to any one of claims 2 or 3 wherein the liming mixture sets to provide a lime mortar with the digested organic waste bonded to the lime mortar or set within the lime mortar to provide a slow release fertiliser.
5. The method according to any one of claims 2 to 4 wherein the liming mixture is composed to provide a lime mortar that degrades, when left exposed to the elements or in the soil, over a period of 1 to 9 months.
6. The method according to any one of claims 2 to 5 further comprising the step of considering the composition of the digested organic waste and adjusting the composition of the liming mixture so that the digested organic waste and liming mixture will provide a slow release fertiliser that degrades over a desired time period in soil under specific conditions, and optionally wherein the mass of liming mixture compared to the mass of digested organic waste is adjusted to provide a fertiliser that degrades faster or slower and/or to provide a fertiliser with a higher or lower neutralising value.
7. The method according to any one of the preceding claims wherein the setting step comprises admixing the digested organic waste mixture with one or more selected from calcium oxide, calcium carbonate and calcium hydroxide to form a slow release fertiliser.
8. The method according to any one of the preceding claims further comprising the step of adding one or more minerals at any stage in the method to provide suitable nutrients in the slow release fertiliser, and/or optionally further comprising the step of adding one or more fungicides, pesticides, herbicides and/or unpalatable compounds to provide desirable properties to the slow release fertiliser, and/or optionally further comprising the step of processing the slow release fertiliser, for example by pelleting, granulating, press-forming or powderising the slow release fertiliser, and/or optionally further comprising the step of coating the fertiliser with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds.

1. AU201428296

National Bibliography

Description

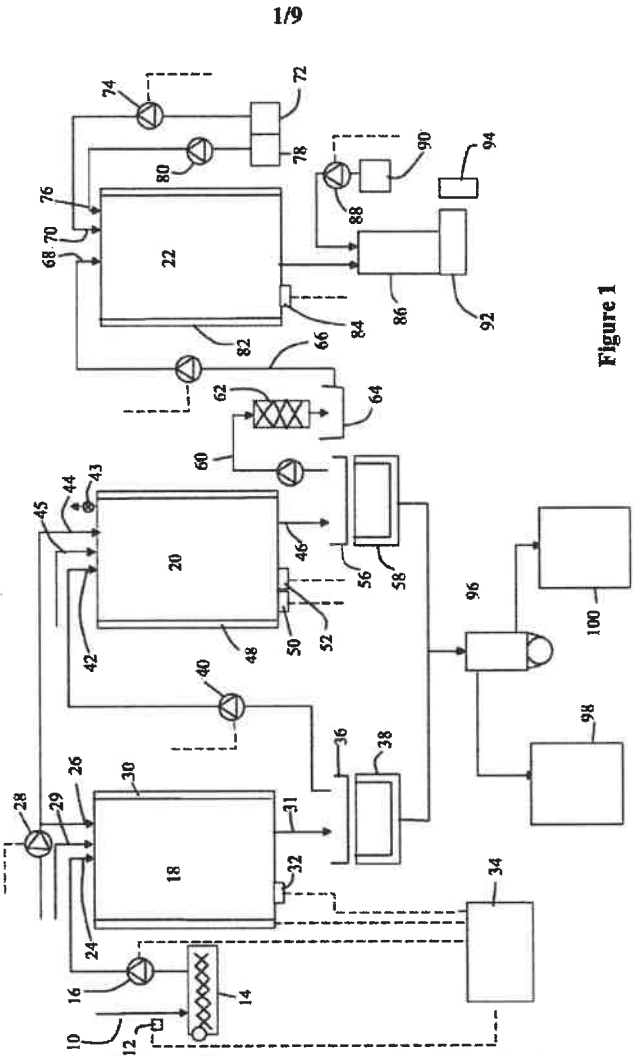


Image 1 of 9



Figure 13



Figure 15



Figure 16

1. AU2014282964

National Filing Date Description

Figure 2

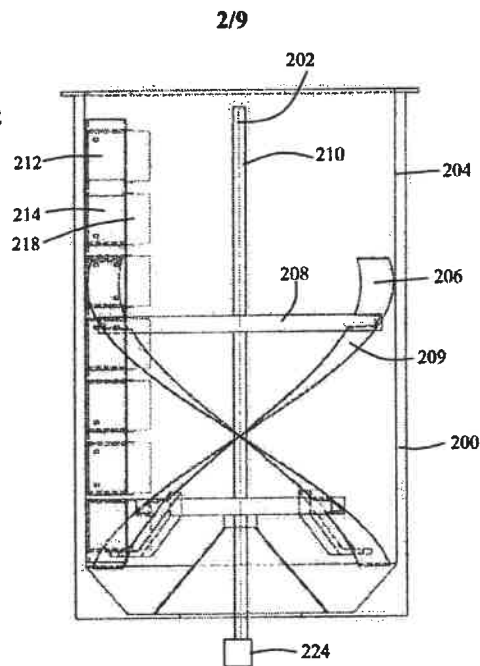


Figure 3

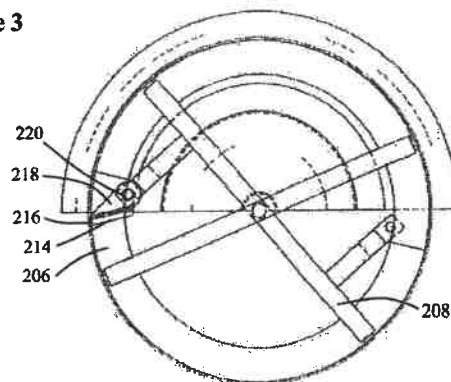


Image 2 of 9



Figure 11

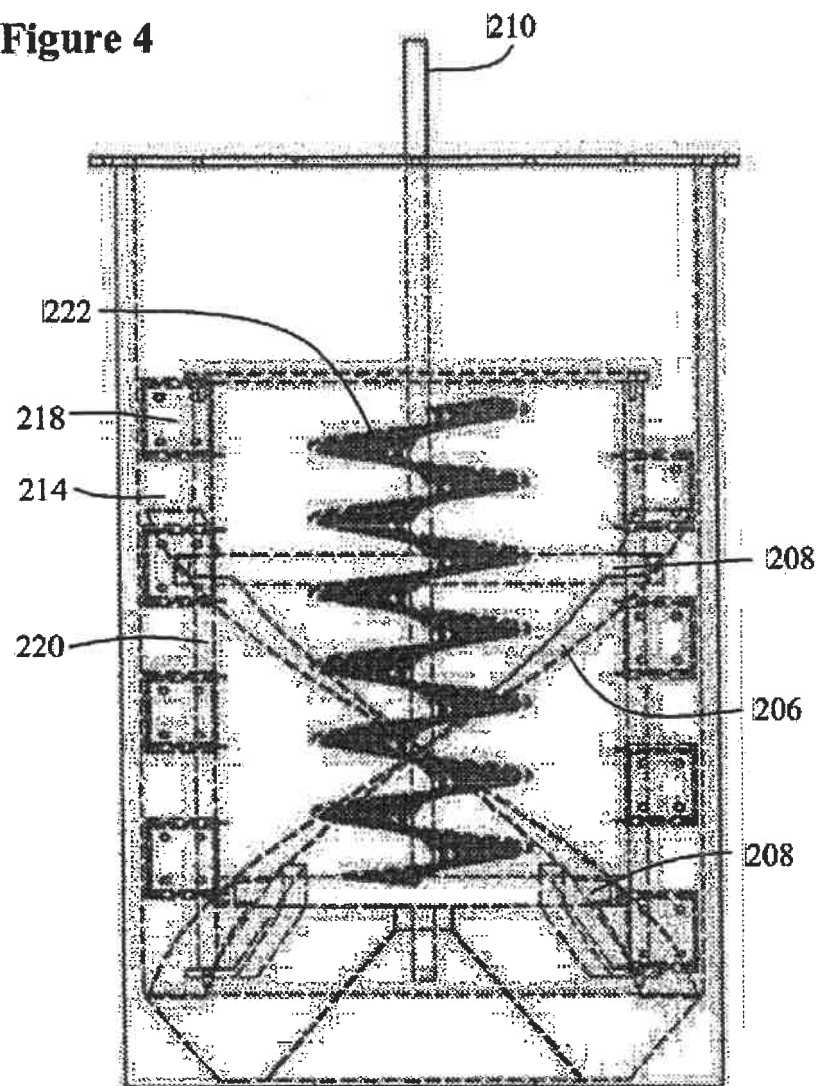
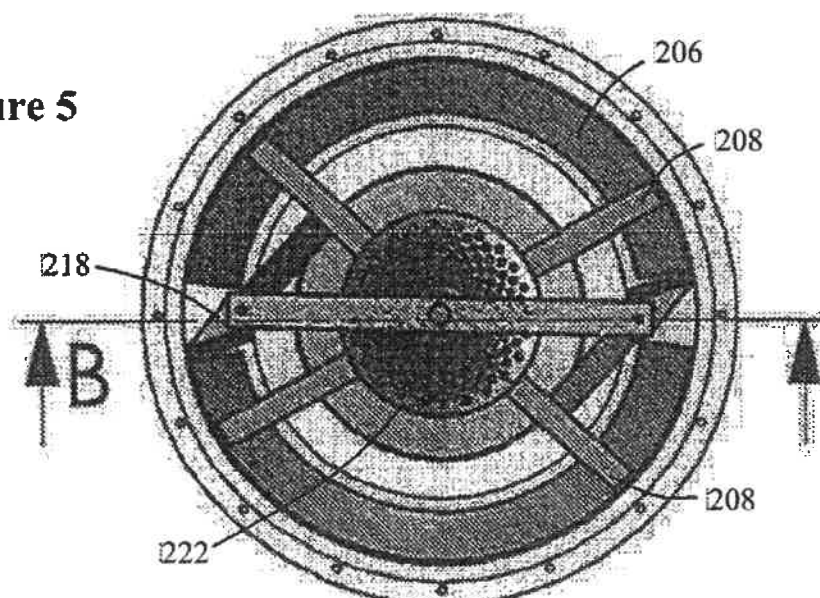


Figure 12



Figure 13

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Figure 4**Figure 5**

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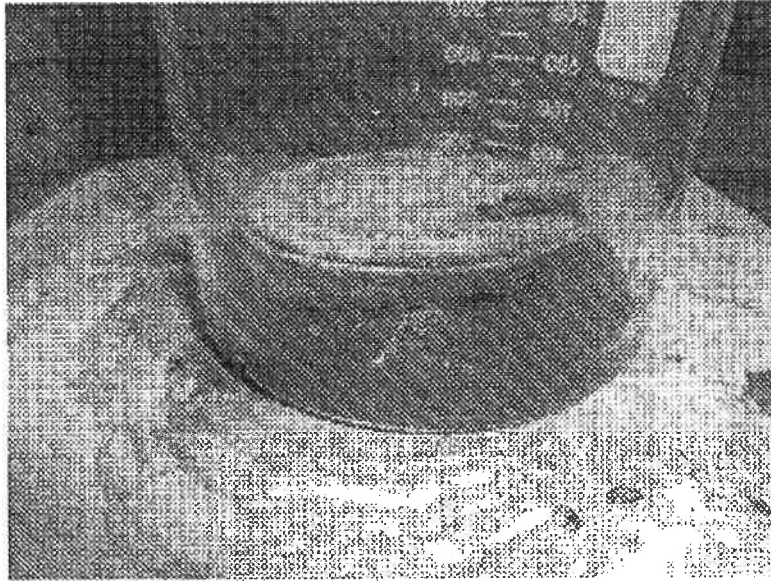


Figure 6

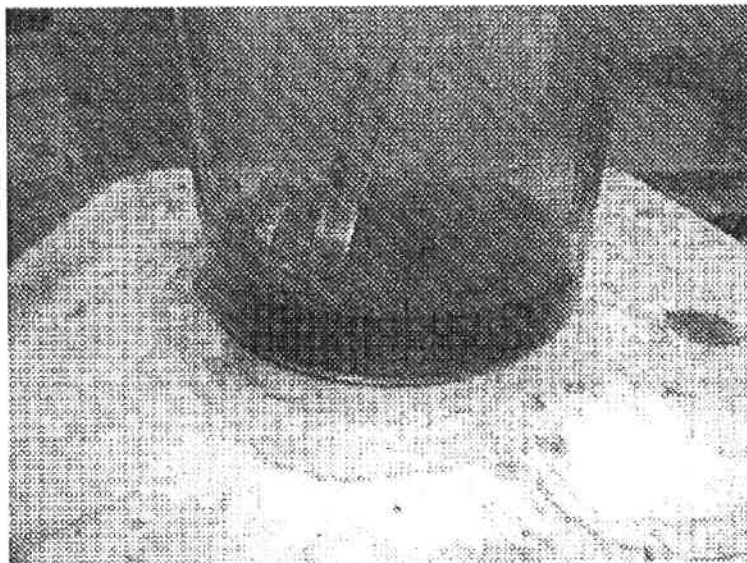
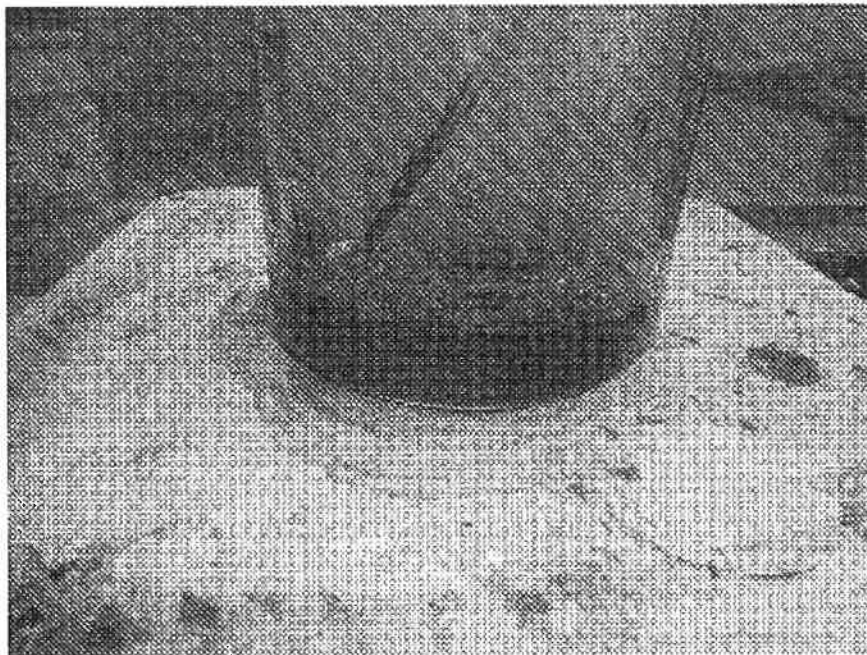
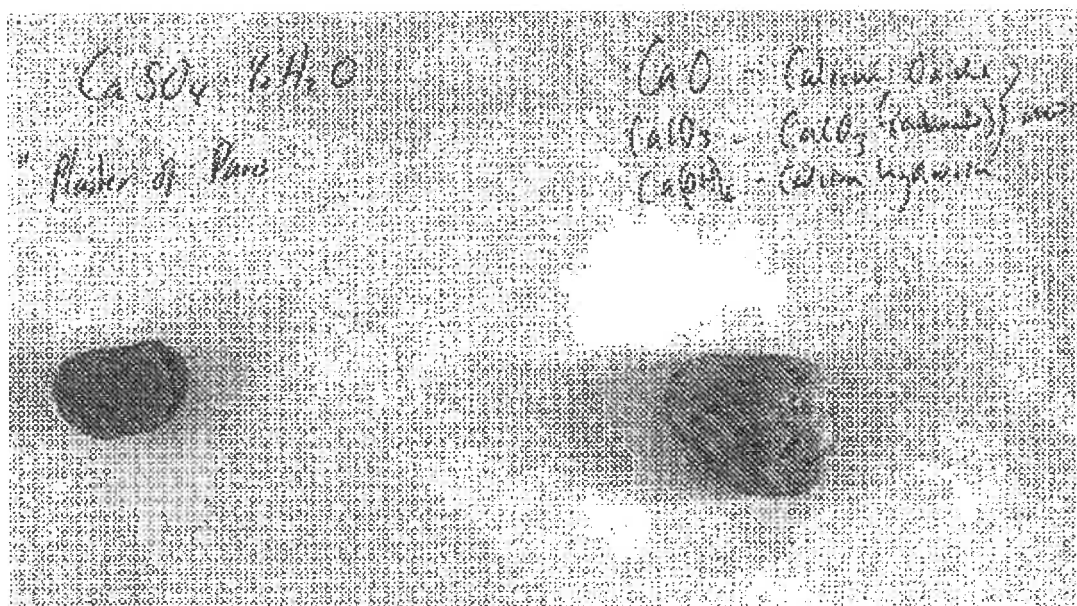


Figure 7

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**Figure 8****Figure 9**

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Figure 10



Figure 11

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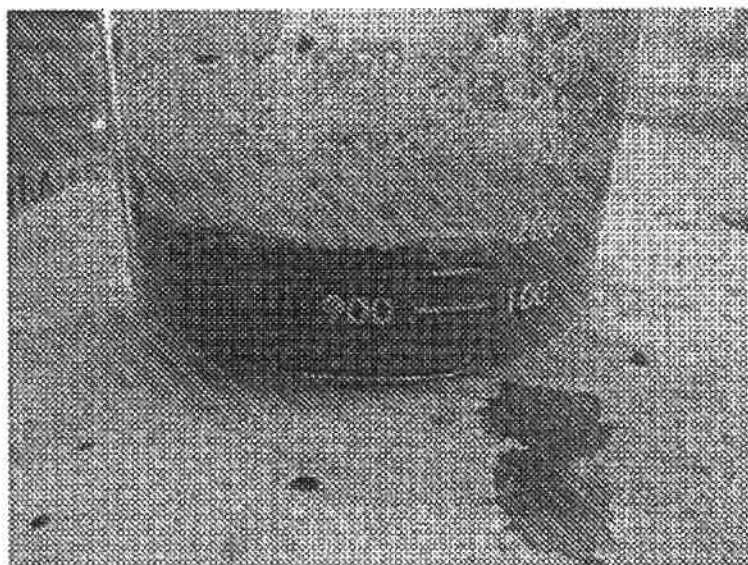


Figure 12

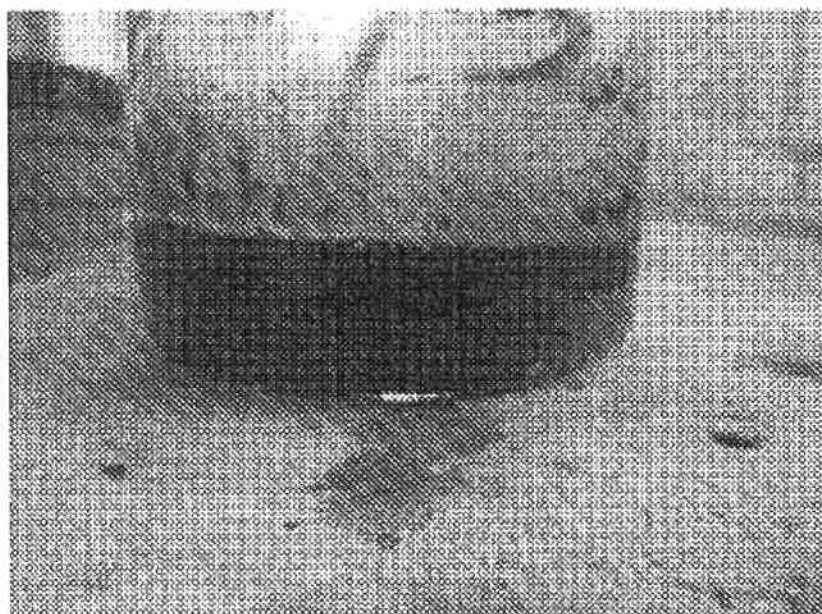


Figure 13

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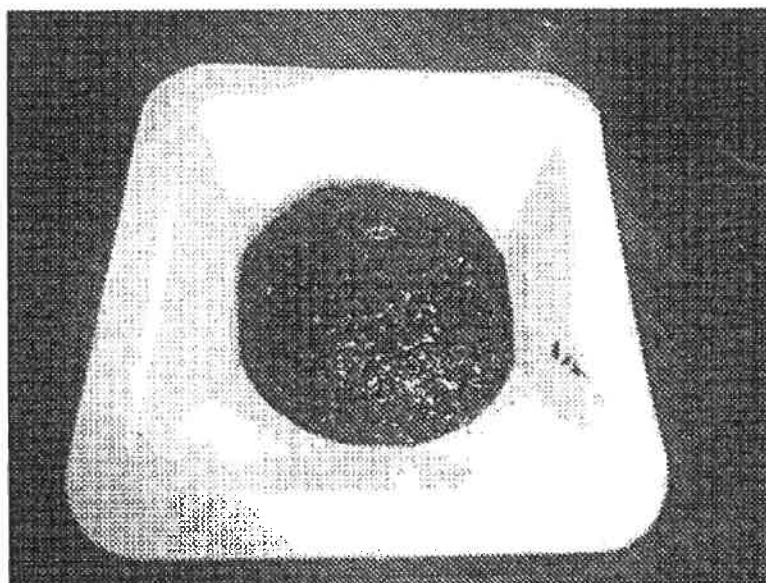


Figure 14

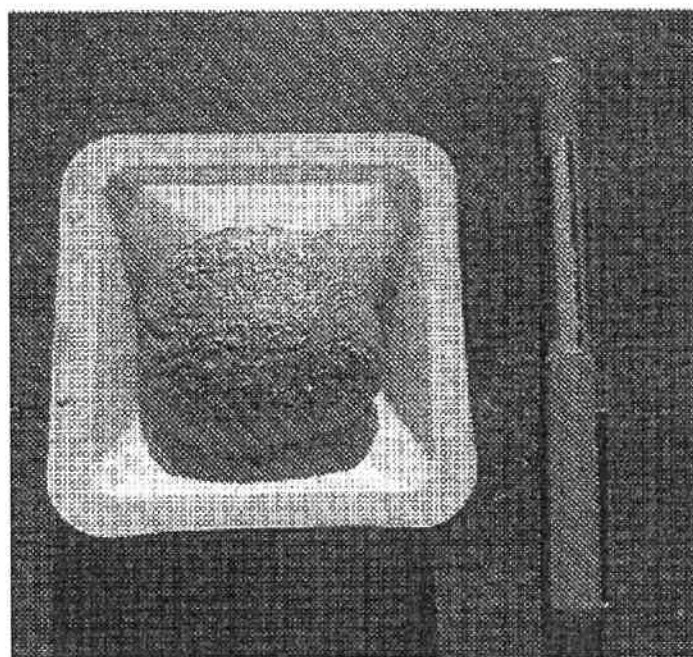


Figure 15

1. AU2014282964

National Bibliographic Data Description

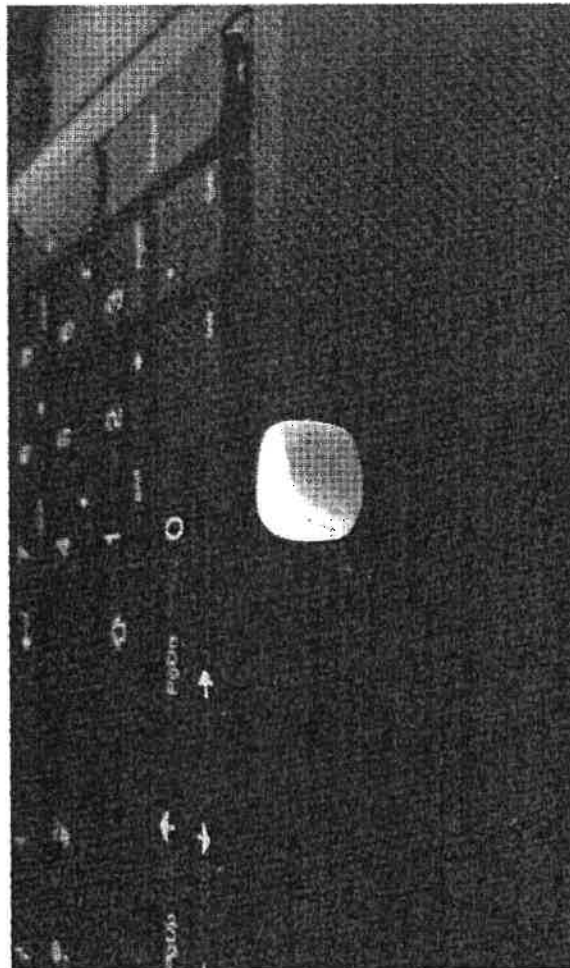


Figure 16

Image 9 of 9



Figure 14



Figure 12

Figure 13

2014282964

1. AU2014282964 - ORGANIC WASTE PROCESSING



National Biblio. Data Description Claims Drawings Patent Family Documents

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C05F 11/08

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[View more classifications](#)

Applicants

Elemental Digest Limited

Inventors

Palmer, Bruce William

Agents

Phillips Ormonde Fitzpatrick

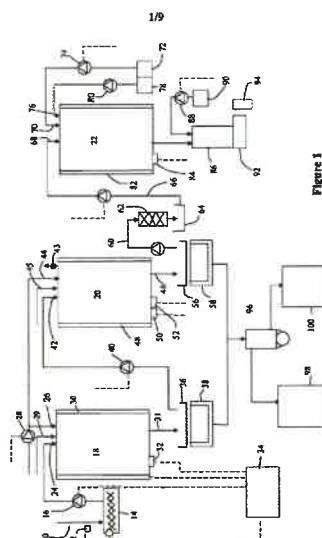
Priority Data

1311108.3 21.06.2013 GB

1322874.7 23.12.2013 GB

Title

(EN) Organic waste processing



Abstract

(EN) The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertiliser from organic waste, especially slow release fertiliser. The present invention also relates to fertilisers made by the method of the invention, especially slow release fertiliser. The method of processing organic waste to provide a slow-release fertiliser may comprise at least one hydrolysis step to provide hydrolysed organic waste and a setting step to provide a slow release fertiliser.

Related patent documents

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 EP3010871 CA2915964 MX366240 IN201627001450 NZ716125 C07590580

1. AU2014282964 - ORGANIC WASTE PROCESSING



National Biblio. Data

Description

Claims

Drawings

Patent Family

Documents

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Machine translation

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[EN]

ORGANIC WASTE PROCESSING

Field of the Invention

The present invention relates to methods of processing organic waste, in particular it relates to a method for making fertiliser from organic waste, especially slow release fertiliser. The present invention also relates to fertilisers made by the method of the invention, especially slow release fertiliser.

Background to the Invention

Organic waste may be animal by-products, vegetable waste, food waste or a mixture of both.

During the processing of animals for meat production or other uses, relatively large amounts of the animal are removed, discarded and not sold on to an end user. Such discarded by-products include animal by-products such as bone, blood, gut content, wool, fur and feathers for example. Animal by-products also include animal waste such as chicken-litter, cow slurry and pig or horse manure. Vegetable waste may include kitchen waste, bedding from animals, fruit or vegetable processing waste, for example, fruit peels, sugar cane waste. This represents a significant waste of the resources that could be harnessed for useful purposes. There is also a cost involved in storing and disposing of such by-products as waste.

Various processes are previously described for processing organic waste. Use of incineration and composting to process the waste into a usable form or extract energy from waste has been previously described.

Organic waste comprises valuable nutrients and there is a need for methods of processing organic waste to provide useful products. There is also a need for and environmentally friendly alternatives to conventional fertiliser for crops.

Slow release or controlled release fertilisers are particularly valuable because they release nutrients to plants over a period of time, for example several months. This means that there is less danger of toxicity to plants and of fertiliser running off the land before it is taken up by plants. Slow release fertilisers can provide a sustained supply of nutrients through a longer period of the growing season compared to fertilisers that are easily soluble and release nutrients quickly.

The discussion of documents, acts, materials, devices, articles and the like is included in this specification solely for the purpose of providing a context for the present invention. It is not suggested or represented that any or all of these matters formed part of the prior art base or were common general knowledge in the field relevant to the present invention as it existed before the priority date of each claim of this application.

Where the terms "comprise", "comprises", "comprised" or "comprising" are used in this specification (including the claims) they are to be interpreted as specifying the presence of the stated features, integers, steps or components, but not precluding the presence of one or more other features, integers, steps or components, or group thereof.

It is an aspect of the present invention to provide a method for disposing of organic waste and a method for processing organic waste to provide a slow release fertiliser.

The present invention provides a method of processing organic waste comprising the step of mixing the organic waste with a hydrolysing composition.

The method of processing organic waste may be a method of making a slow release fertiliser.

The present invention provides a method of processing organic waste to provide a slow-release fertiliser comprising at least one chemical disintegrating step to provide solubilised organic waste and a setting step to provide a slow release fertiliser, wherein the chemical disintegration step comprises combining the organic waste with a metal catalyst, concentrated sulfuric acid and concentrated hydrogen peroxide, wherein said catalyst is titanium dioxide, copper or a copper compound; and heating at between about 70 °C and about 200 °C and about 3 bar absolute pressure for at least 20 minutes, and wherein the setting step comprises combining the solubilised organic waste with a mixture comprising or consisting of one or more of lime, ammonia, calcium sulphate, calcium carbonate, calcium oxide and calcium hydroxide.

The method may be carried out in a suitable apparatus comprising one or more vessels. The one or more vessels may be equipped with a mechanism for lifting the mixture upwards in a central region of the vessel by means of a lifting mechanism, resisting downward movement of the mixture in an outer region of the vessel by means of a heat transfer member, and transferring heat within the mixture using the heat transfer member. The vessels may also be equipped with a heating mechanism to heat the contents of the vessel.

The heat transfer member may be rotatably mounted in the vessel and may have a leading end and a trailing end. The leading end may be below the trailing end

such that rotation of the heat transfer member generates a lifting force acting on the by-product, which may be arranged to resist its downward movement.

The heat transfer member may be formed from a strip of metal. The heat transfer member may be part helical.

The lifting mechanism may comprise a helical lifting member rotatable to lift the by-product. The lifting mechanism and the heat transfer member may be mounted on a common apparatus may comprise scraping means, for example one or more scrapers, arranged to scrape the inside of the vessel wall, or to scrape by-product, or other contents of the vessel, from the side wall. The scraping means may be mounted on the common rotatable support means for rotation with the heat transfer member.

The vessel wall may be cylindrical, for example having a central axis that is vertical. The vessel may further comprise heating means arranged to heat the by-product in the vessel.

The apparatus may further comprise a further processing vessel also comprising: a vessel wall; a lifting mechanism, which may be located in a central region of the vessel, arranged to lift the by-product within the vessel; a metal heat transfer member, which may extend around the lifting mechanism and may be in an outer region of the vessel, the heat transfer member optionally being rotatable around the outer region of the vessel to resist downward movement of the by-product in the outer region of the vessel and to dissipate heat through the by-product in the outer region of the vessel.

The vessel, or one of the vessels, may have an inlet connected to a supply of water. The vessel, or one of the vessels, may have an inlet connected to a supply of steam. The vessel, or one of the vessels, may have inlet means connected to a supply of acid and hydrogen peroxide.

The method may further comprise heating the mixture in the vessel.

The method may be carried out in a plant according to the invention.

The present invention may be carried out in an apparatus for processing organic waste the apparatus comprising a heating vessel having a product inlet arranged to receive the by-product, a water inlet arranged to receive water, and a steam inlet arranged to receive steam, and heating means arranged to heat the first heating vessel, a separator arranged to receive the contents of the first heating vessel and separate it into first and second components, and an acidification vessel having a product inlet arranged to receive the solid component, and acid inlet means arranged to receive acid. The acid inlet means may further be arranged to receive hydrogen peroxide.

Preferably the apparatus further comprises means to deliver a lime mix to the acidified by-product.

Preferably the apparatus further comprises means to deliver magnesium salts to the acidified by-product.

The apparatus may comprise a dryer operative to receive and dry the acidified by-product.

A by-product sizing apparatus may be provided operative to re-size the by product into pieces below a predetermined size.

The sizing apparatus may comprise a crusher operative to crush the raw organic waste into pieces below a predetermined size.

The sizing apparatus may comprise a plurality of crushers arranged in sequence and operative to sequentially crush the raw organic waste into smaller pieces.

Preferably the sizing apparatus is operative to output pieces of less than 40mm.

A metal detector may be provided at the inlet operative to detect any metal content in the raw organic waste and to generate a corresponding signal that is used to prevent raw organic waste entering the by-product sizing apparatus if metal content is detected.

Preferably the apparatus further comprises a pre-heating vessel operative to pre-heat the sized by-product and to deliver the sized by-product to the heating vessel.

The apparatus may further comprise a supply of at least one additive and means to deliver the additive to the further heated by-product to alter the properties of the further heated by-product.

The apparatus may comprise a pellet press or granulator operative to receive dried by-product and to press or granulate the dried by-product into pellets or granules.

The apparatus may further comprise a packaging unit or bagging machine operative to receive and package or bag the pellets or granules.

The method may comprise a step of sizing the organic waste, for example by crushing, grinding or chopping solid waste to provide suitable sized pieces. Suitable sized pieces may be less than 200mm, less than 150 mm, less than 100 mm, less than 50 mm or less than 20 mm sized pieces.

Water may be added to the organic waste before or after sizing the organic waste.

The method may comprise a heating step. The organic waste may be heated, for example to between 70 °C and 200°C. The organic waste may be heated for between 10 and 60 minutes. The organic waste may be heated for 20 minutes at 135 °C and 3 bar pressure to provide microbiological safety. The organic waste may be heated for a time and/or temperature as specified in legislation regarding disposal of organic waste or animal waste and/or microbiological safety. The organic waste may be heated by adding steam to the organic waste, for example steam may be added in a vessel as shown at [18] or [20] in Figure 1. The organic waste may be heated at higher than atmospheric pressure, for example in a pressurised vessel shown at [20] in Figure 1. The organic waste may be heated by heating the vessel in which the organic waste is held. The organic waste may be heated by an exothermic reaction when a hydrolysing composition is added to the organic waste, for example in a vessel as shown at [22] in Figure 1. The method may comprise one or more, two or more, three or more, four or more or five or more heating steps. Each heating step may be performed in a different vessel for example a heating step may be performed in each of vessels [18], [19] and [20] shown in Figure 1. The method may comprise two or more heating steps performed at different temperatures or pressures. Heating steps may be used to remove gelatine and fat; degrade the organic waste and/or provide a microbiologically safe product from organic waste.

The heating step may be an independent step in the method of processing organic waste performed at a different time from other steps. The heating step may be performed at the same time as another step in the method, for example the hydrolysis step. Combining the organic waste with a hydrolysing composition may cause an exothermic reaction that heats the organic waste. During the heating step the organic waste may be heated using a combination of heating methods.

During the acid-peroxide/high temp/high pressure and/or digestion step a metal catalyst may be added. The metal catalyst may be any suitable metal or metal compound that is able to increase the rate of reaction with respect to breaking apart protein material such as the hydroxyproline in collagen into the constituent amino acid building blocks. The subsequently released nitrogen atoms from the peptide links are released in the form of ammonium ions. In particular these catalysts could be very useful in the dissolving of bone in acid/peroxide digestion step. The catalysts may include copper or copper compounds specifically copper II sulphate or copper II oxide, selenium or selenium compounds, titanium and/or titanium compounds but specifically titanium dioxide as either a powder or a nano-powder. Some of the catalysts, such as selenium and copper, would add benefit to the micronutrient composition of the final fertiliser product. The catalyst may be added in very small quantities, for example the mass of catalyst may be in the range of 0.1% to 0.01% of the mass of undigested organic waste, for example bone paste.

The method may comprise a step of draining liquid from the organic waste. The step of draining liquid from the organic waste may be performed after the heating step. Liquid may be removed from the mixture. The liquid may comprise fat and/or gelatine which may be separated and processed separately.

The method may comprise one or more hydrolysis steps where a hydrolysing composition is combined with the organic waste. The hydrolysing composition may be for example:

a combination of an acid and hydrogen peroxide. These may be added to the organic waste together or sequentially. For example Sulphuric acid H₂SO₄ 50-100% (for example 95%) followed by or mixed with Hydrogen Peroxide H₂O₂ 5-100% (for example 30% v/v) in a ratio of 1:1 to 1:10 sulphuric acid solution : hydrogen peroxide or for example, nitric acid 37% nitric acid [70% or 1.42 specific gravity] followed by or mixed with 7% hydrogen peroxide [35% w/w];

a combination of a base and hydrogen peroxide. These may be added to the organic waste together or sequentially. For example ammonium hydroxide NH_4OH and Hydrogen Peroxide H_2O_2 30%;

a base, for example Potassium hydroxide KOH or sodium hydroxide NaOH ; or

an acid, for example Phosphoric acid H_3PO_4 ; or Hydrochloric acid HCl , Nitric Acid HNO_3 , Sulphuric acid H_2SO_4 .

The method may comprise one hydrolysis step, two hydrolysis steps or three hydrolysis steps. The hydrolysis steps may be performed sequentially. For example, the method may comprise an acid hydrolysis step and a base hydrolysis step performed sequentially. The method may comprise an acid or a base hydrolysis step and a hydrolysis step using acid or base and hydrogen peroxide.

The hydrolysis step may be an exothermic reaction and may cause the organic waste mixture to heat up. Heat may be removed from the organic waste mixture during this step to keep the mixture within suitable temperature limits.

If the hydrolysis step is exothermic it may contribute heat to the heating step.

The one or more hydrolysis steps may be performed in a vessel, for example as shown at [22] in Figure 1.

After the one or more hydrolysis steps the hydrolysed organic waste may provide an acidified hydrolysed organic waste.

The method may comprise a step of draining liquid from the organic waste. The step of draining liquid from the organic waste may be performed after the hydrolysing step. Liquid may be removed from the mixture. The liquid may comprise fat and/or gelatine and/or other components which may be separated from each other and processed separately.

The method may comprise an acidification step where the hydrolysed organic waste is combined with additional acid to bring the pH of the organic waste to an acid pH below pH 7. An acid, for example phosphoric acid, nitric acid, or sulphuric acid may be added if the organic waste mixture is not acidic after the hydrolysis step or if the organic waste mixture is not acidic enough after the hydrolysis step. The hydrolysed organic waste may be acidified to a pH between pH -2 and pH 7, between pH 1 and pH 7, between pH 2 and pH 6, between pH 3 and pH 6 or between pH 4 and pH 5. The amount of acid or base added may be enough to make the organic waste mixture slightly acid, for example pH 6.5 to pH 4.

The one or more acidification or neutralisation steps may be performed in a vessel, for example as shown at [22] in Figure 1.

The method may further comprise a step of adding nutrients or minerals, for example magnesium salts to the organic waste mixture. This is advantageous because additional nutrients that are required in a fertiliser may be added to the processed organic waste so that it can be made into a fertiliser with a suitable balance of nutrients. Nutrients or minerals may be selected in order to complement the known components of the organic waste to provide nutrients or minerals that are known to be lacking in the organic waste. Nutrients or minerals may be selected to address a deficiency in the area where the processed organic waste is going to be used as a fertiliser.

The method may further comprise a step of adding ammonia and/or other basic or carbonate materials to increase the nitrogen or other minerals in the organic waste so that the final fertiliser has higher nitrogen or other minerals. For example ammonia and/or other minerals may be added to the organic waste at any point in the processing in order to increase the nitrogen content, the phosphate content or the potassium content of the final fertiliser product. Ammonia and/or other minerals may be added to the organic waste at any point in the processing in order to produce a final fertiliser product with a particular balance of nitrogen phosphate and potassium (NPK content) that is suitable for a specific soil type and/or crop type.

The method may comprise the step of drying the neutralised organic waste mixture. This may be done by a combination of heating and the addition of a dehydrating granulating mixture followed by further heating in the granule drying stage of the process. The acidified organic waste mixture may be dried until it comprises 10-12% moisture.

The method may comprise a step of setting the organic waste mixture by combining acidic, hydrolysed organic waste with a setting mixture or liming mixture. This may be any compound or composition that can form a lime mortar.

The method may comprise a re-sizing step to re-size the organic waste mixture into pieces below a predetermined size. The re-sizing step may, for example, comprise crushing, chopping or grinding the product. The product may be re sized to pieces of less than 40mm.

The method may further comprise adding at least one additive to alter the properties of the organic waste mixture. For example, an unpalatable ingredient may be added to deter animals from eating the fertiliser.

The method may comprise the step of pelleting or granulating the organic waste mixture after the setting step.

Granular fertilisers may be treated with hydrophobic anti-caking agents such as Kaolin by externally coating of the granules, or incorporated as part of the granulation mix; and or paraffin based additives which may be sprayed onto hot granules.

The method may further comprise a packaging step where the pellets or granules are packaged.

Organic waste may comprise, for example, waste from abattoirs, waste from meat processing or packaging, fallen stock, animal carcasses, food waste, animal excrement, for example, cow slurry, pig or cow dung, poultry litter, animal bedding, waste from the food industry, abattoir blood waste. Organic waste may be animal waste, vegetable waste or mixed animal and vegetable waste.

The method of processing organic waste may be a method of making fertiliser. The hydrolysing composition may be chosen in order to provide suitable ingredients to the fertiliser product, for example, use of nitric acid in the hydrolysing composition may provide additional nitrogen in the fertiliser product. Use of phosphoric acid in the hydrolysing composition may provide additional phosphorus in the fertiliser product. Use of sulphuric acid in the hydrolysing composition may provide additional sulphur in the fertiliser product.

The hydrolysing composition, neutralising composition and additives may be chosen to provide a suitable level of potassium, phosphorous and nitrogen in the final fertiliser when they are combined with the organic waste.

The hydrolysing composition may also be selected because of its suitability for hydrolysing the particular components of the organic waste that it is combined with. For example a hydrolysing composition comprising a base is particularly suitable for hydrolysing organic waste comprising cellulose. A hydrolysing composition comprising acid may be particularly suitable for hydrolysing organic waste comprising bone.

Ammonium hydroxide NH_4OH and Hydrogen Peroxide H_2O_2 30% may be particularly suitable for hydrolysing poultry litter, pig manure, cow manure, horse manure and/or contents of intestinal tract.

The method of the present invention may comprise a hydrolysis step using potassium hydroxide KOH and then a neutralising step using nitric acid to bring the pH close to neutral but keep the mixture slightly alkaline or slightly acidic as required for granulation into granules. The mixture may be set using lime, plaster of paris $\text{CaSO}_4 \cdot \frac{1}{2}\text{H}_2\text{O}$, Calcium carbonate CaCO_3 , Calcium oxide CaO or Calcium hydroxide Ca(OH)_2 .

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one hydrolysis step.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step and at least one hydrolysis step.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step, at least one hydrolysis step and at least one neutralising step.

The method according to the present invention may be a method of processing organic waste to provide a fertiliser comprising at least one heating step, at least one hydrolysis step, at least one neutralising step and at least one pelleting step.

The method of the present invention may be a method of processing organic waste to provide a slow-release fertiliser comprising at least one hydrolysis step to provide hydrolysed organic waste and one setting step to provide a slow release fertiliser.

The at least one hydrolysis step may comprise an acid hydrolysis step to provide an acidified hydrolysed organic waste.

The at least one hydrolysis step may comprise an acid hydrolysis step and a base hydrolysis step to produce an acidified hydrolysed organic waste.

The at least one hydrolysis step may comprise an acid hydrolysis step using any acid, for example sulphuric acid, nitric acid and/or phosphoric acid to hydrolyse organic waste and produce an acidified hydrolysed organic waste. The at least one hydrolysis step may comprise an acid hydrolysis step using sulphuric acid, to hydrolyse organic waste and produce an acidified hydrolysed organic waste.

The method may comprise an optional acidification step to make the pH of the acidified organic waste below pH 7.

The acidified hydrolysed organic waste may comprise sulphuric acid, phosphoric acid and nitric acid.

The method may comprise a step of setting the acidified hydrolysed organic waste mixture by combining acidic, hydrolysed organic waste with a setting mixture or a liming mixture. This may be any compound or composition that can form a lime mortar. The organic waste may be bonded to the lime mortar or embedded in it. The liming mixture may comprise, for example lime, calcium sulphate CaSO_4 , Calcium carbonate CaCO_3 , Calcium oxide CaO and/or Calcium hydroxide Ca(OH)_2 .

The liming mixture may further comprise clay bases materials such as kaolin.

The liming mixture may set to provide a lime mortar. During the setting step the organic waste and any additives that have been added to it may become bonded to the lime mortar or set within the lime mortar to provide a slow release fertiliser.

The lime mortar may degrade when left exposed to the elements or in the soil over a period of 1 to 6 months, or over a period of more than 9 months, or over a period of 3 to 9 months, or over a period of 3 to 6 months, or over a period of 3 to 5 months or over a period of 4 to 5 months. The lime mortar may degrade more quickly in soil that is more acidic.

The composition of the setting mixture or liming mixture may be adjusted to provide a slow release fertiliser that degrades faster or slower depending on what is required. In order to adjust the time that the fertiliser takes to degrade different proportions of each of the liming mixture ingredients may be added to the acidified organic waste. For example, the liming mixture can be adjusted so that more calcium sulphate is produced in the setting step to provide a fertiliser that breaks down more slowly. More calcium sulphate may be produced by reducing the amount of lime in the liming mixture so that the excess sulphuric acid reacts with the lime to make more calcium sulphate. The liming mixture can be adjusted so that it degrades faster by including more lime in the liming mixture.

The specific conditions where the slow release fertiliser will be used may be considered when selecting the composition of the organic waste and the composition of the liming mixture. The specific conditions may be the pH of the soil, the expected rain fall, the type of crop, the nutrient requirements of the crop, the nutrient profile of the soil, the ambient temperature that the crop will grow at, the length of growing time of the crop and/or the rate at which the crop requires nutrients during particular phases of its growth.

The percentage of liming mixture compared to acidified organic waste may be adjusted to provide a fertiliser that degrades faster or slower. The percentage of liming mixture used to set the acidified organic waste may be 10% to 50%, may be 20% to 50%, or may be 30% to 50%, or may be 40% to 50% of the mass of the organic waste.

The percentage of liming mixture added compared to the amount of acidified organic waste may be adjusted to provide a fertiliser with a higher or lower neutralising value. Neutralising value may be the amount of base compounds in the fertiliser. For example, a higher neutralising value may be required to partially neutralise acid soil or to provide lime to crops that prefer alkaline conditions. The percentage of liming mixture used to set the acidified organic waste may be 10% to 50% of the mass of the organic waste.

Lime mortar may comprise lime, calcium sulphate CaSO_4 , Calcium carbonate CaCO_3 , Calcium oxide CaO and/or Calcium hydroxide Ca(OH)_2 , which react with water and/or acid and set to form solid lime mortar. The liming mixture may further comprise clay bases materials such as kaolin to change the properties of the lime mortar. Other components of the hydrolysed organic waste may become part of the lime mortar by chemically bonding to components of the lime mortar or by being immobilised inside the lime mortar but not chemically bonded to it. As the lime mortar degrades in soil it may release lime into the soil and may also release components of the organic waste into the soil which provide nutrients and act as a fertiliser.

The setting step may comprise admixing the acidic mixture with lime and/or one or more calcium compounds wherein the calcium compounds are able to react with water and/or acid to form lime mortar which sets the mixture to form a slow release fertiliser. One or more calcium compounds within the liming mixture may react with sulphuric acid to form calcium sulphate which sets the mixture to form part of the lime mortar of the slow release fertiliser.

The lime mortar may set the mixture to form a solid slow release fertiliser with the components of the acidic mixture bonded inside the lime mortar matrix.

The setting step may comprise admixing the acidic hydrolysed organic waste mixture with one or more selected from calcium oxide, calcium carbonate and calcium hydroxide to form a slow release fertiliser.

The setting step may comprise admixing the acidic mixture with calcium oxide and calcium carbonate to form a slow release fertiliser.

The setting step may comprise admixing the acidic mixture with calcium oxide, and calcium hydroxide to form a slow release fertiliser.

The setting step may comprise admixing the acidic mixture with calcium carbonate and calcium hydroxide to form a slow release fertiliser.

The proportions of acidic mixture, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product is slightly alkali.

The proportions of acidic mixture, lime, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product takes more than a month, more than two months, more than 3 months, more than 4 months, more than 5 months or more than 6 months to degrade in soil. The soil may be between pH 5 and pH 7. The soil may be between pH 5 and pH 9.

The proportions of acidic mixture, calcium oxide, calcium hydroxide and/or calcium carbonate may be adjusted so that the slow release fertiliser product takes between one and 6 months, between 1 and 4 months or between 2 and 3 months to degrade in slightly acidic soil. For example the soil may be between pH 5 and pH 7.

The method may comprise the step of adding one or more minerals to provide suitable nutrients in the slow release fertiliser.

The method may comprise the step of adding one or more fungicides to provide antifungal properties to the slow release fertiliser.

The method may comprise the step of adding one or more pesticides to provide pesticide properties to the slow release fertiliser.

The method may comprise the step of adding one or more herbicides to provide herbicide properties to the slow release fertiliser.

The method may comprise the step of adding one or more unpalatable compounds deter animals from eating the slow release fertiliser.

The method may comprise the step of processing the slow release fertiliser, for example by pelleting, granulating, press-forming or powderising the slow release fertiliser.

The method may comprise the step of coating the fertiliser, for example coating pellets or granules of the fertiliser, with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds.

In a second aspect the present invention provides a slow release fertiliser provided by the method of the present invention.

In a further aspect the present invention provides a slow release fertiliser comprising a proportion of lime mortar to set the fertiliser.

The slow release fertiliser may comprise more than 10%, more than 20%, more than 30%, more than 40%, more than 50%, more than 60% or more than 70% lime mortar.

The slow release fertiliser may comprise between 10% and 70%, between 20% and 50% or between 30% and 40% lime mortar.

The slow release fertiliser may comprise lime mortar that takes between one and 6 months, between 1 and 4 months or between 2 and 3 months to degrade in slightly acidic soil. For example the soil may be between pH 5 and pH 7.

The slow release fertiliser may further comprise one or more fungicides, one or more pesticides, one or more herbicides, one or more pest control agents and/or one or more unpalatable compounds. Pellets of the fertiliser may be coated with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds.

In a further aspect the present invention provides a composition obtained or obtainable by the method of the invention.

In a further aspect the present invention provides a fertiliser composition obtained or obtainable by the method of the present invention.

There now follows by way of example only a detailed description of the present invention with reference to the accompanying drawings, in which;

Figure 1 is a schematic view of an apparatus for processing organic waste;

Figure 2 is a cut-away side view of a heating vessel forming part of the plant of Figure 1;

Figure 3 is a plan view of the vessel of Figure 2;

Figure 4 is a cut-away side view of a heating vessel of Figure 2 with further components shown; and

Figure 5 is a plan view of the vessel of Figure 4;

Figure 6 shows a beaker containing 95g of congealed frozen blood and 30 g of water;

Figure 7 shows a view of the hydrolysis and heating stage being performed on 95 g of blood with 30 g of water. 10g of potassium Hydroxide pellets have been added and the mixture was stirred for two minutes to form a deep red solution. This reaction is very exothermic and so hydrolysis and heating happen at the same time;

Figure 8 shows the mixture of figure 7 with 8g of 70% (1.42 specific gravity) nitric acid added to neutralise the potassium hydroxide this step is also very exothermic and is the neutralisation step and a heating step at the same time. The resulting mixture is an alkaline solution of potassium nitrate containing dissolved hydrolysed blood;

Figure 9 shows a view of the solid set product that results when the mixture shown in figure 8 is combined with $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ (left hand side) or with CaO , CaCO_3 and $\text{Ca}(\text{OH})_2$;

Figure 10 shows a view of 63 g of chicken litter and bedding;

Figure 11 shows a view of the 63 g of chicken litter and bedding mixed with 150 ml and blended to form a workable paste;

Figure 12 shows a view of the blended mixture shown in Figure 11 with 3 g (0.0306 mole) of H_3PO_4 [orthophosphoric acid (1.75 specific gravity)] added and the mixture brought to the boil. This is the heating stage and a first hydrolysis stage [acid hydrolysis] done at the same time;

Figure 13 shows a view of the mixture in Figure 12 with 0.0918 mol of potassium hydroxide (5.14 g) added to neutralise the acid, then 5 g more potassium hydroxide added to increase the pH to 14 and hydrolyse fat, proteins, lipids and nucleic acids, this stage is highly exothermic and is therefore a second hydrolysis step [base hydrolysis] performed at the same time as a heating step;

Figure 14 shows a view of the resulting mixture after pH of the mixture shown in Figure 13 is neutralised using using 8 g of 70% (1.42 specific gravity) nitric acid;

Figure 15 shows a view of the mixture shown in Figure 14 after addition of calcium carbonate, calcium hydroxide and calcium oxide to produce a solid pellet; and

Figure 16 shows a pellet of fertiliser made from chicken litter using a process of the present invention.

The method of the present invention may be carried out in an apparatus for processing organic waste, an example of which is set out below and in Figures 1 to 5.

Referring to Figure 1, an apparatus for processing organic waste comprises an inlet 10 which receives raw organic waste. A metal detector 12 is provided at the inlet 10 to detect any metal in the raw organic waste and to generate a control signal indicative of any metal content detected. The inlet 10 feeds into a particle sizing module, which in this case is a crushing module 14, arranged to reduce the raw product to pieces of a suitable size. In this embodiment the crushing module 14 comprises a first crusher arranged to crush the raw organic waste into pieces of less than a predetermined size, said size being 150mm in this example, and a second crusher arranged to crush the raw organic waste into smaller pieces, of less than 40mm in this example. A conveyor 16, which may be a screw conveyor, is arranged to transfer the crushed product from the outlet of the crushing module 14 to the first 18 of three heating vessels 18, 20, 22.

The first heating vessel 18 comprises an enclosed steel container having an inlet 24 at the top through which the crushed product can be introduced, a further inlet 26, also at the top, which is connected to a water supply via a pump 28 so that water can be added to the product in the vessel 18, a further inlet 29 connected to a supply of steam, and an outlet 31 at the bottom. Flow control valves are provided in the water and steam inlets 26, 29, and the outlet 31. A heater 30 is provided around the wall of the vessel, and a temperature gauge 32 is arranged to measure the temperature in the vessel 18. A controller 34 is arranged to control the operation of the whole plant, including all the conveyors, pumps and heaters, and the inlet and outlet valves, as will be described in more detail below.

A first separation vessel 36 is arranged to receive the product from the outlet 31 from the first heating vessel, and is arranged to hold the solid product and allow liquid to drain into a collection tank 38 situated below it. A second conveyor 40 is arranged to transport the solid product from the separation vessel 36 to the inlet 42 of the second heating vessel 20.

The second heating vessel 20 is pressurized, and has a pressure vent controlled by a vent valve 43 so that the pressure in the vessel 20 can be controlled. It also has a further inlet 44 for water, a further inlet 45 for steam, and an outlet 46 at its bottom end, each with a flow control valve controlled by the controller

34. It also has a heater 48 around its side walls, and a temperature gauge 50 and pressure gauge 52 arranged to measure the temperature and pressure of the contents of the second vessel 20. These are connected to the controller 34 which is arranged to control the temperature and pressure, as well as the quantities of water, steam, and product, in the second vessel 20 as required.

A second separation vessel 56 is arranged to receive the product from the outlet 46 from the second heating vessel 20, and is arranged to hold the solid product and allow liquid to drain into a collection tank 58 situated below it. A third conveyor 60 is arranged to transport the solid product from the separation vessel 56 to a second crusher unit 62. The second crusher unit 62 is arranged to break down the solid product into pieces no larger than 2mm.

The outlet from the crusher unit 62 is arranged to output the product into a pan 64, which may be arranged to pre-heat the product prior to it being received in the third heating vessel 22. A further conveyor 66 is arranged to transport the product from the pan 64 to the inlet 68 of the third heating vessel 22.

The third vessel 68 also has a further inlet 70 at its top end connected to a source 72 of acid via a pump 74, and a further inlet 76 connected to a source 78 of hydrogen peroxide, via pump 80. These are controlled by the controller 34 so that acid and hydrogen peroxide can be added into the vessel 22 in the required amounts and rates and at the required time. The third vessel also has a heater 82 and a temperature gauge 84 connected to the controller 34. An outlet 84 at the bottom end of the vessel 22 is connected to a neutralization vessel 86 which is arranged to receive the product from the third heating vessel 22, and also arranged to receive an alkaline additive, in this embodiment a lime mix, via a conveyor 88, from an alkaline additive source 90, under the control of the controller 34.

The final neutralized product is in the form of a paste and an extruder 92 is arranged to receive this product from the neutralization vessel 86 and extrude it into pellets, and a packaging module 94 is arranged to receive the extruded pellets and package them for transport away from the plant.

Each of the liquid collection vessels 38, 58 has an outlet that is connected to a separator in the form of a centrifuge 96 which is arranged to separate out the lighter and heavier components of the liquid. The lighter components will generally comprise fats, and the heavier components generally comprise gelatine, and these components are collected in separate collection vessels 98, 100.

Operation of the plant will now be described with particular reference to the treatment of bone, although it will be appreciated that it can be used for a wide variety of other waste products.

Raw bone is introduced into the crusher module 14 where it is crushed into pieces no larger than 40mm. From there it is transferred into the first vessel 18.

The controller 34 is arranged to monitor the amount of product entering the vessel 18, for example using a load cell to measure its weight, and to stop the transfer of product when a specified amount has entered the vessel. The controller 34 is also arranged to control the introduction of water and steam into the vessel 18, and to heat the vessel 18 to the required temperature. The bone and water mixture is heated to approximately 95°C to form an organic waste slurry which is then transferred to the separation vessel 36.

In the separation vessel 36, liquid drains from the slurry into the collection tank 38 while the slurry is transferred along the vessel. The drier slurry is then transferred via the conveyor 40 to the second heating vessel 20. The amount of slurry introduced into the second vessel 20 is controlled by the controller, for example using weighing cells to weigh the amount added, and then controlled amounts of water and steam are added by controlled operation of the inlet valves. The second vessel 20 is then heated up to about 135°C, and the vent valve 43 closed so that the pressure increases to about 5bar. This temperature and pressure is maintained for about twenty minutes.

At the end of the treatment in the second vessel 20, the vent valve 43 is opened to reduce the pressure in the vessel, and the outlet 46 is opened to allow the product, which is still in the form of a slurry, to empty into the second separation vessel 56. From that vessel 56, liquid drains into the second collection vessel 58, and the solid component of the slurry is transferred via conveyor 60 into the second crusher unit 62, where it is crushed to a particle size of about 2mm and transferred to the pre-heating vessel 64, where it is pre heated.

From the pre-heating vessel 64 the solid product is transferred via conveyor 66 into the third heating vessel 22 via the conveyor 66 and the inlet 68. Again the amount of product introduced is controlled, and also sulphuric acid and hydrogen peroxide are added in controlled amounts via the inlets 70, 76. The reaction in the vessel 22 is exothermic and the temperature rises to about 95°C, and then falls off when the reaction is complete. This heat and the acid and hydrogen peroxide kills any disease or bacteria in the product, and also oxidises any carbon in the product resulting in an inorganic product which is suitable for use as a fertiliser.

When the acidification step is complete, the outlet 22 from the third vessel 22 is opened and the product transferred to the neutralization vessel 86, to which lime mix is added in controlled quantities in order to increase the pH of the mixture to neutralize the product. The lime mix may comprise a mixture of chalk and lime, or quick lime for example. The lime mix may also comprise additives to alter the properties of the mix, such additives for example comprising selected macro/micro nutrients. The lime mix, and/or the additives, can be varied as required to deliver the required fertilizer properties. The lime mix causes an exothermic reaction. This serves to at least partially dry the acidified organic waste.

The increased pH mixture, which is typically in the form of a paste, is then dried if necessary, and passed into the extruder 92, from which it is extruded as pellets or granules, and packaged at packaging module 94 for transport.

The liquid from the collection vessels 38, 58 is separated in the centrifuge 96, with fat being transferred to fat storage tank 98 and gelatine to the gelatine storage tank 100.

The third heating vessel 22 may also comprise additive inlets arranged to deliver further additives to the slurry. These additives can be of any suitable ingredient to adjust the mineral or nutritional content of the solid slurry particles to suit a particular use or location of use. Thus the additives can be controlled to alter the macro/micro nutrient bases of the solid slurry particles to suit different soil or ground conditions for example, or to provide optimum nutrition for a particular type of crop.

Referring to Figures 2 and 3, the first heating vessel 18 comprises a cylindrical steel body 200, about 1.25m high and 750mm in diameter, arranged so that its central axis 202 and curved side walls 204 extend vertically. A pair of part helical heat transfer blades 206 are mounted by means of cross members 208 on a central rotatable shaft 210 which is located on the central axis 202. Each of the heat transfer blades 206 is formed of a strip of sheet metal 209 about 15 mm wide, which is formed into a part-helix shape extending around the outer part of the vessel close to the side wall 204 but spaced slightly from it. The blades each extend from a point about two thirds of the way up the side of the vessel, down to close to the bottom of the vessel, and through half a turn around the central axis. The two blades 206 are diametrically opposite each other, so that they extend around opposite sides of the vessel.

A pair of scrapers 212 are also supported on the cross members 208. Each scraper 212 comprises a flat scraper blade 214 extending vertically down the side wall 204 of the vessel with an outer scraping edge 216 just clear of the side wall 204. Each of the scraper blades 214 is supported on a set of support brackets 218, which in turn are supported on a vertical support pole 220 which extends vertically between the cross members 208 near their outer ends.

Referring to Figures 4 and 5, an inner helical lifting flight 222 is also mounted on the shaft 210. The lifting flight 222 is formed of a flat strip of metal 224 about 100mm wide, formed into a helix centred on the shaft 210, and turning through about seven turns. The top of the lifting flight 222 is about level with the top of the heat transfer blades 206 and the bottom of the lifting flight 222 is slightly above the bottom of the heat transfer blades 206. The turning sense of the lifting flight 222 is the same as both of the heat transfer blades 206. A motor 224 is arranged to rotate the shaft 210, so that the lower end of the lifting flight forms a leading end and the upper end of the lifting flight forms a trailing end, and the lifting flight provides lift in the central part of the vessel 18 close to the central axis 202. At the same time, the lower ends of the heat transfer blades 206 form leading ends and the upper ends of the heat transfer blades form trailing ends, and the heat transfer blades also provide a small amount of lift, or in fact resistance to downward flow, in the outer part of the vessel closer to the outer wall 204. As the shaft rotates, the scrapers 212 move around the wall 204 of the vessel, scraping from the wall any of the by-product or other contents of the vessel that may have accumulated on the wall.

The central shaft 210 is rotated as the product is introduced into the vessel 18 through the inlet 24, the water is introduced through the inlet 26, this forms a slurry, and the steam is introduced through the inlet 29 which helps to heat the slurry. As the heater 30 heats up the contents of the vessel 18 and the steam is injected into it, the central lifting flight 222 causes a general upward flow of the mixture in the central area of the vessel. The mixture then moves outwards

at the top of the vessel 18 and cascades downwards in the outer region of the vessel 18 close to the wall 204. The rotating heat exchange blades 206 provide some resistance to the downward flow, and therefore mix the mixture, but also help to conduct heat through the mixture quickly allowing it to be heated quickly to the required temperature. This helps to reduce the time required for the heating step carried out in the first heating vessel 18.

The second vessel 20 has the same mixing and heat exchange mechanism as the first vessel 18 as shown in Figures 2 to 5. In this vessel the heating process is similar to the first vessel, though at increased pressure, and the heat exchange coils and the lifting flight serve to distribute heat from the heater and steam throughout the vessel in a similar way. Similarly the third vessel 22 also has the same mixing and heat exchange mechanisms as shown in Figures 2 to 5. In this case the heat is generated by the exothermic reaction in the vessel. However, rapid heat dissipation is still helpful to ensure that the temperature remains approximately equal throughout the vessel, and that the reaction proceeds at about the same rate throughout the vessel.

It will be appreciated that various modifications can be made to the embodiment described above, and that the design of the heating vessel or vessels can be varied as appropriate for the process and products to be treated. For example, the two heat transfer blades could be replaced by one continuous helical coil, or more than two blades. Also the blades may not be strictly part helical. For example the angle of the blade or blades to the vertical may vary along its length. Similarly the lifting flight may be of a different shape, or may be rotated independently of the heat transfer coils. Indeed a different type of lifting mechanism may be provided which is not a helical flight.

A number of pilot studies, described in the examples below, were performed to assess the effectiveness of different compositions for use in the hydrolysis step of the present method. The hydrolysis step may take place in one of the reaction vessels, for example vessel [22] in Figure 1. The pilot studies were mostly done on a small scale but may be scaled up and the method may be performed on large quantities of organic waste, for example in an apparatus as shown in Figure 1. The inlets for acid and hydrogen peroxide described in relation to Figure 1 may alternatively be used to add other ingredients for the hydrolysis and neutralisation steps.

Example 1 Acid and Hydrogen Peroxide - Bone Paste

The preferred ranges of the temperature and reaction times.

Operating temperature range: Start temp of bone paste 10°C to 100°C giving an end temp after the hydrolysis step of 100°C to 200 °C.

Preferred temp for process: Start temperature 50-90 °C - end temp 130-170 °C.

Ratio of wet bone paste to acid to hydrogen peroxide by mass 5-20:5:1 (wet bone:acid:peroxide), for example 13.5:5:1 (wet bone : acid : peroxide).

For example the ratio may be: 1000kg bone paste (wet) to 370kg of 95% sulphuric acid to 74kg of 30% hydrogen peroxide.

Treatment time Broad range - 1min to 30mins, for example, the treatment time may be 5-10 mins.

The amounts of acid and hydrogen peroxide - Broad Range - 1:1 to 1:10 Concentrated sulphuric acid to hydrogen peroxide solution.

Preferred ratio is 1:5 concentrated sulphuric acid to hydrogen peroxide for bone digestion but will vary for other waste streams.

Concentrated sulphuric acid used is 95% (1.83g/ml) but this could work over a range of 50 - 100%.

Concentrated hydrogen peroxide used is 30%w/v [weight to volume] but concentrations between 5-100% w/v although greater than 30% can carries an explosion risk.

Example 2 Base and Hydrogen Peroxide

Base peroxide

Broad range of concentrated ammonium hydroxide, and/or potassium hydroxide and/or sodium with hydrogen peroxide from 1:1 to 1:10 using 0.880sg (specific gravity) ammonium hydroxide (ammonia solution) and 30% hydrogen peroxide solution.

Preferred mixture 3:1 of 0.880sg ammonia solution to 30% hydrogen peroxide solution for the purpose of dissolving for example poultry litter, pig manure, cow slurry, cow manure, horse manure, contents of intestinal tract material from abattoir waste etc.

Example 3 Hydrolysis of Blood

Overview - blood contents are hydrolysed using sodium hydroxide or potassium hydroxide. The resulting solution is neutralised using nitric acid or sulfuric acid forming a potassium nitrate, sodium nitrate, sodium sulphate or potassium sulphate solution depending on the acid and base used solution containing the dissolved, heat treated, hydrolysed blood components.

Method:

To 100g of congealed blood add 30g of water and bring to the boil (water not required if blood is fresh and uncongealed). NB - frozen blood used can be seen melting in Figure 1.

Hydrolysis: Add 10g of Potassium Hydroxide pellets KOH (solid, form preferred) or 7g of sodium hydroxide [solid form preferred] and stir for two minutes until deep red solution forms - very exothermic reaction. The product of this reaction is shown in Figure 2.

Neutralisation: Add 16g of 70% 1.42 specific gravity nitric acid to neutralise the potassium hydroxide [concentrated] although a dilute nitric acid solution would suffice - very exothermic. The resulting mixture is shown in Figure 3.

The resultant mixture is then treated with further nitric acid to achieve the desired pH for granulating - this could be acidic or alkaline.

Solution of potassium nitrate containing dissolved, hydrolysed blood.

If a small amount of solid precipitates from the blood it may be separated and hydrolysed/oxidised by a small, additional dose of nitric(or sulphuric) acid and hydrogen peroxide in a ratio of 5:1 before recombining with original solution for pasting or setting.

The neutralised hydrolysed blood may be blended with other soluble ingredients and the resultant solution sold as a liquid fertiliser or the neutralised hydrolysed blood may be granulated and sold as a solid granular form after setting using plaster of paris [$\text{CaSO}_4 \cdot 1/2\text{H}_2\text{O}$] Figure 4 left or Calcium carbonate/oxide/hydroxide $\text{CaCO}_3/\text{CaO}/\text{Ca}(\text{OH})_2$ Figure 4 right.

Example 4 Acid hydrolysis using phosphoric acid (other acids could be used nitric acid or sulphuric acid) followed by alkaline hydrolysis using potassium hydroxide or sodium hydroxide - resultant mixture neutralised using nitric acid or sulphuric acid.

Phosphoric acid increases the phosphorus content (P) and hydrolyses cellulose content. Nitric acid increases the N content.

Potassium hydroxide increases the K content and hydrolyses bulk content. Nitric acid neutralises the potassium hydroxide or sodium hydroxide forming potassium nitrate or sodium nitrate and increases the N content.

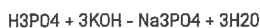
Approximate ratios by mass - 42[chicken litter] : 100[water] : 2[phosphoric acid] : 6.6[potassium hydroxide] or 4.7 sodium hydroxide: 8.0 [nitric acid].

Overview 63g of chicken litter and bedding is shown in Figure 5.

Addition of 150ml of water followed by 'blending' to form a suitable, workable paste shown in Figure 6.

Acid hydrolysis: - high-heat stage of process. 3g (0.0306 mole) of H₃PO₄ (Orthophosphoric acid 1.75 specific gravity) or 2.75g 70% nitric acid is added to drop pH to pH 3-4 and mixture brought to the boil (high heat stage of process).

Alkaline hydrolysis - in heliotherm stage. First the phosphoric acid is neutralised by potassium hydroxide:



3.00g of phosphoric acid (0.0306 mol) requires 0.0918 mol of potassium hydroxide = 5.14g then an addition 5g of KOH is required to facilitate the increase in pH to 14 and hydrolyse fat/protein/lipids/ nucleic acids etc [total 10g of KOH] - lots of heat evolved NaOH could also be used (less required 7.14g).

Neutralisation - still in heliotherm stage.

Neutralisation using 17g of 70% (1.42 specific gravity) nitric acid.

Resultant mixture is an alkaline, sweet (molasses type smell) brown paste containing potassium nitrate, potassium phosphate and all the components of chicken litter suitable for plant uptake.

Addition of calcium carbonate/hydroxide/oxide/calcium sulphate heptahydrate mix to produce the solid pellet form is shown in Figure 10.

Sodium hydroxide is useful for the hydrolysis of blood or chicken litter followed by neutralisation using sulphuric acid. For example 100g of blood requires 5g of sodium hydroxide at around 90°C to hydrolyse, followed by neutralisation using concentrated sulphuric acid 3.6ml (6.63g) and then adjustment of pH to desired level.

The present invention may be further described in the following numbered paragraphs:

1. A method of processing organic waste comprising one or more step of combining the organic waste with a hydrolysing composition.

2. The method according to paragraph 1 further comprising a heating step.

3. The method according to paragraph 2 wherein the organic waste is heated to between 70 °C and 200°C.

4. The method according to paragraph 2 or paragraph 3 wherein the organic waste is heated for between 10 and 60 minutes.

method according to any one of the preceding paragraphs further comprising the step of removing liquid comprising fat and/or gelatine from the organic waste.

method according to any one of the preceding paragraphs wherein the hydrolysing composition comprises: an acid and hydrogen peroxide; a strong base and hydrogen peroxide; a base; an acid; or hydrogen peroxide.

method according to any one of the preceding paragraphs further comprising a sizing step wherein the organic waste is sized to pieces less than 50mm.

method according to any one of the preceding paragraphs further comprising a neutralisation step, wherein acid or base is added to the mixture to bring the pH near to neutral.

method according to any one of the preceding paragraphs further comprising a step of adding minerals or nutrients to the mixture.

The method according to any one of the preceding paragraphs further comprising drying the organic waste mixture.

The method according to any one of the preceding paragraphs further comprising adding a setting agent such as calcium carbonate, calcium hydroxide, calcium sulphate hemihydrate and/or calcium oxide to the organic waste in a quantity suitable to set the mixture.

The method according to any one of the preceding paragraphs further comprising the step of re-sizing the set organic waste into pieces of less than 40 mm.

13. The method according to any one of the preceding paragraphs further comprising adding one or more unpalatable ingredients to the organic waste mixture.

14. The method according to any one of the preceding paragraphs further comprising a step of pelleting or granulating the organic waste mixture.

15. The method according to any one of the preceding paragraphs wherein the method of processing organic waste is a method of producing fertiliser.

16. An organic waste composition obtained by the method according to any one of the preceding paragraphs.

17. A fertiliser composition obtained by the method of any one of the preceding paragraphs.

1. AU2014282964 - ORGANIC WASTE PROCESSING

[National Biblio. Data](#) [Description](#) [Claims](#) [Drawings](#) [Patent Family](#) [Documents](#)[PermaLink](#) [Machine translation](#)

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[EN]

THE CLAIMS DEFINING THE INVENTION ARE AS FOLLOWS:

1. A method of processing organic waste to provide a slow-release fertiliser comprising at least one chemical disintegrating step to provide solubilised organic waste and a setting step to provide a slow release fertiliser, wherein the chemical disintegration step comprises combining the organic waste with a metal catalyst, concentrated sulfuric acid and concentrated hydrogen peroxide, wherein said catalyst is titanium dioxide, copper or a copper compound; and heating at between about 70 °C and about 200 °C and about 3 bar absolute pressure for at least 20 minutes, and wherein the setting step comprises combining the solubilised organic waste with a mixture comprising or consisting of one or more of lime, ammonia, calcium sulphate, calcium carbonate, calcium oxide and calcium hydroxide.
2. The method according to claim 1 wherein sulfuric acid of a concentration of 50% 100% sulfuric acid is used.
3. The method according to claim 1 or claim 2 wherein hydrogen peroxide of a concentration 30 %-100% hydrogen peroxide is used.
4. The method according to any one of the preceding claims wherein sulfuric acid and hydrogen peroxide are used in a ratio of 1:1 to 1:10.
5. The method according to any one of the preceding claims, wherein the method comprises a step of crushing, grinding or chopping solid organic waste before the chemical disintegration step.
6. The method according to any one of the preceding claims wherein the method further comprises a drying step to remove water after the setting step.
7. The method according to any one of the preceding claims, wherein the pH of the solubilised organic waste is increased by mixing it with chalk or limestone (calcium carbonate) or dolomite calcium magnesium carbonate).
8. The method according to any one of the preceding claims, wherein the copper II compound is selected from copper II sulphate, copper II oxide or other soluble copper II compound.
9. The method according to any one of the preceding claims, wherein the copper II compound or titanium dioxide is added at 0.1% - 0.01% of the weight of organic waste.
10. The method according to any one of the preceding claims, wherein the organic waste: sulfuric acid: hydrogen peroxide ratio is 5-20:5:1.
11. The method according to any one of the preceding claims wherein the liming mixture further comprises clay base materials such as kaolin.
12. The method according to any one of the preceding claims wherein wherein the mixture comprising or consisting of one or more of lime, calcium sulphate, calcium carbonate, calcium oxide and calcium hydroxide is a liming mixture, and the liming mixture sets to provide a lime mortar with the acidified solubilised organic waste bonded to the lime mortar or set within the lime mortar to provide a slow release fertiliser.
13. The method according to any one of the preceding claims wherein wherein the mixture comprising or consisting of one or more of lime, calcium sulphate, calcium carbonate, calcium oxide and calcium hydroxide is a liming mixture, and the liming mixture is composed to provide a lime mortar that degrades, when left exposed to the elements or in the soil, over a period of 1 to 9 months.
14. The method according to any one of the preceding claims, wherein wherein the mixture comprising or consisting of one or more of lime, calcium sulphate, calcium carbonate, calcium oxide and calcium hydroxide is a liming mixture, and the mass of liming mixture compared to the mass of acidified decomposed organic waste is adjusted to provide a fertiliser that degrades faster or slower and/or to provide a fertiliser with a higher or lower neutralising value.
15. The method according to any one of the preceding claims, wherein wherein the mixture comprising or consisting of one or more of lime, calcium sulphate, calcium carbonate, calcium oxide and calcium hydroxide is a liming mixture, and the mass of liming mixture is 10% to 50%, of the mass of the acidified solubilised organic waste.
16. The method according to any one of the preceding claims further comprising the step of adding one or more minerals at any stage in the method to provide suitable nutrients in the slow release fertiliser.
17. The method according to any one of the preceding claims further comprising the step of adding one or more fungicides, pesticides, herbicides, biocide and/or unpalatable compounds to provide desirable properties to the slow release fertiliser.
18. The method according to any one of the preceding claims further comprising the step of processing the slow release fertiliser, for example by pelleting, granulating, press-forming or powderising the slow release fertiliser.
19. The method according to any one of the preceding claims further comprising the step of coating the fertiliser with a coating comprising one or more bacteria, fungal spores, fungicides, pesticides, herbicides, pest control agents, and/or one or more unpalatable compounds.

20. A slow release fertiliser provided by the method of any one of the preceding claims.
21. The slow release fertiliser according to claim 20 comprising between 10% and 70% lime mortar.
22. The slow release fertiliser according to claim 20 or claim 21 wherein the lime mortar takes between 1 and 9 months to degrade in soil.

1. AU201428296

national: ipsc date description

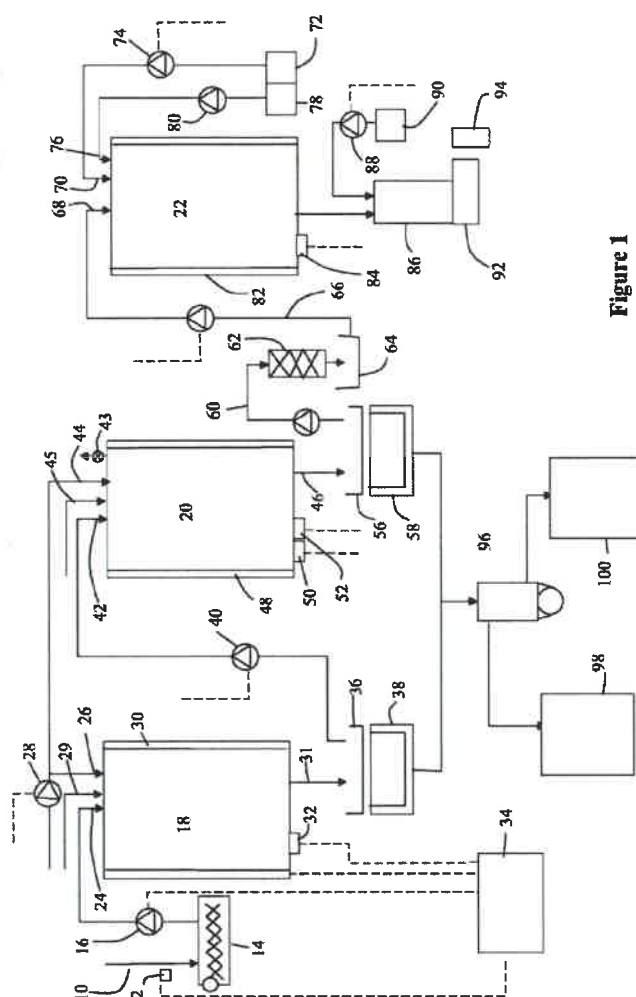


Figure 1

Image 1 of 9



Figure 13



Figure 15



Figure 16

1. AU2014282964

Figure. 2

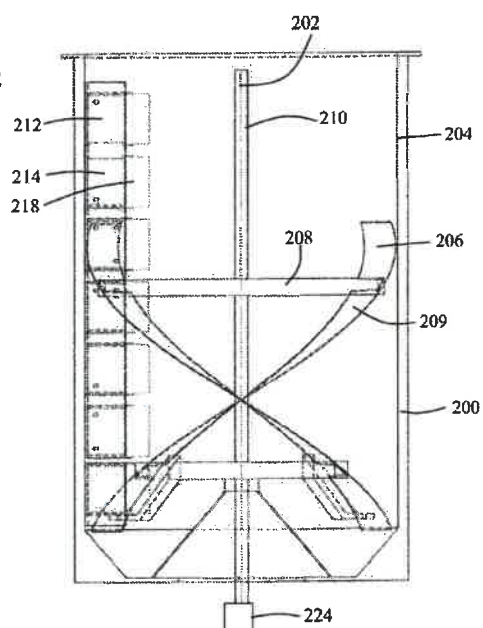


Figure 3

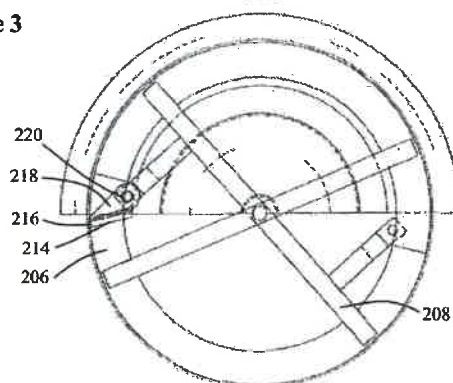


Image 2 of 9



Figure 13



Figure 14



Figure 15

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Figure 4

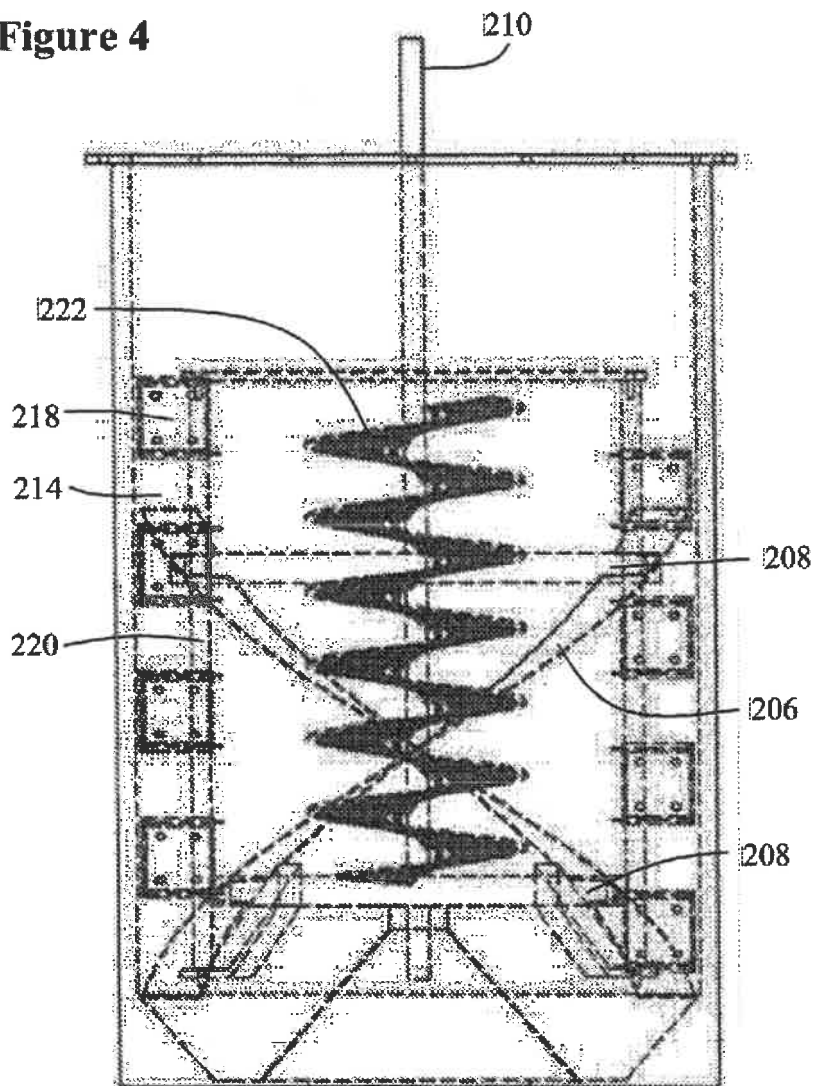
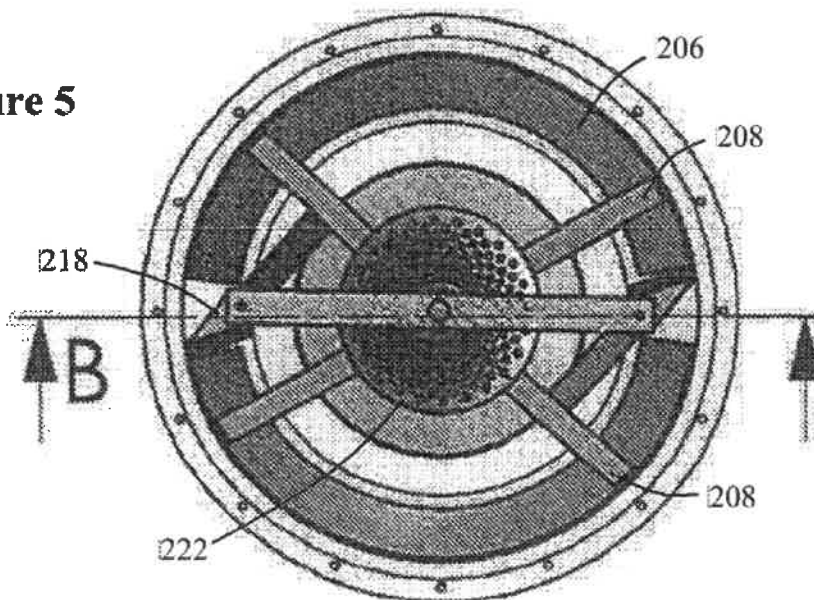


Figure 5



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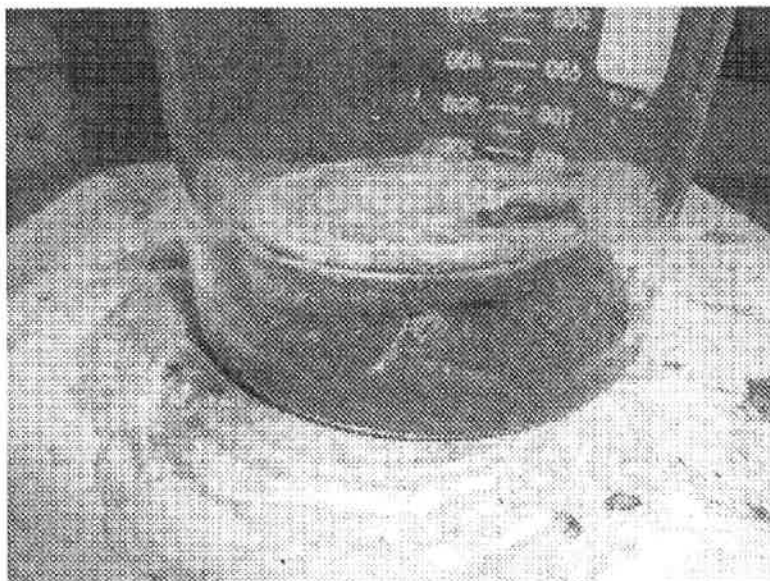


Figure 6

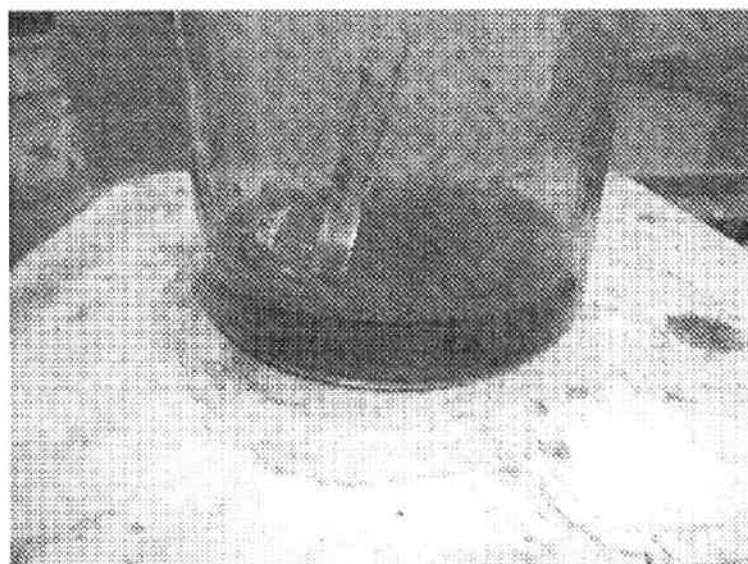
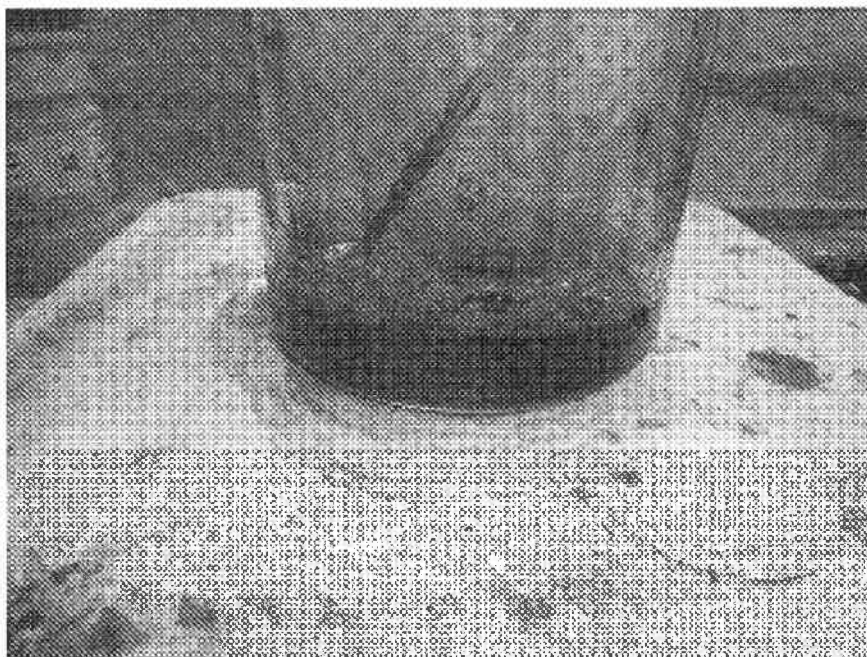
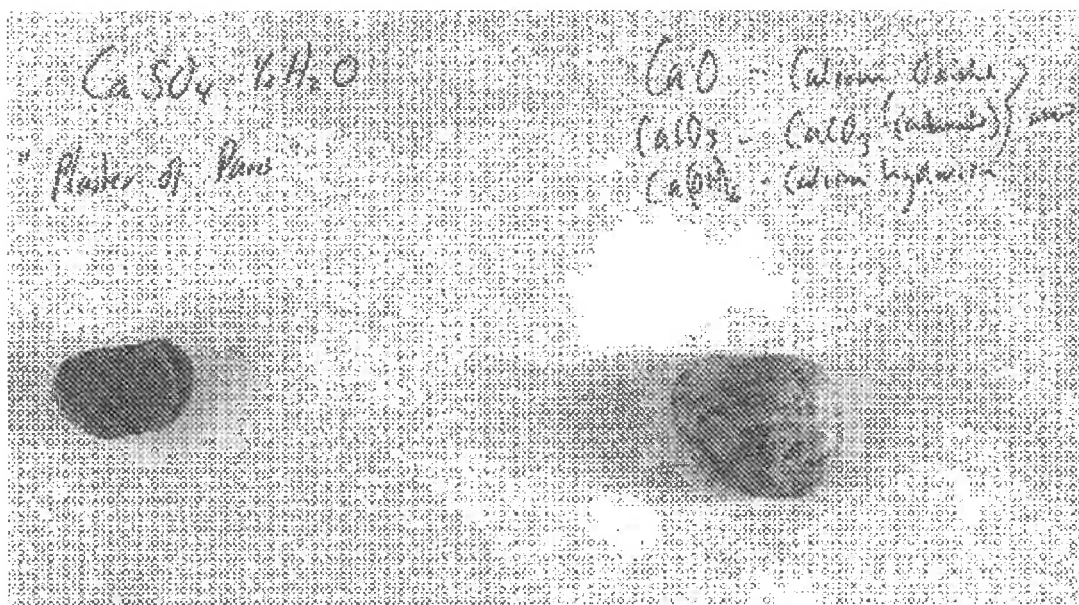


Figure 7

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**Figure 8****Figure 9**

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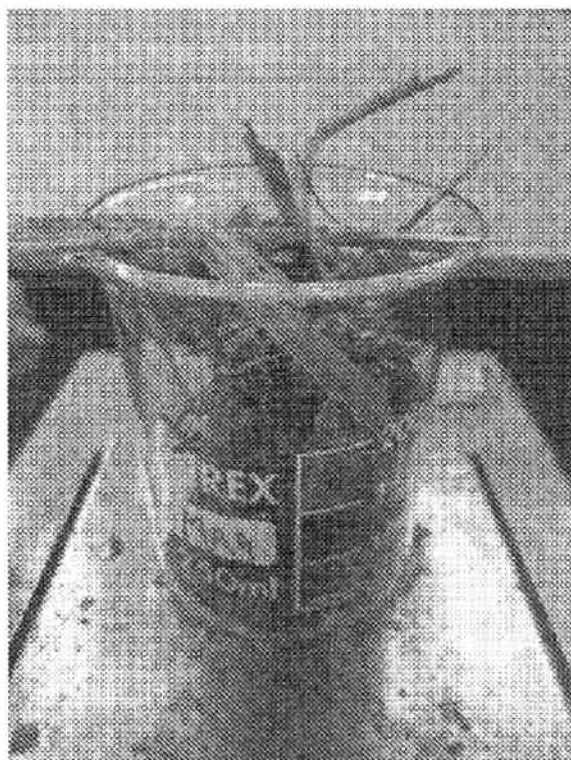


Figure 10

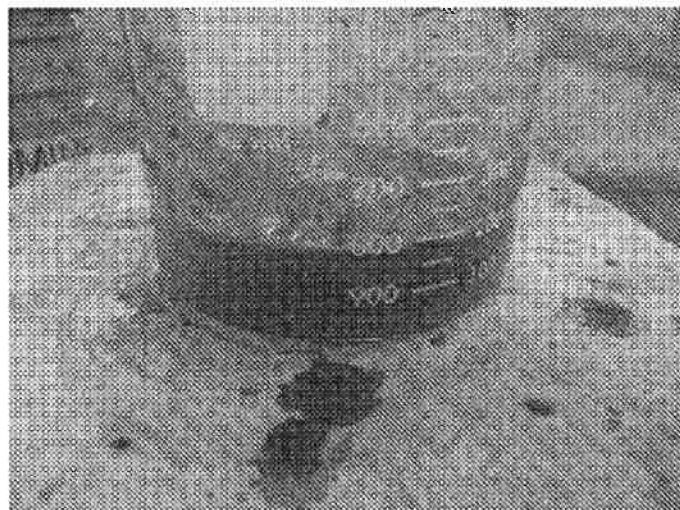


Figure 11

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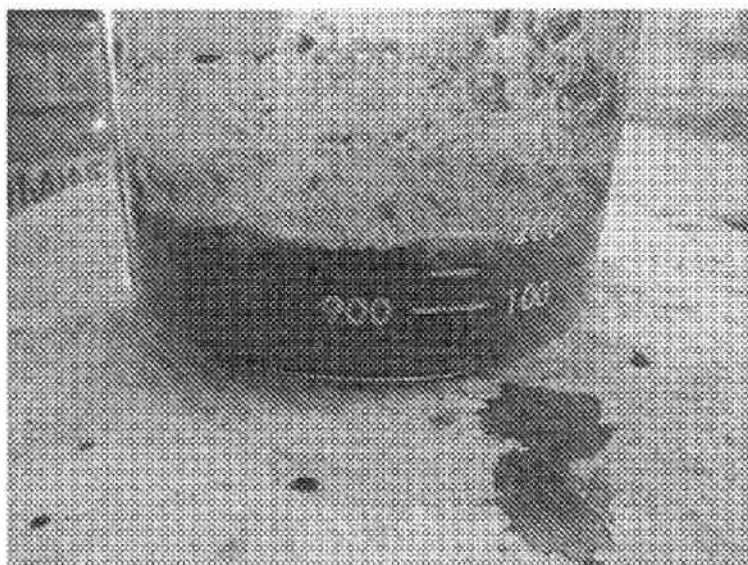


Figure 12

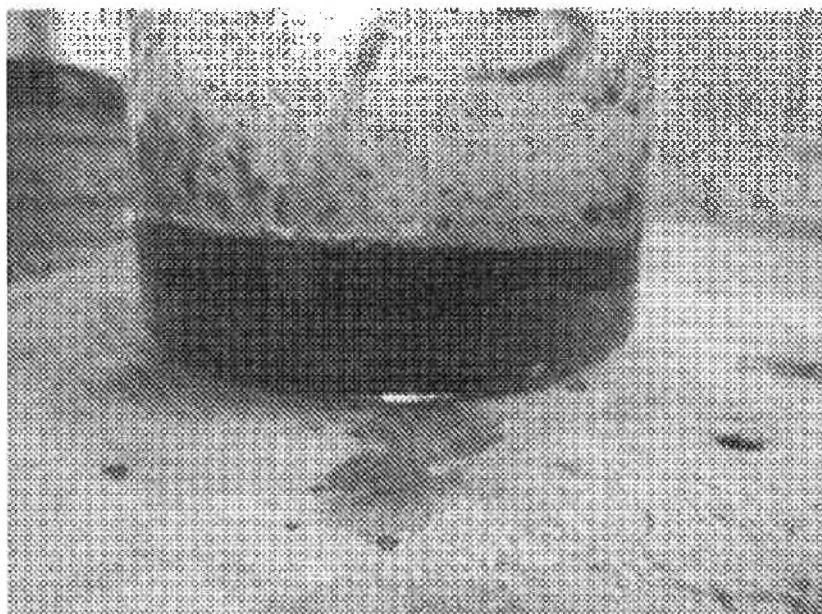


Figure 13

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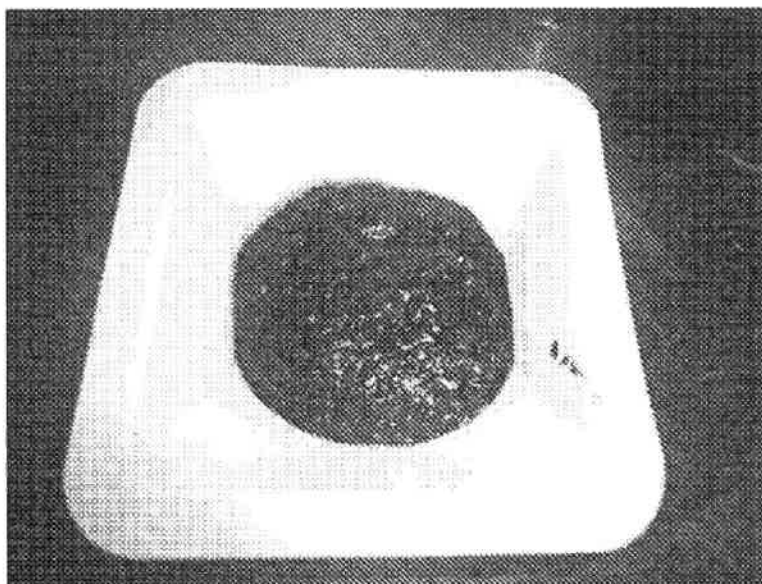


Figure 14

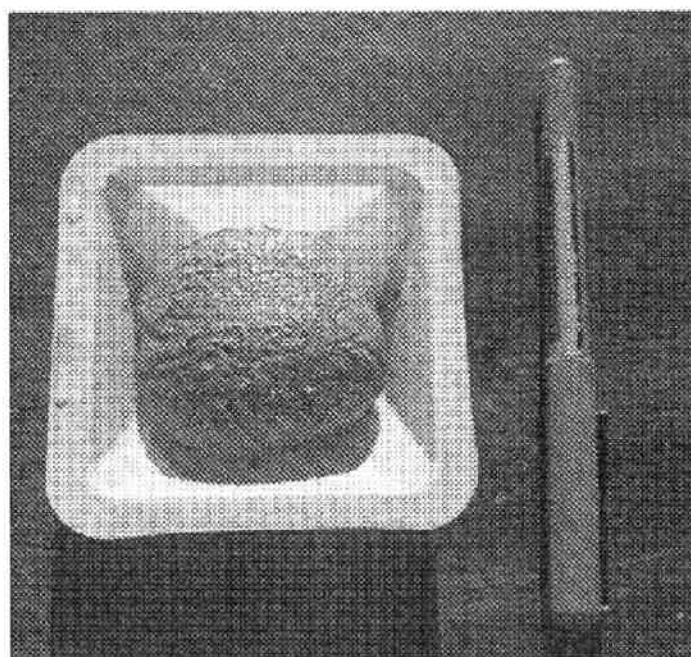


Figure 15

TECHNICAL MAINTENANCE: Tuesday, March 11 from 18:00 to 20:00 (CET/CEST) Favorites and IP Portal Dashboard will not be accessible. We apologize for any inconvenience. X

1. AU2014282964

Publication Date Description

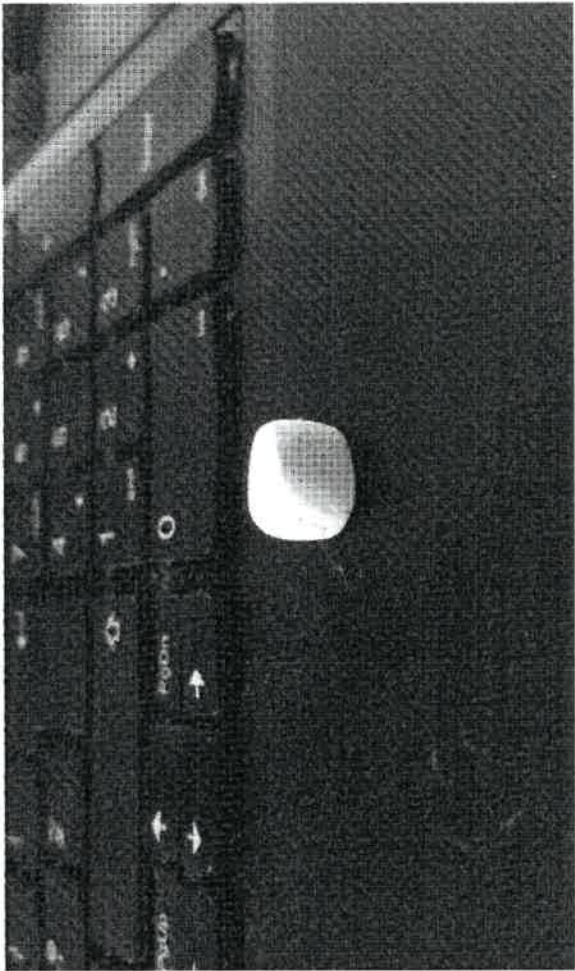


Figure 16

Image 9 of 9

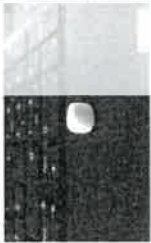


Figure 15



Figure 13



Figure 14